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Emission reduction of non-degradable and non-CO₂ greenhouse gas: Efficient destruction of CF₄ in an excess enthalpy combustor

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Background: CF₄ (tetrafluoromethane) being widely used as a cleaning and etching agent in semiconductor or display industry has a large global warming potential. In addition, it is chemically stable so that it is hardly decomposed even below 1600 °C by thermal methods. As it is usually utilized together with other explosive or toxic gases, it is diluted to around several thousands of ppm by N₂ before being disposed of. This excessive dilution makes it difficult to destruct the waste gas effectively. In this study, we developed an energy-efficient method of CF₄ destruction in an excess enthalpy combustor.

Experiment: An excess enthalpy combustor is a sort of two-section porous medium burner; two silicon carbide honeycombs with different cell sizes but with the same cylindrical shape were axially stacked. The emulated waste gas (CF₄, N₂) and the fuel-oxidant (CH₄, O₂) were fully premixed before being supplied to the combustor. A reaction front of combustion was stabilized around an interface between two honeycombs and the CF₄ inlet and outlet concentrations were measured by FTIR to determine a destruction efficiency of CF₄.

Findings: As a representative result presented in Fig. 1 shows, 94.6% of CF₄ was destructed and chemically transformed to HF, CO₂, and H₂O when the inlet CF₄ concentration was 2,150 ppm with the fuel usage of 18 LPM. This fuel usage in the destructing unit volume of CF₄ is much less than that of commercial abatement systems by about 30%. In addition, our combustor attained much higher CF₄ destruction efficiency than the commercial systems.

Conclusion: Our excess enthalpy combustor was found to have an advantage in reducing the greenhouse gas emissions, not only of CF₄ but also of CO₂ via reduced fuel usage.

Biography

Dae Keun Lee is a Principal Researcher at Korea Research Institute of Energy Research (KIER). He received his PhD degree from Korea Advanced Institute of Science and Technology (KAIST). His major concerns are fundamental understanding and practical applications of thermo-chemical and fluid dynamics, especially including combustion, by using mathematical and experimental methods.

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