

# OMICS Journals are welcoming Submissions

OMICS International welcomes submissions that are original and technically so as to serve both the developing world and developed countries in the best possible way.

OMICS Journals are poised in excellence by publishing high quality research. OMICS International follows an Editorial Manager® System peer review process and boasts of a strong and active editorial board. Editors and reviewers are experts in their field and provide anonymous, unbiased and detailed reviews of all submissions.

The journal gives the options of multiple language translations for all the articles and all archived articles are available in HTML, XML, PDF and audio formats. Also, all the published articles are archived in repositories and indexing services like DOAJ, CAS, Google Scholar, Scientific Commons, Index Copernicus, EBSCO, HINARI and GALE.

**For more details please visit our website:**

**<http://omicsonline.org/Submitmanuscript.php>**

# MODELLING OF AUTOMOTIVE FUEL DROPLET HEATING AND EVAPORATION

*Sergei S Sazhin*

The Sir Harry Ricardo Laboratories, School of Computing, Engineering and Mathematics,  
University of Brighton, UK

## Concept

### (Engineering Mathematics)

Modelling of individual processes (numerical or analytical)

*Example:* modelling of deformation of individual droplets

*Strength:* allows in-depth analysis and understanding of underlying physics

*Weakness:* does not take into account the effects of other processes in realistic engineering systems

# Concept

## (Computational Fluid Dynamics)

Modelling of realistic engineering systems (numerical)

*Example:* modelling of Diesel engines

*Strength:* takes into account the interaction between various processes (fluid dynamics, heat/mass transfer etc)

*Weakness:* too simplistic models for individual processes are used; difficult to understand the underlying physics

***Analysis of realistic engineering systems  
requires a combined approach using  
the methods of engineering mathematics  
and Computational Fluid Dynamics (CFD)***

## MODELS FOR DOPLET HEATING AND EVAPORATION (*Conventional CFD models*)

1. The Infinite Thermal Conductivity (ITC) models based on the energy balance equation of the droplet as a whole were used.
2. Droplets were assumed to be opaque grey spheres.
3. Droplet radii were assumed to be constant during each time step.
4. Too complex or too simplistic models for heating and evaporation of multi-component droplets were used.
5. Kinetic and molecular dynamic effects on droplet evaporation were ignored

## Conduction limit. Effective conductivity

$$c_1 \rho_1 \frac{\partial T}{\partial t} = k_1 \left( \frac{\partial^2 T}{\partial R^2} + \frac{2}{R} \frac{\partial T}{\partial R} \right) + P(R)$$

$$h(T_g - T_s) = -\rho_1 L \dot{R}_d + k_1 \left. \frac{\partial T}{\partial R} \right|_{R=R_d}$$

$$k_{\text{eff}} = \chi_T k_l,$$

where the coefficient  $\chi_T$  varies from about 1 (at droplet Peclet number  $\text{Pe}_{d(t)} = \text{Re}_{d(t)} \text{Pr}_l < 10$ ) to 2.72 (at  $\text{Pe}_{d(t)} > 500$ ) and can be approximated as

$$\chi_T = 1.86 + 0.86 \tanh \left[ 2.225 \log_{10} (\text{Pe}_{d(t)} / 30) \right].$$

$$P(R)\Big|_{R < R_d} = 3 \times 10^6 \quad a \quad \sigma \quad R_d^b \theta_R^4 / c_l \rho_l$$
$$P(R)\Big|_{R > R_d} = 0$$

where  $\vartheta_R$  is the radiation temperature,  $R_d$  is the droplet radius,

$$a = a_0 + a_1 \theta_R / 10^3 + a_2 (\theta_R / 10^3)^2,$$

$$b = b_0 + b_1 \theta_R / 10^3 + b_2 (\theta_R / 10^3)^2,$$

$\vartheta_R$  can be assumed equal to the external temperature  $T_{\text{ext}}$  in the case of an optically thin gas in the whole domain.



