Nanof ormulation by multiple emulsion-evaporation (w/o/w) in solid lipid matrices can be achieved by means of ultrasonic cavitation. This two-step technique requiring a pre-emulsification step followed by an evaporation step employs two acoustic phenomena; namely emulsification and cavitation. Since the ultimate characteristics of the nanoparticles almost exclusively depend on the aforementioned two-steps, we investigated the influence of various acoustic phenomena and duration on the physical properties of the nanoparticles produced. We also report the influence of lipid composition on the properties of formed NLC.

NLC were formulated by multiple emulsion-evaporation using insulin as model drug. Organic solvent (Cl2CH2) was evaporated off in 2 hours at ambient temperature, by heat or by spraying. Characterization of the NLC was based on particle size analysis, SEM and DSC.

Aply, there was an inverse relationship between z-average of the NLC and amplitude of ultrasonication from 40% down 20%. Further reduction in z-average was observed when the primary emulsion was vortex-mixed prior to ultrasonication as compared to the same subjected to bath-ultrasonication. Solvent removal after emulsification was most efficient from the conical measure where NLC were fabricated compared to the use of heat or spray drying.

As palmitic acid content is increased, the state of entropy (ΔS) of the NLC decreases in the direction of less ordered structure. Conversely, ΔS increases with increasing amount of tripalmitin, with a less disrupted or more crystalline network structure. A slight lowering of melting temperature was observed with increasing palmitic acid content.

Very minor alteration in acoustic energy has a direct impact on properties of the NLC at both pre-emulsification as well as evaporation stage. This highlights the need to consider re-optimisation of formulation parameters, even with the slightest change to formulation variables. There is an indication that addition of chemically different solid lipid favour increase in the state of disorder, similarly to mixtures of solids and liquid lipids. However, an increase in state of disorder in the present instance does not appear to equate increase in drug accommodation. Therefore, it is likely that the state of disorder within chemically different solid lipid composites does not necessarily favour drug accommodation, as opposed to composites comprised of solid and liquid lipid.