

# Determination of Heavy Metal Profile in Bottled Water and Sachet Water Samples Obtained From Various Markets in Lagos, Nigeria

## Bolawa OE\* and Adelusi OS

Department of Biochemistry, College of Medicine, University of Lagos, Nigeria

## Abstract

The provision of safe drinking water is a very important public health priority. Most of the diseases in developing countries are caused by the consumption of contaminated water. The concentrations of six heavy metals and six trace metals were determined in sachet water samples and bottled water samples obtained in Lagos markets, Nigeria using Atomic Absorption Spectrophotometry (AAS). Lead concentrations were below the maximum permissible levels except in Bevap sachet water, Osan bottled water, Christo bottled water and Cway bottled water samples. Cadmium was not found in most of the water samples except in Pringad bottled water, Sunar bottled water and Cway bottle water samples together with Nimbus satchet water. These levels were above the maximum permissible levels of the WHO. Chromium was found present in all the water samples analyzed although in trace amounts. All other heavy metals analyzed were below the maximum permissible levels. The water samples also contained iron, manganese, calcium, magnesium, sodium and potassium in concentrations that are not harmful to the body. Examination of heavy metals in the water samples showed traces of lead in five sachet water samples and seven bottle water samples together with cadmium in three satchet water samples and three bottled water samples. This may be harmful after being consumed for a long time by the populace.

**Keywords:** Heavy metals; Lead; Chromium; Copper; World Health Organization

## Introduction

Water is a very essential component of our environment. The provision of safe drinking water is an important public health priority [1]. Water in its purest form is odorless, tasteless and colorless. Most of the diseases in developing countries are caused by the consumption of polluted water [2]. Over one thirds of deaths in most developing countries are caused by water pollution [3]. Water is necessary for the proper functioning of the human body. About 70% of the human body is made up of water. Potable water is good quality water that has neither smell nor odor and can be taken with no risk of harm [4]. Water is a very important component of biochemical processes in the body. Drinking water should conform to standards set by the World Health Organization. Water is a life sustaining drink and it is essential for the survival of all organisms. It is a very important part of metabolic process and it is involved in most biological processes [5]. The consumption of water containing toxic chemicals leads to damages in the human body. The metals can accumulate in the human body and poses serious health risks to the people consuming such unwholesome water [6]. Water can be rendered undrinkable if its physical qualities are undesirable. Due to this fact, the assessment of drinking water quality and its continuous monitoring are of utmost importance.

Most of the people living in developing countries lack access to clean water due to environmental pollution [7]. It is the right of every individual to have access to safe drinking water. The presence of heavy metals in drinking water leads to many diseases in the human populace [8]. Heavy metals are metals that have atomic weights of more than 40. They include arsenic, cadmium, lead, mercury, etc. They do not easily break down once they are in the environment and they have toxic effects on plants and human beings [9]. Chromium found in water is usually in the hexavalent form which is carcinogenic and highly toxic [10]. Lead has no essential function in man and it can be found occurring as metallic lead, lead salts and lead inorganic ions. Food and water are one of the major sources of lead exposure. In humans, about 20-50% of inhaled lead is absorbed while about 5-15% of ingested inorganic lead is absorbed. Once in the blood stream, lead is distributed among the soft tissue, mineralizing tissue and blood. Children are more sensitive to lead because of their rapid growth rate and metabolism [11].

Cadmium is naturally present in the environment in soils, the air sediments and sea water. It is emitted into the air by industries using cadmium compounds for pigments, batteries, plastic alloys, etc. [12]. People are exposed to cadmium when consuming plant and animal based foods, together with sea foods [13]. Cadmium accumulates in the human bodies and it affects the lungs, liver, kidney, brain central nervous system, etc. Other damages include hepatoxic toxicity, reproductive, hematological and immunological toxicities [14]. The WHO guideline for cadmium in drinking water is 0.003 mg/L [10].