## Analytical solution for $h=\text{const}$

$$T(r, t) = \frac{1}{r} \sum_{n=1}^{\infty} \left\{ \frac{p_n}{\kappa \lambda_n^2} + \exp[-\kappa \lambda_n^2 t] \left( q_n - \frac{p_n}{\kappa \lambda_n^2} \right) - \frac{\sin \lambda_n}{\|v_n\|^2 \lambda_n^2} \mu_0(0) \right. \\ \left. \times \exp[-\kappa \lambda_n^2 t] - \frac{\sin \lambda_n}{\|v_n\|^2 \lambda_n^2} \int_0^t \frac{d\mu_0(\tau)}{d\tau} \exp[-\kappa \lambda_n^2 (t - \tau)] d\tau \right\} \sin \lambda_n r + T_{\text{eff}}(t),$$

where  $\lambda_n$  are positive solutions (in ascending order) of the equation:

$$\lambda \cos \lambda + h_0 \sin \lambda = 0, \quad h_0 = (hR_d/k_l) - 1 = \text{const}, \quad r = R/R_d, \quad \kappa = \frac{k_l}{c_l \rho_l R_d^2},$$

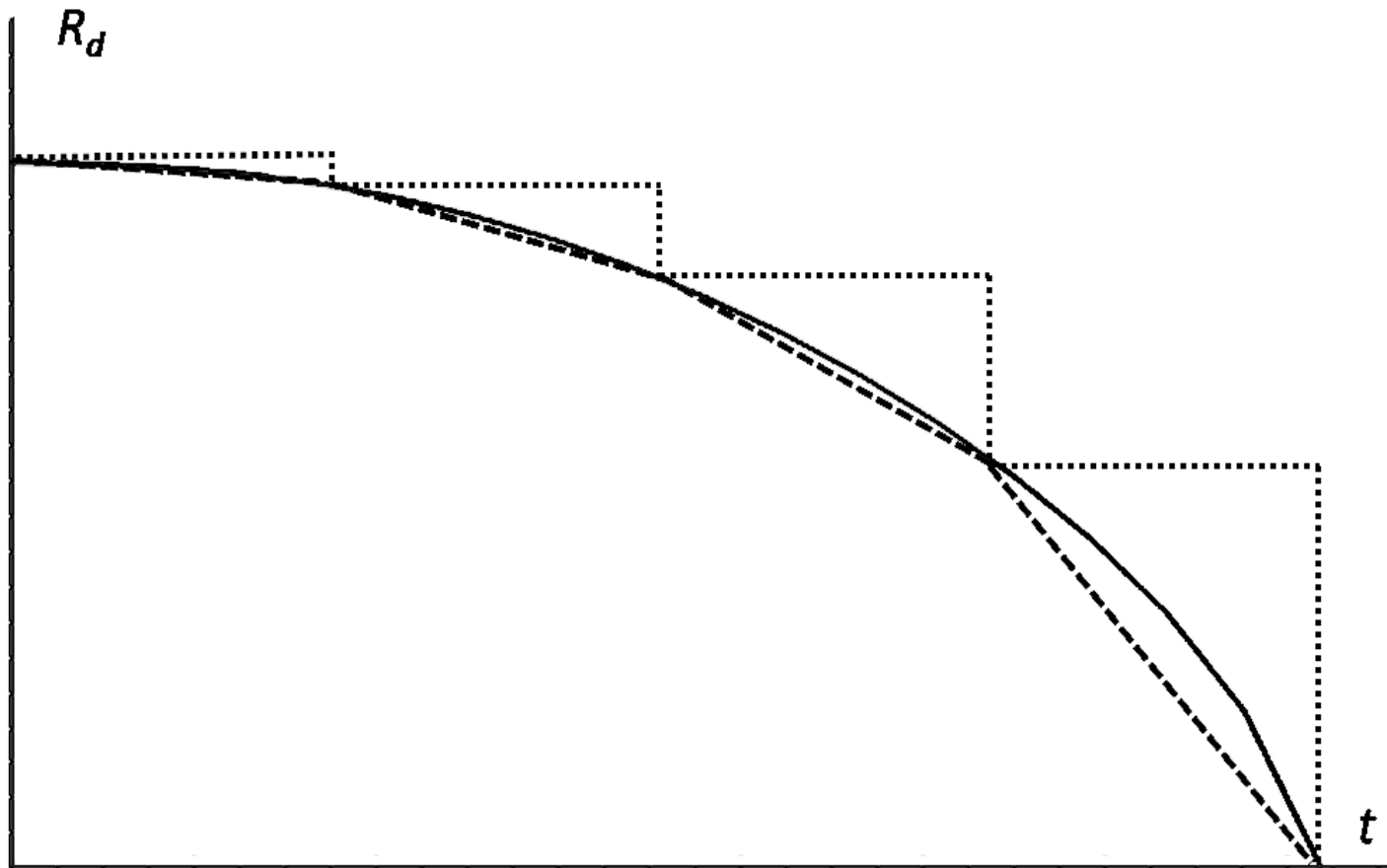
$$T_{\text{eff}} = T_g + \frac{\rho_l L \dot{R}_d}{h}, \quad \mu_0(t) = \frac{h T_{\text{eff}}(t) R_d}{k_l},$$

