

# Exploiting Chemical Industrial and Academic Wastes for Supported Photocatalysts: A Potential Source

Wilsmith Silva\*

Instituto de Química, Universidade Federal do Rio Grande do Sul, Brazil

## Description

The synthesis of supported photocatalysts holds immense promise for addressing environmental challenges and advancing sustainable chemistry practices. While the utilization of traditional materials for photocatalyst support is common, there exists a vast, untapped resource in the form of chemical industrial and academic wastes. This commentary explores the potential of such wastes as a sustainable source of supported photocatalysts, offering insights into their unique properties, challenges, and opportunities [1, 2].

Chemical industrial and academic activities generate a vast amount of waste materials, ranging from spent catalysts to unused chemicals and laboratory by-products. Traditionally, these wastes are viewed as environmental liabilities and are often disposed of through landfilling or incineration, contributing to pollution and resource depletion. However, there exists a significant untapped potential in these waste streams, particularly in the context of sustainable chemistry and materials synthesis [3, 4].

The utilization of chemical industrial and academic wastes as a source of supported photocatalysts represents a promising avenue for addressing environmental challenges and advancing sustainable practices. Photocatalysis, the process of harnessing light energy to drive chemical reactions, has garnered considerable attention for its applications in environmental remediation, energy conversion, and chemical synthesis. Supported photocatalysts, which consist of photocatalytic materials immobilized on a support matrix, offer enhanced stability, recyclability, and catalytic efficiency compared to their unsupported counterparts [5].

In recent years, researchers have begun to explore the feasibility of utilizing chemical industrial and academic wastes as support materials for photocatalysts. By repurposing these wastes, which possess unique physical and chemical properties, researchers can develop sustainable and cost-effective photocatalytic materials while simultaneously mitigating waste generation and environmental impact [6]. However, this approach presents both opportunities and challenges that warrant further investigation and discussion.

**Utilization of chemical industrial and academic wastes:** Chemical industrial processes and academic laboratories generate a diverse array of wastes, ranging from spent catalysts and by-products to unused chemicals and materials. Instead of discarding these wastes, they can be repurposed and transformed into valuable resources for synthesizing supported photocatalysts [7]. By harnessing the inherent properties of these wastes, such as surface chemistry, morphology, and composition, researchers can tailor the characteristics of supported photocatalysts for specific applications.

**Advantages and challenges:** The utilization of chemical industrial and academic wastes as a source of supported photocatalysts offers several advantages. Firstly, it promotes resource efficiency and waste minimization, aligning with principles of green chemistry and sustainability. Secondly, it reduces the environmental footprint associated with waste disposal and landfilling, contributing to circular

economy initiatives. However, challenges such as waste characterization, variability in composition, and scalability of synthesis methods must be addressed to realize the full potential of these materials [8].

**Case studies and examples:** Several studies have demonstrated the feasibility of utilizing chemical industrial and academic wastes for synthesizing supported photocatalysts. Examples include the utilization of spent catalysts from petrochemical refineries, laboratory glassware, and semiconductor manufacturing by-products [9]. By employing innovative synthesis approaches, such as sol-gel methods, hydrothermal treatment, and impregnation techniques, researchers have successfully transformed these wastes into functional supports for photocatalytic materials.

**Future directions and opportunities:** The exploration of chemical industrial and academic wastes as a source of supported photocatalysts opens up new avenues for research and innovation. Future studies should focus on developing scalable synthesis methods, optimizing material properties, and exploring novel waste sources. Collaborations between academia, industry, and waste management sectors are essential to harnessing the full potential of these materials and accelerating their translation into real-world applications [10].

## Conclusion

In conclusion, chemical industrial and academic wastes represent a valuable yet underutilized resource for synthesizing supported photocatalysts. By repurposing these wastes, researchers can create sustainable solutions for environmental remediation, energy conversion, and chemical synthesis applications. However, addressing challenges related to waste characterization, synthesis scalability, and material performance is crucial for realizing the full potential of these materials. Through interdisciplinary collaboration and innovative research efforts, the utilization of chemical industrial and academic wastes as a source of supported photocatalysts can contribute to advancing sustainable chemistry practices and addressing global environmental challenges.

## Acknowledgement

None

\*Corresponding author: Wilsmith Silva, Instituto de Química, Universidade Federal do Rio Grande do Sul, Brazil, E-mail: will@iq.ufrgs.br

**Received:** 01-Jan-2024, Manuscript No ico-24-126486; **Editor assigned:** 04-Jan-2024, PreQC No. ico-24-126486(PQ); **Reviewed:** 18-Jan-2024, QC No. ico-24-126486; **Revised:** 25-Jan-2024, Manuscript No. ico-24-126486(R); **Published:** 30-Jan-2024, DOI: 10.4172/2469-9764.1000268

**Citation:** Silva W (2024) Exploiting Chemical Industrial and Academic Wastes for Supported Photocatalysts: A Potential Source. Ind Chem, 10: 268.

**Copyright:** © 2024 Silva W. This is an open-access article distributed under the terms of the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

## Conflict of Interest

None

## References

1. H Abdel-Wahab (1998) Surfactants Selector a Guide to the Selection of I&I and Household Product Formulations, Akcros Chemicals (now part of Akzo Nobel Surface Chemistry AB).
2. Further formulation information available from Akzo Nobel Surface Chemistry AB, S 444 85 Stenungsund, Sweden, on request.
3. Valappil K, Lalitha S, Gottumukkala D, Sukumaran R K, Pandey A (2015) White Biotechnology in Cosmetics. *Indus BiorefinWhite Biotech* 607-652.
4. Kenneth DK, Stephen JL, Joan SV, Cynthia JB (2015) Solving 21st Century Problems in Biological Inorganic Chemistry Using Synthetic Models. *Acc Chem Res* 48: 2659-2660.
5. Hannah H, Gerlinde G, Christian GH (2016) Electrophoretic separation techniques and their hyphenation to mass spectrometry in biological inorganic chemistry. *Electrophoresis* 37: 959-972.
6. Williams DR (2000) Chemical speciation applied to bio-inorganic chemistry. *J Inorg Biochem* 79: 275-283.
7. Complexing agents Environmental and Health Assessment of Substances in Household Detergents and Cosmetic Detergent Products. Danish Environmental Protection Agency Accessed.
8. Schrodter (2008) Klaus Bettermann Gerhard Staffel, Thomas; Wahl, Friedrich; Klein, Thomas; Hofmann, Thomas Phosphoric Acid and Phosphates. *Ullmann's Ency Ind Chemistry*.
9. Oxford Dictionaries English (2016) glycerol Definition of glycerol in English by Oxford Dictionaries". Archived from the original.