Trace metals in minute amounts are essential in the human body because of their involvement in many processes within the human bodies. At higher concentrations, they are associated with increased risks for diabetes, cancer, liver disease, heart disease, endocrine disease, etc. [15]. Trace metals leak into groundwater from natural and anthropogenic sources thereby rendering them undrinkable [12].

One of the UN Millennium Development Goals is to reduce by half, the proportion of people without sustainable access to safe drinking water by the year 2015. The UN convention on the Rights of the child stipulates that all children have the right to safe drinking water [1]. The presence of heavy metals or trace metals in water renders it unfit for

\*Corresponding author: Bolawa OE, Biochemistry Department, college of medicine, university of lagos, p.m.b 12003, Idi-Araba, Lagos state, Nigeria, Tel: 08180495357; 08183359379; 07041248280; E mail: obolawa@unilag.edu.ng; olatundunebolawa@yahoo.com; tundun@safe-mail.net

Received March 10, 2016; Accepted March 20, 2017; Published March 27, 2017

**Citation:** Bolawa OE, Adelusi OS (2017) Determination of Heavy Metal Profile in Bottled Water and Sachet Water Samples Obtained From Various Markets in Lagos, Nigeria. Environ Pollut Climate Change 1: 114.

**Copyright:** © 2017 Bolawa OE, et al. This is an open-access article distributed under the terms of the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

human consumption. This may lead to accumulation of these metals in the body [16].

Trace metals in minute amounts are essential in the human body because of their involvement in many processes within the human body. At high concentrations, they are associated with increased risks for diabetes, cancer, liver disease, endocrine diseases etc. [5,12,15,17,18].

Trace metals leak into the ground water from natural and anthropogenic sources and render them undrinkable.

## Materials and Methods

## Sampling

Fifteen bottled water samples and ten sachet water samples were randomly purchased from different markets in Lagos, Nigeria.

pH: The pH of all the water samples was determined using the pH meter and the values recorded.

## Heavy metal analysis

100 ml of each water sample were acidified with 20 ml of nitric acid. The mixture was digested in a fume cupboard for one hour at 100°C until a clear solution was seen and the volume reduces to 20 ml. The mixture was transferred to 100 ml volumetric flask and diluted with deionized water and the mixture made up to 100 ml mark.

The mixture was filtered with filter paper after cooling and analyzed for lead, copper, chromium, calcium, zinc, nickel and cobalt using the Atomic Absorption Spectrophotometer.

## Results

Results found after performing the above methods are shown below (Tables 1-6).

## Discussion

Heavy metals enter the environment via natural and anthropogenic means. Such sources include industrial discharges, mining, erosion, sewage discharge water, waste effluents etc. the main route of exposure for most people is through food and water. Consistent exposure to heavy metals at low levels can cause great adverse effects [4,11,19].

Labelling	Names of bottled water samples	рН
Α	Cascade table water	6.76
В	Osan	6.80
С	Nestle	6.21
D	Eva	6.78
E	Aquarite	6.81
F	Christo	6.02
G	Pringad	5.60
Н	UncleToby	5.90
I	Mowa	6.30
J	Sunar	5.68
K	Cway	5.90
L	AquaDominion	6.70
М	Shower	6.83
N	Aquadana	6.76
0	Gossy	6.21

Standard pH: 6.5-8.5

Table 1: pH of the bottled water samples.

Labelling	Names of the sachet water samples	<b>pH</b> 6.81	
A1	Vee Vee sachet		
B1	LaMonde	5.28	
C1	Bako	6.10	
D1	Fizco	6.60	
E1	Living Seeds	5.78	
F1	Bevep	6.36	
G1	Nimbus	6.84	
H1	Chavis	5.77	
11	Denitol	6.86	
J1	Austin	6.90	