$$\|v_n\|^2 = \frac{1}{2} \left( 1 - \frac{\sin 2\lambda_n}{2\lambda_n} \right) = \frac{1}{2} \left( 1 + \frac{h_0}{h_0^2 + \lambda_n^2} \right),$$

$$q_n = \frac{1}{\|v_n\|^2} \int_0^1 \tilde{T}_0(r) v_n(r) dr, \quad \tilde{T}_0(r) = r T_0(r R_d).$$

$$p_n = \frac{1}{\|v_n\|^2} \int_0^1 \tilde{P}(r) v_n(r) dr, \quad \tilde{P}(r) = r P(r), \quad r = R/R_d$$

## Effect of the moving interface



## Effect of the moving interface

$$\frac{\partial T}{\partial t} = \kappa \left( \frac{\partial^2 T}{\partial R^2} + \frac{2}{R} \frac{\partial T}{\partial R} \right) + P(R)$$

for  $0 \leq t \leq t_e$ ,  $0 \leq R \leq R_d(t)$ , where  $\kappa$  is liquid thermal diffusivity ( $\kappa = k_l / (c_l \rho_l) = \text{const}$ ),  $k_l$  is the liquid thermal conductivity,  $c_l$  is the liquid specific heat capacity,  $\rho_l$  is the liquid density,  $R$  is the distance from the centre of the droplet.

$$\left( k_l \frac{\partial T}{\partial R} + hT \right) \Big|_{R=R_d(t)} = hT_g + \rho_l L \dot{R}_d(t)$$

$$T(t=0) = T_0(R) \qquad \dot{R}_d = - \frac{k_g \ln(1 + B_M)}{\rho_l c_{pg} R_d}$$

$$R_d(t) = R_{d0}(1 + \alpha t)$$

## Analytical solution

$$T(R) = \frac{1}{R\sqrt{R_d(t)}} \exp \left[ -\frac{\alpha R_{d0} R^2}{4\kappa R_d(t)} \right] \left[ \sum_{n=1}^{\infty} \Theta_n(t) \sin \left( \lambda_n \frac{R}{R_d(t)} \right) + \frac{\mu_0(t)}{1+h_0} \frac{R}{R_d(t)} \right]$$

$$\Theta_n(t) = q_n \exp \left[ -\frac{\kappa \lambda_n^2 t}{R_{d0} R_d(t)} \right] + f_n \mu_0(t)$$

$$- f_n \kappa \lambda_n^2 \int_0^t \frac{\mu_0(\tau)}{R_d^2(\tau)} \exp \left[ \frac{\kappa \lambda_n^2}{\alpha R_{d0}} \left( \frac{1}{R_d(t)} - \frac{1}{R_d(\tau)} \right) \right] d\tau.$$

