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## Supermagnetic recyclable Bi/Fe-based nanomaterial with multiple functionalities and its potential practical application in environmental decontamination

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Advanced oxidation processes (AOPs) and physical adsorption are efficient and green approaches in environmental decontamination. As everyone knows,  $\text{TiO}_2$  can drive strong photocatalysis in slurry type and  $\text{Fe}^{2+}$  ion can induce Fenton oxidation at  $\text{pH} \sim 3$  while there are many investigations on adsorbents (e.g., urchin-like  $\alpha\text{-FeOOH}$  can adsorb  $80 \text{ mg g}^{-1}$  of  $\text{Pb(II)}$ , which is much higher than others). However, those used nanomaterials are difficult to separate from the treated water and the post-treatment will be high cost. Herein, we propose a Bi/Fe-based nanomaterial with hierarchical morphology, which can effectively drive AOPs in heterogeneous type at  $\text{pH} \sim 7$ , has outstanding physical adsorption and supermagnetic property. It can be used to remove organic pollutants and heavy metals, and can be recovered quickly via an environmental-friendly magnetic separation technology. The magnetic property for pristine bismuth ferrites is too weak to be used in practical application effectively. Here Bi/Fe-based materials with coral-like hierarchical morphology were fabricated using solvothermal treatment in methanol system. Its saturation magnetization ( $M_s$ ) marvelously increase from  $0.375$  to  $30.7 \text{ emu g}^{-1}$ , while the adsorption of methyl orange (a dye pollutant) ranges from  $0.5$  to  $46.6\%$ . Besides, it also can effectively induce visible light photo-Fenton oxidation which can be used to degrade different types of organic pollutants (e.g., dyes, pharmaceuticals, pesticides). Even at a low catalyst loading of  $0.12 \text{ g L}^{-1}$ , the removal rate of organic pollutants can be  $\sim 99\%$  in  $100 \text{ min}$  by degradation and/or adsorption. Its adsorption ability also can be used to remove different kinds of heavy metal ions (e.g.,  $\text{Pb(II)}$ ,  $\text{Cd(II)}$ ,  $\text{As(V)}$ ,  $\text{Cr(VI)}$ ,  $\text{Cu(II)}$ ,  $\text{Mn(II)}$ ,  $\text{Ba(II)}$  and  $\text{Co(II)}$ ), especially for  $\text{Pb(II)}$ , for which its maximal adsorption capacity can reach a new height of  $214.5 \text{ mg g}^{-1}$ . The outstanding performances are possibly ascribed to its coral-like hierarchical morphology which was investigated by several characterization techniques. It was proved that it is self-assembled by 1D nanowires ( $\sim 6 \text{ nm}$  in diameter) and 2D ultrathin nanoflakes ( $\sim 4.5 \text{ nm}$  in thickness). This product has remarkable optical properties with absorption of UV, visible light and even IR as well.

### Biography

Zhong-Ting Hu is a PhD student from Nanyang Technological University (Singapore). He holds BSc in Applied Chemistry and MSc in Environmental Engineering. He was a R&D researcher of NanoMaterials Technology Pte Ltd. in Singapore (2007-2012). He has experiences in nanoparticle synthesis, surface modification, wet coating and nanomaterial production in pilot plant. He was a Team Leader of a research project for undergraduate students (chemical plating &  $\text{H}_2$  energy) and their paper won the 1<sup>st</sup> Prize of the 1<sup>st</sup> ZJNU Natural Science's Academic Paper Competition. His current research interests are material and environment including advanced nanomaterials fabrication/optimization (morphology, self-assembly, nanocomposite, doping, synthesis), environmental photochemistry, heterogeneous catalysis, water treatment, solar energy, magnetic separation.

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