Page 2 of 5

Standard pH: 6.5-8.5

#### Table 2: pH of the sachet water samples.

Heavy metals have toxic effects on humans, animals, fisheries and plants [20-23]. They are persistent in the environment and they bioaccumulate in plants and human beings [13,24,25]. From the analysis carried out on the water samples, it was discovered that lead levels in the sample were within WHO limits with the exception of Bevep sachet water sample, Osan bottled water sample, Christo bottled water sample and Cway bottled water sample. Lead was not detected at all in Vee vee sachet, Living sachet, Nimbus sachet, Chavis sachet, Denifol sachet, Pringad bottled water, Uncle Toby bottled water, Mowa bottled water, Sunar bottled water and Aquandana bottled water samples. Out of the fifteen bottled water samples analyzed, cadmium was only detected in four water samples. Nickel and Copper levels were below the WHO permissible limits with the exception of Osan bottled water sample which had a nickel value of 0.06 mg/ml. Zinc level was above the WHO permissible limit in only Aqua dominion bottled water sample.

Children are more sensitive to lead because of their rapid growth rate and metabolism [26,27]. Cadmium is naturally present in the environment in soils, air, sediments and in sea water. It is emitted into the air by industries using cadmium compounds for pigments, plastics, alloys etc. People are exposed to cadmium when consuming plant and animal based foods, together with sea foods [13,28]. Cadmium accumulates in the human body and it affects the lungs, liver, kidney, brain, central nervous system etc. Other damages include hepatic toxicity, reproductive, hematological and immunological toxicities [19]. The WHO Maximum Permissible Level for cadmium in drinking water is 0.003 mg/l [29].

For the trace metals analyzed, iron was present in all the drinking water samples and it was below the permissible limits with the exception of Aqua dominion bottled water sample. Manganese concentration was high in Bako sachet, La monde sachet, Osan bottled water, Nestle bottled water and Pringad bottled water samples.

Previous studies have shown iron concentrations to be in high levels in ground and drinking water. The excess concentrations of iron in ground water may be due to the dissolution of rocks into the water system [30]. High levels of iron in drinking water may alter the appearance, taste, odor of water and may even promote the growth of bacteria in the water system. Copper is a potent neurotoxin and it accumulates in soft tissues and bones after prolonged exposure. The supply of water through piped water system leads to copper water pollution. High levels of copper in drinking water can cause vomiting, chronic anemia, abdominal pain, nausea, diarrhea, etc. [27]. Hussain et al. [31], collected water samples to determine the concentrations of metals such as nickel, lead, chromium, cadmium, zinc and copper. The concentrations of heavy metals found in drinking water samples ranged

Page 3 of 5

Water Samples	Pb (mg/ml)	Ni (mg/ml)	Cd (mg/ml)	Cr (mg/ml)	Cu (mg/ml)	Zn (mg/ml)
Vee Vee sachet	ND	0.01 ± 0.04	ND	0.05 ± 0.02*	0.04 ± 0.03	0.36 ± 0.07
La monde sachet	$0.01\pm0.02^{*}$	0.01 ± 0.01	ND	0.01 ± 0.01	0.02 ± 0.01	0.33 ± 0.05
Bako sachet	0.01 ± 0.01	0.01 ± 0.02	ND	0.01 ± 0.01	0.02 ± 0.04	1.49 ± 0.06
Fizco sachet	0.01 ± 0.02*	0.01 ± 0.01	ND	0.01 ± 0.03	0.03 ± 0.03	0.74 ± 0.09
Living sachet	ND	ND	ND	0.01 ± 0.02	0.01 ± 0.01	$0.43 \pm 0.08$
Bevep sachet	0.02 ± 0.03*	0.02 ± 0.04	ND	0.01 ± 0.01	$0.02 \pm 0.02$	1.43 ± 0.07
Nimbus sachet	ND	0.01 ± 0.02	$0.01 \pm 0.01^{*}$	0.01 ± 0.03	0.02 ± 0.04	1.42 ± 1.09
Chavis sachet	ND	ND	ND	0.01 ± 0.01	0.17 ± 0.09	1.00 ± 0.08
Denifol sachet	ND	0.01 ± 0.02	ND	0.01 ± 0.01	0.03 ± 0.06	1.33 ± 1.86
Austin sachet	0.01 ± 0.03*	ND	ND	0.01 0.02	0.02 ± 0.06	2.17 ± 1.98*
WHO MPL	0.01	0.02	0.003	0.05	2.00	3.00