$$f_n = \frac{1}{\|v_n\|^2} \int_0^1 f(\xi) v_n(\xi) d\xi = -\frac{\sin \lambda_n}{\|v_n\|^2 \lambda_n^2}$$

$$\lambda \cos \lambda + h_0 \sin \lambda = 0$$

$$\|v_n\|^2 = \frac{1}{2} \left( 1 - \frac{\sin 2\lambda_n}{2\lambda_n} \right) = \frac{1}{2} \left( 1 + \frac{h_0}{h_0^2 + \lambda_n^2} \right)$$

$$h_0 = \frac{h(t)}{k_l} R_d(t) - 1 - \frac{R_d'(t) R_d(t)}{2\kappa}$$

## *Discrete Components Model*

$$k_{\text{eff}} = \chi k_l \quad \frac{\partial Y_{li}}{\partial t} = D_l \left( \frac{\partial^2 Y_{li}}{\partial R^2} + \frac{2}{R} \frac{\partial Y_{li}}{\partial R} \right)$$

$$\alpha(\epsilon_i - Y_{lis}) = -D_l \left. \frac{\partial Y_{li}}{\partial R} \right|_{R=R_d-0} \quad \alpha = \frac{|\dot{m}_d|}{4\pi\rho_l R_d^2}$$

$$D_{\text{eff}} = \chi_Y D_l \quad p_{vi} = X_{li} \gamma_i p_{vi}^*$$

# Solution to the species diffusion equation

$$Y_{li} = \epsilon_i + \frac{1}{R} \left\{ \exp \left[ D_l \left( \frac{\lambda_0}{R_d} \right)^2 t \right] [q_{i0} - \epsilon_i(0)Q_0] \sinh \left( \lambda_0 \frac{R}{R_d} \right) + \sum_{n=1}^{\infty} \exp \left[ -D_l \left( \frac{\lambda_n}{R_d} \right)^2 t \right] [q_{in} - \epsilon(0)Q_n] \sin \left( \lambda_n \frac{R}{R_d} \right) \right\}.$$

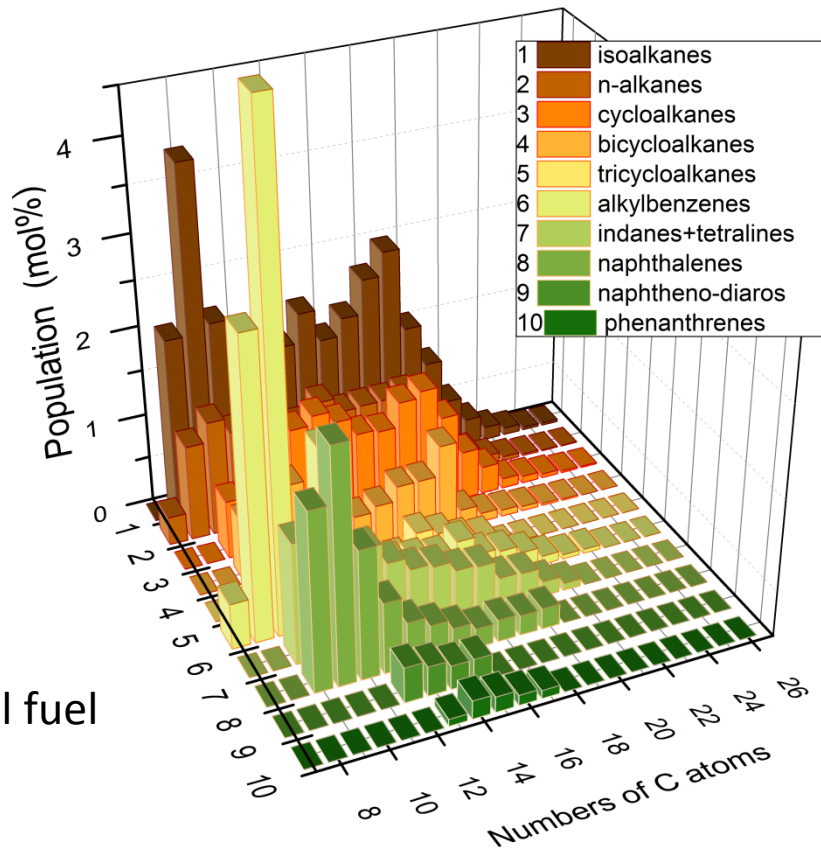
$$\tanh \lambda = -\frac{\lambda}{h_0}$$

