

# Renewable Energy and Resources & Energy Materials and Fuel Cell Research

August 27-28, 2018 | Boston, USA

## Development of new bi-functional dense ceramic-carbonate membrane reactors for CO oxidation and subsequent CO<sub>2</sub> permeation

Pedro Sanchez-Camacho, Oscar Ovalle-Encinia and Heriberto Pfeiffer  
National Autonomous University of Mexico, Mexico

Gas separation processes have become one of the most promising strategies for different industrial applications; carbon dioxide (CO<sub>2</sub>) and carbon monoxide (CO) separation among them. Within this context, syngas is composed of hydrogen and carbon oxides, and it is usually produced at high temperatures. Therefore, H<sub>2</sub> enrichment, through the carbon oxides separation is of great interest. On CO<sub>2</sub> separation, different membrane systems have been developed, including zeolites, polymers, and ceramics. Especially, the so-called dense dual-phase membranes have shown very interestingly CO<sub>2</sub> separation properties at high temperatures. This kind of membrane systems is produced by a porous solid support infiltrated with molten carbonates, where ceramic supports must ideally have oxygen ionic and electronic conductive properties. CO<sub>2</sub> permeation is performed by the reaction of oxygen ions from the ceramic oxide with the CO<sub>2</sub> present on the upstream side, producing carbonate ions, which diffuse through the molten carbonate phase due to CO<sub>2</sub> partial pressure gradients on both membrane sides. CO<sub>2</sub> is desorbed on the downstream side through the reversible decarbonation process and swept from the surface. Oxygen ions are reincorporated and diffused on the ceramic phase as a consequence of decarbonization. Since perovskites have good ionic-electronic conduction properties, some of them have been incorporated to dense ceramic-carbonate membranes, increasing CO<sub>2</sub> permeation. In this work, a composite (doped ceria and perovskite) was synthesized, sintered and infiltrated with molten carbonates, showing that it is able to perform both processes; the CO oxidation at the surface and subsequent CO<sub>2</sub> permeation through the molten carbonate phase. CO conversion and CO<sub>2</sub> recovery efficiencies were 39.6 and 64.6% at 900 °C, respectively. Moreover, results in evidence that the perovskite phase importantly improve the oxygen permeation from sweep to feed side (inverse permeation), enhancing CO oxidation and CO<sub>3</sub><sup>2-</sup> formation, without releasing oxygen on the feed side.

### Biography

Pedro Sanchez-Camacho received a Chemistry's degree from the Universidad Nacional Autónoma de México (UNAM) in 2012. Later, he obtained a master's degree in Material Science and Engineering in same University. Nowadays, he studies a PhD in a Materials Science and Engineering focusing in the development of ceramic oxide membranes in order to separate CO<sub>2</sub> from a gas mixture. He is especially interested in solid state chemistry, ceramic synthesis, sorption processes on ceramics, catalysis, inorganic membranes for gas separation and solid oxide fuel cells (SOFC's). As part of his work, some papers have been published on referred journals such as Journal of Physical Chemistry and Journal of Energy Chemistry, among others.

pedsacam@gmail.com

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