<sup>∗</sup>p>0.05

#### Table 3: Concentration of heavy metals in sachet water samples.

Water Samples	Pb (mg/ml)	Ni (mg/ml)	Cd (mg/ml)	Cr (mg/ml)	Cu (mg/ml)	Zn (mg/ml)
Cascade	ND	0.02 <u>+</u> 0.01	ND	0.01 <u>+</u> 0.01	0.03 <u>+</u> 0.04	1.33 <u>+</u> 0.42
Osan	0.03 <u>+</u> 0.02 <sup>*</sup>	0.06 <u>+</u> 0.09 <sup>*</sup>	ND	0.01 <u>+</u> 0.01	0.03 <u>+</u> 0.02	1.26 <u>+</u> 0.52
Nestle	0.01 <u>+</u> 0.01	0.01 <u>+</u> 0.02	ND	0.01 <u>+</u> 0.01	0.02 <u>+</u> 0.01	2.09 <u>+</u> 0.88 <sup>*</sup>
Eva	0.01 <u>+</u> 0.01	0.01 <u>+</u> 0.01	ND	0.02 <u>+</u> 0.02 <sup>*</sup>	0.02 <u>+</u> 0.02	1.30 <u>+</u> 0.47
Aquarite	0.01 <u>+</u> 0.01	0.01 <u>+</u> 0.01	ND	0.01 <u>+</u> 0.02	0.02 <u>+</u> 0.01	0.78 <u>+</u> 0.55
Christo	0.02 <u>+</u> 0.03 <sup>*</sup>	0.01 <u>+</u> 0.02 <sup>*</sup>	ND	0.01 <u>+</u> 0.01	0.03 <u>+</u> 0.05	0.78 <u>+</u> 0.52
Pringad	ND	0.01 <u>+</u> 0.01	0.001 <u>+</u> 0.001	ND	0.01 <u>+</u> 0.01	1.43 <u>+</u> 0.99
Uncle Toby	ND	ND	ND	0.01 <u>+</u> 0.01	0.03 <u>+</u> 0.02	1.27 <u>+</u> 0.78
Mowa	ND	ND	ND	ND	0.01 <u>+</u> 0.01	1.31 <u>+</u> 0.76
Sunar	ND	0.003 <u>+</u> 0.001	0.01 <u>+</u> 0.01 <sup>*</sup>	0.01 <u>+</u> 0.01	0.02 <u>+</u> 0.01	0.79 <u>+</u> 0.52
Cway	0.02 <u>+</u> 0.01 <sup>*</sup>	0.01 <u>+</u> 0.01	0.01 <u>+ 0</u> .01 <sup>*</sup>	0.01 <u>+</u> 0.01	0.06 <u>+ 0</u> .08 <sup>*</sup>	5.04 <u>+</u> 0.97 <sup>*</sup>
Aqua dominion	ND	ND	ND	0.01 <u>+</u> 0.02	0.04 <u>+</u> 0.06*	8.23 <u>+</u> 0.89*
Shower	0.01 <u>+</u> 0.01	0.01 <u>+</u> 0.01	ND	0.01 <u>+</u> 0.01	0.03 <u>+</u> 0.02	2.29 <u>+</u> 0.57
Aquadana	ND	0.01 <u>+</u> 0.01	ND	0.02 <u>+</u> 0.01	0.05 <u>+</u> 0.03	0.28 <u>+</u> 0.33
WHO MPL	0.01	0.02	0.003	0.05	2.00	3.00