$$v_0(R) = \sinh \left( \lambda_0 \frac{R}{R_d} \right)$$

$$\tan \lambda = -\frac{\lambda}{h_0}$$

$$v_n(R) = \sin \left( \lambda_n \frac{R}{R_d} \right)$$

# Multi-Dimensional Quasi-Discrete Model



Realistic Diesel fuel

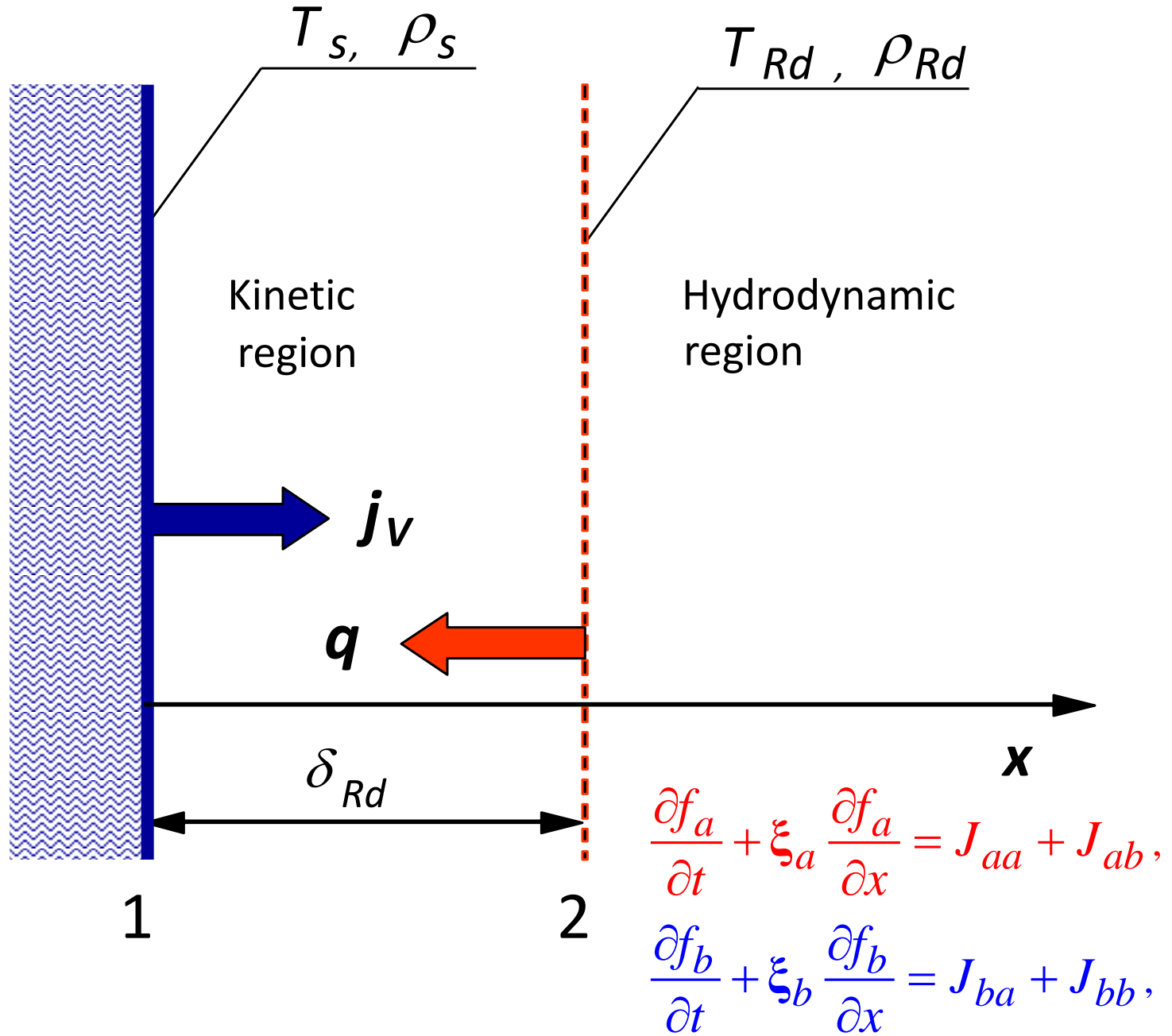
## Multi-Dimensional Quasi-Discrete Model

