A Cost Effective Strategy for Production of Bio-surfactant from Locally Isolated *Penicillium chrysogenum* SNP5 and Its Applications

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**Abstract**

The current work is enlightened about a cost effective bioprocess using one factor at a time approach for the production of bio-surfactant through solid state fermentation. A fungal strain *Penicillium chrysogenum* SNP5 isolated from grease contaminated soil was reconnoitered for the production of bio-surfactant. Various physiochemical parameters i.e., substrate composition, nitrogen supplements, extraction media and pH were optimized in order to optimized the production in terms of emulsification index and oil displacement assay. Maximum oil displacement area produced using grease waste and wheat bran (1:1 w/w), waste cooking oil and wheat bran (1:1 v/w) as a substrate were 3.5 cm and 5 cm, respectively. Whereas, considered values for emulsification activity with oil and diesel were 43% and 22% during optimization of substrate composition. Variable ratios of grease waste and wheat bran were capable to enhance the emulsification activity with oil and diesel up to 45% and 24% in presence of grease and wheat bran (1:5:1). The strain also showed enhancement of emulsification activity 45% and 23% with oil and diesel respectively to utilized yeast extract as a nitrogen source and the highest emulsification activity 38% in diesel. 47% in oil and oil displacement 5.5 cm was found at pH 8 with green and wheat bran as a substrate. Preliminary characterizations by thin layer chromatography showed that the bio-surfactant was lipopeptide in nature and was also confirmed through FTIR analysis. Metabolization of industrial grease waste through solid state fermentation has never been reported before for the production of bio-surfactants therefore would be applicable in petroleum and biodiesel industry. The partially purified biosurfactants was further investigated for antimicrobial activity and enhanced oil recovery. It displayed effective zones of inhibition against both gram +ve (1.67 cm) and gram –ve (1.93 cm) as well as 16.5% enhanced recovery of oil. Both results also give a positive support to its role in pharmaceuticals as well as in petroleum and oil industry.

**Keywords:** Biosurfactant; Lipopeptides; Solid state fermentation; FTIR; Production and extraction

**Introduction**

Science, mysteries, miracles, inventions and Eureka! all are required for invigoration of a new approach to develop a product. Bio-surfactant (BS) word implies to "A surface active substance which synthesized from microbes to metabolize water insoluble substance such as hydrocarbons and lipids etc." BSs are amphiphilic compounds contain hydrophobic and hydrophilic moieties that reduce surface and interfacial tensions between individual molecules at the surface and interface, respectively. BSs are generally characterized as anionic (usually due to sulphonate or sulphur group), cationic (positively charged quaternary ammonium group) and nonionic (lack their ionic constituents). They may be characterized in terms of molecular weight as high mass containing BS (polymeric and particulate surfactants) or low mass BSs (i.e., lipopeptides, glycolipids and phospholipids etc.) [1,2]. Unlike petroleum based surfactant which are usually non biodegradable thus remain toxic to the environment, biosurfactants have many advantages in terms of biodegradability, biocompatibility, digestibility, low toxicity, availability of raw material for production and surface activeness (can lower down surface tension), which allows their application in cosmetic, pharmaceuticals, as functional food additives, agriculture, medicine, petroleum and industry etc.

Due to this increasing awareness on the need to protect the ecosystem and mankind, necessitate an increased interest in surfactants of microbial origin as possible alternatives to chemically synthesized ones [3,4]. According to a new market report of transparency market research global market of BSs was worth USD 1,735.5 million in 2011 and is expected to attain USD 2,210.5 million in 2018, growing at a CAGR of 3.5% from 2011 to 2018 with leading position of European region sharing 53.3% of global BSs market revenue share [5].

To meet such boost in demand of BS production must be cost effective and at industrial scale. Although scale up process has lacunae due to its high production cost with low rate of production compare to synthetic surfactant. So it is very imperative that our target should be relying on development of successive effort and strategies in such a way that our needs replenish the constancy of nature without any major changes in next generation [6] and environmental sustainability emphasizing, the production of eco-friendly natural bioactive compounds from renewable substrates those have the potential to replace chemically synthesized surfactants [7].

In this concern various microorganisms have been explored for the production of BS and major of studies used submerged fermentation. However, it leads to a main disadvantage of foaming which further has a tendency to accumulate micro-organisms on foam and finally lead to washing out of cells from media. Instead of that, solid state fermentation represents a promising future of production with two main advantages firstly, use of low cost substrate for production and secondly it evades foaming problems [8,9].