\*p>0.05

#### Table 4: Concentration of heavy metals in bottled water samples.

Water Samples	Fe (mg/ml)	Mn (mg/ml)	Ca (mg/ml)	Mg (mg/ml)	Na (mg/ml)	K (mg/ml)
Vee Vee	0.10 <u>+</u> 0.09	0.07 <u>+</u> 0.06	6.60 <u>+</u> 1.47	0.14 <u>+</u> 0.30	99.50 <u>+</u> 10.03	9.84 <u>+</u> 1.67 <sup>*</sup>
La monde	0.08 <u>+</u> 0.36	0.45 <u>+</u> 0.12⁺	3.93 <u>+</u> 1.94	0.12 <u>+</u> 0.53	98.37 <u>+</u> 17.11	4.59 <u>+</u> 1.44
Bako	0.08 <u>+</u> 0.88	0.60 <u>+</u> 0.47 <sup>*</sup>	7.07 <u>+</u> 1.40	0.13 <u>+</u> 0.69	121.15 <u>+</u> 14.00	3.60 <u>+</u> 1.72
Fizco	0.16 <u>+</u> 0.21 <sup>*</sup>	0.30 <u>+</u> 0.22	10.81 <u>+</u> 1.99 <sup>•</sup>	0.14 <u>+</u> 0.22	130.05 <u>+</u> 15.87	1.64 <u>+</u> 1.33
Living	0.01 <u>+</u> 0.01	0.06 <u>+</u> 0.53	1.09 <u>+</u> 1.01	0.14 <u>+</u> 0.16	100.45 <u>+</u> 13.66	0.99 <u>+</u> 1.01
Bevep	0.15 <u>+</u> 0.31 <sup>*</sup>	0.04 <u>+</u> 0.03	13.75 <u>+</u> 1.68 <sup>*</sup>	0.15 <u>+</u> 0.39⁺	155.15 <u>+</u> 14.72 <sup>*</sup>	1.95 <u>+</u> 1.76
Nimbus	0.07 <u>+</u> 0.57	0.08 <u>+</u> 0.08	11.96 <u>+</u> 1.45	0.13 <u>+</u> 0.44	188.45 <u>+</u> 17.99 <sup>*</sup>	`3.32 <u>+</u> 1.49
Chavis	`0.05 <u>+</u> 0. 48	0.04 <u>+</u> 0.05	8.09 <u>+</u> 1.55	0.14 <u>+</u> 0.28	100.15 <u>+</u> 14.68	4.65 <u>+</u> 1.44 <sup>*</sup>
Denifol	0.07 <u>+</u> 0.62	0.07 <u>+</u> 0.09	18.51 <u>+</u> 1.73 <sup>*</sup>	0.16 <u>+</u> 0.71 <sup>*</sup>	194.55 <u>+</u> 12.77 <sup>⁺</sup>	2.38 <u>+</u> 1.79
Austin	0.18 <u>+</u> 0.29 <sup>*</sup>	0.08 <u>+</u> 0.07	17.73 <u>+</u> 1.33 <sup>*</sup>	`0.13 <u>+</u> 0.47	`120.15 <u>+</u> 15.32	1.67 <u>+</u> 1.22
MCL	0.30	0.40			200.00	10.00

\*p>0.05

#### Table 5: Concentration of trace metals in sachet water samples.

from 0.01-0.10 mg/ml (nickel), 0.00-0.03 mg/ml (lead), 0.01-0.02 mg/ml (chromium), 0.01-0.16 mg/ml (zinc) and 0.00-0.01 mg/ml (copper). These results are similar to the results gotten in this study, with the exception of zinc concentrations which was higher in this study. The water of Mardan tube well can only be fit for drinking after magnesium is removed by boiling the water. Mebrahthu et al. [32] showed that concentrations of heavy metals in drinking water samples collected from Northern Ethopia have some physiochemical parameter values higher than WHO recommended limits. Samples analyzed showed

that arsenic, cadmium, chromium, iron, nickel and lead were above the WHO Maximum Permissible Limits.