$m$	Component
1	alkanes
2	cycloalkanes
3	bicycloalkanes
4	alkylbenzenes
5	indanes & tetralines
6	naphthalenes
7	tricycloalkane
8	diaromatic
9	phenanthrene

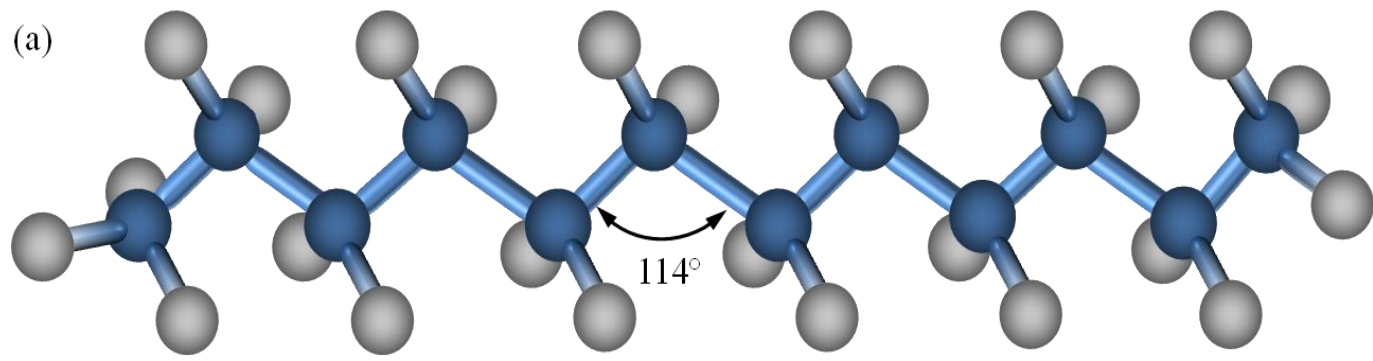
$$\left. \begin{aligned} \bar{n}_{1m} &= \frac{\sum_{n=n_{1m}}^{n=n_{(\varphi_m+1)m}} (nX_{nm})}{\sum_{n=n_{1m}}^{n=n_{(\varphi_m+1)m}} X_{nm}}, \\ \bar{n}_{2m} &= \frac{\sum_{n=n_{(\varphi_m+2)m}}^{n=n_{(2\varphi_m+2)m}} (nX_{nm})}{\sum_{n=n_{(\varphi_m+2)m}}^{n=n_{(2\varphi_m+2)m}} X_{nm}}, \\ \bar{n}_{3m} &= \frac{\sum_{n=n_{(2\varphi_m+3)m}}^{n=n_{(3\varphi_m+3)m}} (nX_{nm})}{\sum_{n=n_{(2\varphi_m+3)m}}^{n=n_{(3\varphi_m+3)m}} X_{nm}}, \\ &\dots\dots\dots \\ \bar{n}_{\ell m} &= \frac{\sum_{n=n_{((\ell-1)\varphi_m+\ell)m}}^{n=n_{k_m}} (nX_{nm})}{\sum_{n=n_{((\ell-1)\varphi_m+\ell)m}}^{n=n_{k_m}} X_{nm}}, \end{aligned} \right\}$$