The present work is centered on achieving a cost effective and...
Effect of substrate sources for biosurfactant production in terms of EI

<table>
<thead>
<tr>
<th>Substrate</th>
<th>Ratio of substrate (w/w)</th>
<th>EI (oil) (%)</th>
<th>EI (diesel) (%)</th>
<th>ODA(cm)</th>
<th>DCA</th>
</tr>
</thead>
<tbody>
<tr>
<td>Grease waste + wheat bran</td>
<td>1:1</td>
<td>43 ± 0.12</td>
<td>22 ± 0.01</td>
<td>3 ± 0.15</td>
<td>++</td>
</tr>
<tr>
<td>water cooking oil+ wheat bran</td>
<td>1:1</td>
<td>40 ± 0.15</td>
<td>20 ± 0.05</td>
<td>5 ± 0.90</td>
<td>++</td>
</tr>
<tr>
<td>soybeans oil+ wheat bran</td>
<td>1:1</td>
<td>53 ± 0.20</td>
<td>37 ± 0.04</td>
<td>5 ± 0.05</td>
<td>+++</td>
</tr>
<tr>
<td>oil cake + wheat bran</td>
<td>1:1</td>
<td>25 ± 0.62</td>
<td>10 ± 0.90</td>
<td>2 ± 0.01</td>
<td>*</td>
</tr>
</tbody>
</table>

Table 1: Influence of substrate sources for biosurfactant production in terms of EI

EI = emulsion height / total height × 100

Drop collapse assay: Drop collapse assay depends on destabilization of a liquid on a solid surface due to presence of surfactant. Drop of collected extract was placed on an oil coated glass slide and the destabilization of liquid drop was observed against distilled water used as a control [12].

Oil displacement assay: 40 ml of distilled water was added in 90 mm petri plate and then 50 µl of oil was added on surface to make a thin layer after that it was followed by addition of 10 µl of collected extract on middle of surface. Diameter of displaced oil was observed after 30 sec under visual light 10 µl distilled water used as control [13] (Figure 1).

Foam height analysis: 10 ml of collected extract was shaken vigorously for 10 min to attain a good frothing and left to stand for 2 minutes. Foaming capacity was calculated according to the following equation [14]:

Foaming = emulsion height / total height × 100

Optimization of BS extraction

For estimation of the most appropriate solvent system for the extraction of BSs; various buffer from range of 3-10 (sodium citrate buffer, 3-6, sodium phosphate buffer, 7-8, sodium carbonate buffer, 9-10 and water) were employed. BS production was further studied with emulsification assay, oil displacement assay and drop collapse assay.

Partial purification of surfactant

Partial purified extraction of surfactant was done through ethanol precipitation method at -20°C for 16 h. After 16 hour precipitate was collected by centrifugation at 4000×g for 15 min at 4°C [15] resulted pellet was treated with hexane to remove excess grease traces and BS was collected in form of pellet after air drying.

Preliminary compositional analysis

Thin layer chromatography (TLC): A part of crude BS was separated on a TLC Sheet using CHCl₃; CH₃OH; H₂O (65:25:4). v/v/v
as a solvent system with different color developing reagents. Ninhydrin reagent (0.5 g ninhydrin in 100 ml anhydrous acetone) was used to detect lipo-peptidase kind of BS as a red spot [16,17] and iodine vapors was used to develop lipid moiety of surfactant as a yellow spot.

Structural characterization

Lowery assay: Presence of protein concentration was assayed by lowery method with the help of standard curve BSA protein [18].

Fourier transform Infra-red spectra analysis: For detection of functional group IR analysis was done by KBr pellet method, using resolution of 2 cm⁻¹ and 4000-400 cm⁻¹ spectral region [19].

Antimicrobial activity

Anti microbial activity of partially purified BS evaluated using agar diffusion method. 20 ml Muller Hinton Agar media was prepared each for petri plates on which 2 wells were made and were named as A and B respectively. The plates were swabbed with P. aeruginosa and S. aureus. To the wells A, 50 µl of partially purified BS was added, to and to the wells B, distilled water (control) was added. The plates were kept in incubation at 37°C for 24 hours. The presence of clear zone marked the antimicrobial activity of BS. Three readings of the clear zone diameter were taken for each well and the mean was calculated to determine the actual zone diameter [20].