It is therefore advisable for manufacturers of drinking water either in sachet or bottle form to treat the water using iron filters before packaging them [6]. Trace metals are required by the body in small amount for various metabolic activities but at high concentrations, they can cause adverse effects to the body. On the other hand, toxic metals have no beneficial effects in humans. Exposure to them leads to toxic human health effects [17].

Page 4 of 5

Water Samples	Fe (mg/ml)	Mn (mg/ml)	Ca (mg/ml)	Mg (mg/ml)	Na (mg/ml)	K (mg/ml)
Cascade	0.06 <u>+</u> 0.03	0.08 <u>+</u> 0.77	8.53 <u>+</u> 1.02	0.16 <u>+</u> 0.88 <sup>*</sup>	97.15 <u>+</u> 12.33	7.95 <u>+</u> 1.08⁺
Osan	0.09 <u>+</u> 0.05	0.49 <u>+</u> 0.13 <sup>*</sup>	2.10 <u>+</u> 1.01	0.14 <u>+</u> 0.25	101.95 <u>+</u> 15.66	5.97 <u>+</u> 1.06 <sup>*</sup>
Nestle	0.09 <u>+</u> 0.08	0.53 <u>+</u> 0.11 <sup>*</sup>	33.59 <u>+</u> 1.80 <sup>⁺</sup>	0.15 <u>+</u> 0.41	101.55 <u>+</u> 16.87	4.98 <u>+</u> 1.22
Eva	0.09 <u>+</u> 0.06	0.40 <u>+</u> 0.16 <sup>*</sup>	6.29 <u>+</u> 1.22	0.16 <u>+</u> 0.10 <sup>*</sup>	129.45 <u>+</u> 13.99	3.26 <u>+</u> 1.02
Aquarite	0.08 <u>+</u> 0.09	0.06 <u>+</u> 0.12	7.52 <u>+</u> 1.53	0.17 <u>+</u> 0.12 <sup>∗</sup>	121.15 <u>+</u> 16.25	1.07 <u>+</u> 1.00
Christo	0.18 <u>+</u> 0.05	0.02 <u>+</u> 0.03	6.10 <u>+</u> 1.32	0.15 <u>+</u> 0.14	164.65 <u>+</u> 19.20⁺	1.21 <u>+</u> 1.01
Pringad	0.09 <u>+</u> 0.07	0.46 <u>+ 0</u> .13 <sup>*</sup>	14.95 <u>+</u> 1.88 <sup>*</sup>	0.15 <u>+</u> 0.13	178.75 <u>+</u> 16.47 <sup>⁺</sup>	3.66 <u>+</u> 1.20
Un. Toby	0.06 <u>+</u> 0.03	0.06 <u>+</u> 0.04	5.93 <u>+</u> 1.10	0.14 <u>+</u> 0.11	95.30 <u>+</u> 10.04	2.13 <u>+</u> 1.09
Mowa	0.06 <u>+</u> 0.05	0.07 <u>+</u> 0.08	10.22 <u>+</u> 1.98	0.16 <u>+</u> 0.18 <sup>*</sup>	145.75 <u>+</u> 19.72⁺	5.98 <u>+</u> 1.53 <sup>*</sup>
Sunar	0.06 <u>+</u> 0.08	0.05 <u>+</u> 0.02	7.36 <u>+</u> 1.33	0.17 <u>+</u> 0.12 <sup>*</sup>	178.15 <u>+</u> 13.98⁺	1.78 <u>+</u> 1.52
Cway	0.28 <u>+</u> 0.12 <sup>*</sup>	0.09 <u>+</u> 0.06	11.25 <u>+</u> 1.79 <sup>*</sup>	0.13 <u>+</u> 0.19	110.95 <u>+</u> 10.88	1.75 <u>+</u> 1.88
Aqua d	0.40 <u>+</u> 0.11 <sup>⁺</sup>	0.06 <u>+</u> 0.02	7.19 <u>+</u> 1.68	0.14 <u>+</u> 0.15	117.15 <u>+</u> 12.57	2.27 <u>+</u> 1.05
Shower	0.18 <u>+</u> 0.13	0.07 <u>+</u> 0.04	7.55 <u>+</u> 1.66	0.16 <u>+</u> 0.14 <sup>*</sup>	156.50 <u>+</u> 12.89 <sup>*</sup>	2.23 <u>+</u> 1.00
Aquadana	0.22 <u>+</u> 0.12 <sup>*</sup>	0.08 <u>+</u> 0.05	8.13 <u>+</u> 1.99	0.14 <u>+</u> 0.14	110.45 <u>+</u> 14.01	1.11 <u>+</u> 1.01
WHO MPL	0.30	0.40			200.00	10.00