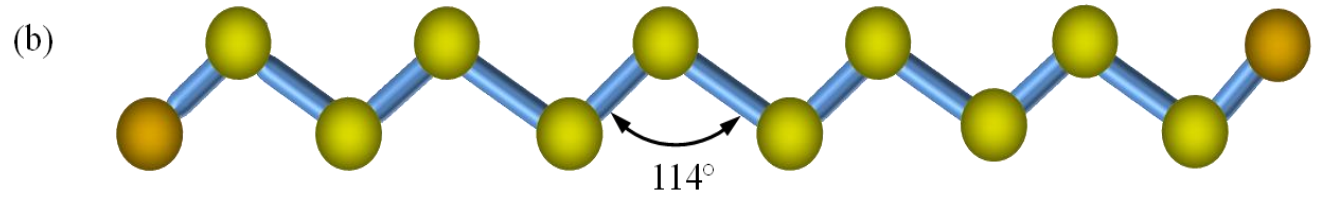
# The Sir Harry Ricardo Laboratories



## United Atom Model (UAM)

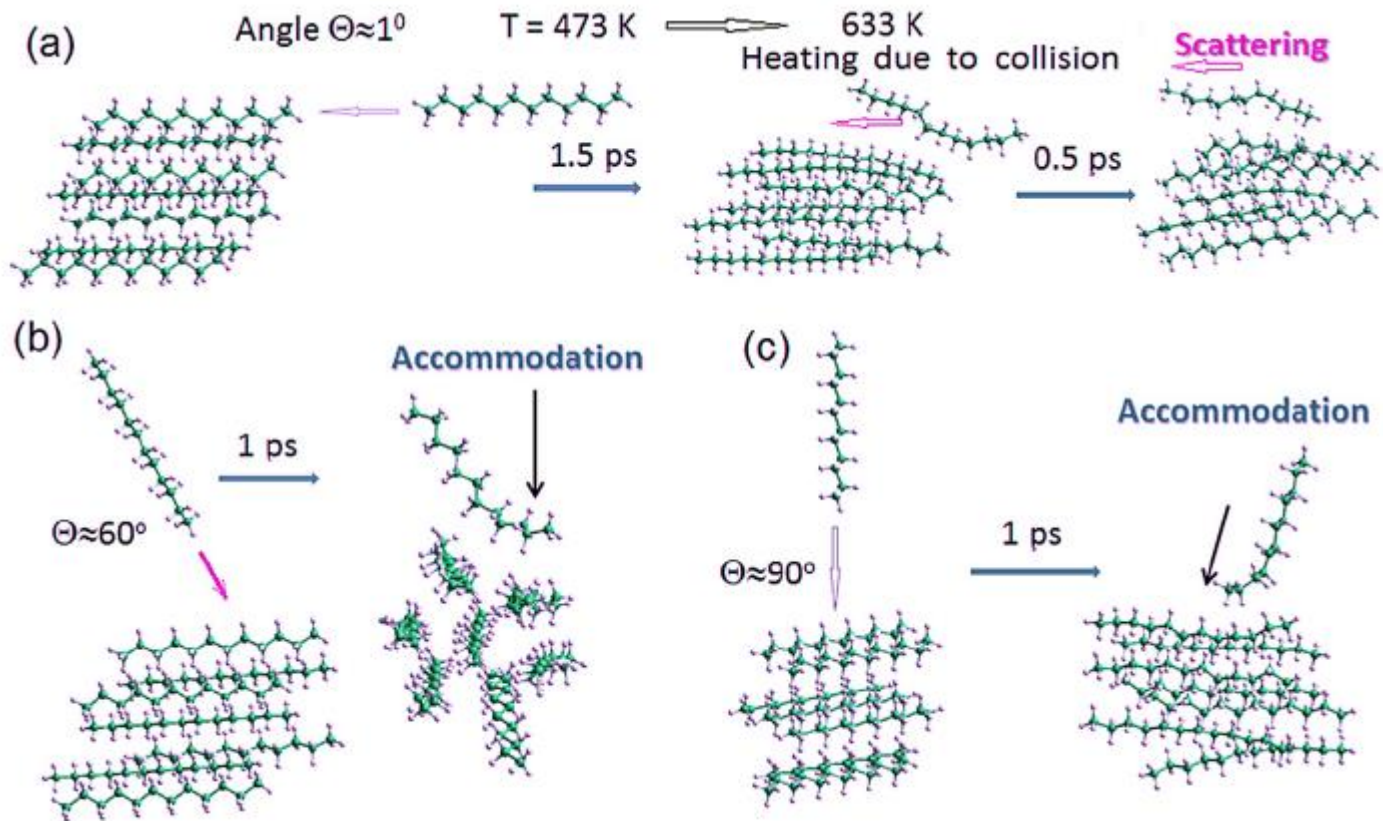


● Hydrogen atom      ● Carbon atom

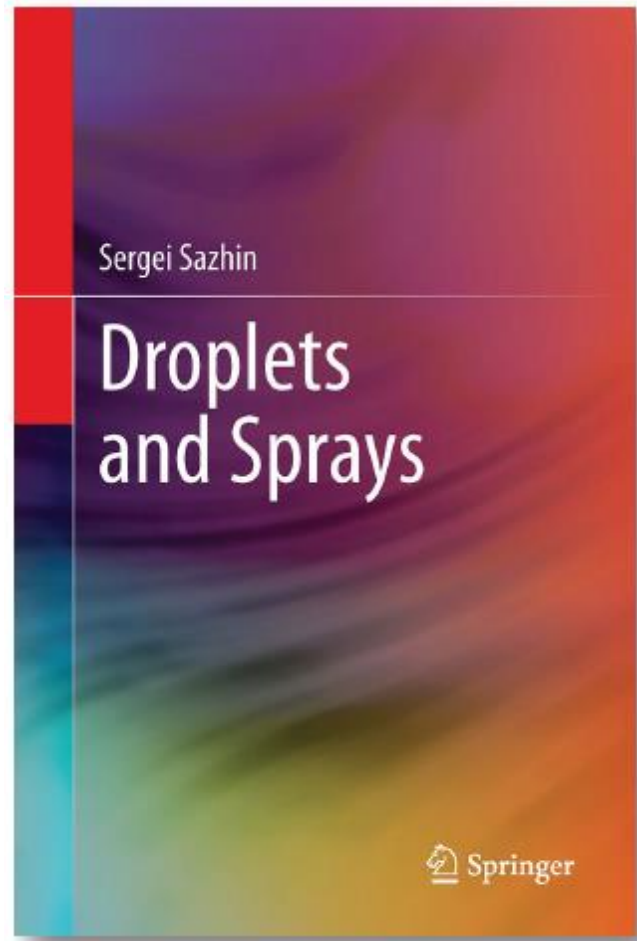


● Methyl group (CH<sub>3</sub>)      ● Methylene group (CH<sub>2</sub>)

A study of the evaporation and condensation of n-alkane clusters and nanodroplets using quantum chemical methods



Gun'ko, V.M., Nasiri, R., Sazhin, S.S. (2014) A study of the evaporation and condensation of n-alkane clusters and nanodroplets using quantum chemical methods, *Fluid Phase Equilibria*, **366** 99-107.





27<sup>th</sup> European Conference on Liquid Atomization and Spray Systems, University of Brighton, Brighton, UK

# Oil & Gas Research Related Journals

- [Journal of Petroleum & Environmental Engineering](#)
- [Pollution Effects & Control](#)
- [Bioremediation & Biodegradation](#)



# Oil & Gas Research Related Conferences

➤ For upcoming Conference visit  
<http://www.conferenceseries.com/>