<sup>\*</sup>p>0.05

Table 6: Concentration of trace metals in bottled water samples.

#### Conclusion

It is recommended that all water samples intended for drinking purposes should be treated considerable to reduce the levels of heavy metals and trace metals present in to them to levels that are not harmful to the human body.

#### References

- WHO/UNICEF (2012): Estimated data from WHO/UNICEF Joint Monitoring Programme (JMP) for water supply and sanitation. Progress on sanitation and drinking water.
- Zuthi MFR, Biswas M, Bahar MW (2009) Assessment of supply water quality in the Chaittagong Town of Bangladesh. ARPN Journal of Engineering and Applied Sciences I 4.
- Chang TJ, Ke DS, Guo HR (2010) The association between arsenic exposure from drinking water and cerebrovascular disease mortality in Taiwan. Water Res 44: 5770-5776.
- WHO (2008) Guidelines for Drinking water quality. Third Edition. Incorporating the 1<sup>st</sup> and 2<sup>nd</sup> Agenda Volume Recommendations. World Health Organization, Geneva.
- Hassan AA, Karzan AH (2013) Evaluation of trace elements in drinking water of Duhok Province/Kurdistan region of Iraq. IJESI 2: 47-56.
- WHO (2011) Guidelines for drinking water quality. 4<sup>th</sup> Edition. World Health Organization, Geneva.
- Gyamfi ET, Achah M, Anim AK, Hanson JK, Kpattah L, et al. (2012): Chemical analysis of potable water samples from selected suburbs of Accra., Ghana. Proc Int Acad Ecol Environ Sci 2: 118-127.
- Al-Saleh, Al-Doush (1998) Survey of trace elements in household and bottled drinking water samples collected in Riyahd, Saudi Arabia. Sci Total Environ 216: 181-192.
- Kumar SM , Gupta OP, Singh DK, Prasad AS (2014) Comparative physiochemical analysis of river water and underground water in winter season of Rewa town, MP, India. Int Res J Environment Sci 3: 59-61.
- WHO (2004) Evaluation of certain food contaminants. 61<sup>st</sup> report of the Joint FAO/WHO Epert Committee on Food Additives. WHO Technical Report Series, No. 922. World Health Organization, Geneva.
- ATSDR Agency for Toxic Substances and Disease Registry (2008) Draft Toxicological profile for cadmium. United States Department of Health and Human Services, Public Health Human Services, Centre for Disease Control, Atlanta.
- 12. Khan AT (2011) Trace elements in Drinking Water and their Possible Health Effects in Aligarh City. J Water Resource and Prot 3: 522-530.
- 13. WHO (2006) Evaluation of certain food contaminants. 64th report of the Joint

FAO/WHO Epert Committee on Food Additives. WHO Technical Report Series No. 930. World Health Organization, Geneva.

- Apostoli P, Catalini S (2011) Metal ions affecting reproduction and development: Metal lons in Life Sciences; Metal ions in Toxicology: Effects, Interactions, Interdependence. ISC Publishing 8.
- Couto CM, Pinto I, Madureirea TV, Rocha MJ, Tiritan ME, et al. (2014) Lower Doureo river basin (Portugal) water quality-focus on trace element changes and anthropogenic sources of contamination. Glob NEST J 16: 252-268.
- UNEP (2007) Global drinking water quality index, development and sensitivity analysis report.
- Buschmann J, Berg M, Stengel C, Winkel L, Sampson ML, et al. (2008) Contamination of drinking water resources in the Mekong Delta flood plains; Arsenic and other trace metals pose serious health risks to the population. Environ Int 34: 756-764.
- Sing S, Mosely LM (2003) Trace metals in drinking water in Viti Levu, Fiji Islands. S Pac J Nat Sci 21: 31-34.
- Mudgal V, Madaan N, Mudgal A, Singh RB, Mishra S (2010) Effect of toxic metals on human health. Open Nutraceuticals J 3: 94-99.
- Bolawa OE, Gbenle GO, Ebuehi OAT (2014) Endocrine disruption by the consumption of fish and its reversal using zinc. International Journal of Aquaculture 4: 85-88.
- 21. Bolawa OE, Gbenle GO (2013) Effect of the consumption of heavy metal contaminated fish on metabolic parameters in rabbits. IJABPT 4: 317-321.
- Camacho LM, Gutierrez WM, Alarcon-Herrera MT (2011) Occurrence and treatment of arsenic in ground water and soil in Northern Mexico and South western USA. Chemosphere 83: 211-225.
- Azizullah A, Khattak MNK, Richer P, Hader DP (2011) Water pollution in Pakistan and its impact on public health - A review. Environ Int 37: 479-497.
- 24. Mridul B, Prasad SH (2013) A GIS based assessment of Pb, As, Cd and Al contamination of ground water in Dhemaji district of Assam, India. Int J Res Chem Environ 3: 1-7.
- Castro-Gonzalez M, Mendez-Armenta M (2008): Implications associated to food consumption. Environ Toxicol Pharmacol 26: 263-270.
- Nwabueze AA (2010) Levels of some heavy metals in tissues of Bonga fish, *Ethmalosa fimbriata* from Forcadoes River. J Appl Environ Biol Sci 1: 44-47.
- Ling MP, Hsu HT, Shie RH, Wu CC, M Hong YS (2009) Health risk of consuming heavy metals in farmed Tilapia in Central Taiwan. Bull Environ Contam Toxicol 83: 558-564.
- Medairos RJ, dos Santos LMG, Freire AS, Santelli RE, Braga AMCB, et al. (2012) Determination of inorganic trace elements in edible marine fish from Rio de Janeiro State, Brazil. Food Control 23: 535-541.

#### Page 5 of 5

- Shyamala R, Shanthi M, Lalitha P (2008) Physiochemical analysis of borehole water samples of Telungupalayam area in Coimbatore district, Tamil Nadu, India. E-Journal of Chemistry 5: 924-929.
- O Akonto, J Adiyah (2007) Chemical analysis of drinking water from some communities in the Brong Ahafo Region. Int J Environ Sci Tech 4: 211
- Hussain J, Shah J, Hussain W, Ali R, Sousa LJ, et al. (2012) Evaluation of the quality of drinking water of Mardan District, KPK, Pakistan. American Eurasian Journal of Agriculture and Environment Science 12: 1047-1051.
- Mebrahthu G, Zerabruk S (2011) Concentration and health implications of heavy metals in drinking water from urban areas of Tigray region, Northern Ethopia. Momona Ethiop J Sci 3: 1-4.

This article was originally published in a special issue, **Environmental Climate Change** handled by Editor(s). Arthur Viterito, Professor of Geography, USA

**Citation:** Bolawa OE, Adelusi OS (2017) Determination of Heavy Metal Profile in Bottled Water and Sachet Water Samples Obtained From Various Markets in Lagos, Nigeria. Environ Pollut Climate Change 1: 114.