Application of BS in oil recovery

To evaluate the efficacy of crude BS produced by P. chrysogenum SNP5 in oil recovery process, sand pack column as a laboratory scale technique was employed. A glass column height 17 cm was packed with 17 g of acid pretreated sand. The brine solution (5% NaCl, w/v) was then injected and kept it for 24 hours. After 24 hours residual oil was discharged from column. BSs was then injected and kept it for 24 hours. After 24 hours residual oil was discharged with following equations.

\[
\text{Initial oil saturation (Soi \%) = } \frac{\text{OOIP}}{\text{PV}} \times 100
\]

\[
\text{Pore volume (PV) (mL) = Volume of brine required to saturate the column.}
\]

\[
\text{Original oil in place (OOIP) (mL) = Amount of brine solution discharged upon displacement by oil sand pack column.}
\]

\[
\text{Initial water saturation (Swi \%) = } \frac{X}{PV} \times 100
\]

Where X = pore volume- volume of brine displaced after injection of oil in sand pack column.

\[
\text{Residual oil saturation (Sor\%) = } \frac{Xi}{O0IP} \times 100
\]

Where Xi = O0IP- Volume of oil displaced after water flooding

\[
\text{Oil recovery after water flooding (Orecwf) = } \frac{Sorwf}{O0IP} \times 100
\]

\[
\text{Sorwf (mL) = Oil released after the feeding of sand pack column saturated with residual oil}
\]

\[
\text{Additional oil recovery after bio surfactant flooding = }
\]

\[
\frac{\text{oil recovered using bio surfactant flooding}}{\text{oil in column after water flooding}} \times 100
\]

Results

Effect of substrate sources

In order to optimize the production of BSs from P. chrysogenum SNP5, different substrates such as waste glycerol, waste cooking oil, soybean oil and oil cake were used for solid state fermentation and mapped in term of Emulsification Index (EI), Oil Displacement Area in mm (ODA) and drop collapse assay (Figure 1). Although soybean oil as a substrate had shown high emulsification activity 56% with oil and 37% with diesel but grease waste was also found promising alternative due to its low cost and good emulsification activity 43% with oil and 22% with diesel.

Growing on waste cooking oil where displaced oil on a water surface was 5 cm without displaying very effective emulsification. Bio-surfactant produced over grease waste and wheat bran displaced the oil (3.5 cm) and shown a good destabilization of surfactant drop on an oil coated glass surface.

Effect of nitrogen sources

Different nitrogen sources like NaNO₃, yeast extract, peptone and malt extract were used and observed that yeast and NaNO₃ were growth as well as production promoting (Table 2). Other complex sources like peptone and malt extract were good for growth but not suitable for BS production.

Effect of substrate ratio

Grease waste in different ratio with wheat bran (NaNO₃ as a constant nitrogen source) shows an effective change in emulsification index and oil displacement activity as observed in Table 3. The best value of emulsification index and oil Displacement activity was detected at 1.5:1 ratio of grease and wheat bran respectively. A significant decrease in activity of BS was observed as the ratio of grease waste was increased.
iodine vapors with a Rf value 0.69, 0.69, 0.73 respectively in soya bean oil, grease and waste cooking oil. Presence of red color in ninhydrin with Rf values 0.62, 0.68, 0.66 respectively in soya bean oil, grease and waste cooking oil, confirms presence of protein and yellow spot with iodine confirmed the lipid moiety of surfactant (Figure 4).

**Lowery assay**

Protein concentration of surfactant was determined through lowery assay [18]. In 1 mg of partially purified BS the protein concentration was 13% (0.13 mg/ml)

<table>
<thead>
<tr>
<th>pH range</th>
<th>El (oil) (%)</th>
<th>El (diesel) (%)</th>
<th>ODA (cm)</th>
<th>DCA</th>
</tr>
</thead>
<tbody>
<tr>
<td>3</td>
<td>37 ± 0.09</td>
<td>14 ± 0.28</td>
<td>2.2cm</td>
<td>+</td>
</tr>
<tr>
<td>4</td>
<td>40 ± 0.05</td>
<td>19 ± 0.04</td>
<td>3.3 cm</td>
<td>+</td>
</tr>
<tr>
<td>5</td>
<td>4 ± 0.30</td>
<td>22 ± 0.07</td>
<td>3.9 cm</td>
<td>+</td>
</tr>
<tr>
<td>6</td>
<td>45 ± 0.03</td>
<td>23 ± 0.80</td>
<td>4.1m</td>
<td>++</td>
</tr>
<tr>
<td>7</td>
<td>45 ± 0.09</td>
<td>23 ± 0.9</td>
<td>4.8cm</td>
<td>***</td>
</tr>
<tr>
<td>8</td>
<td>47 ± 0.61</td>
<td>38 ± 0.09</td>
<td>5.5</td>
<td>***</td>
</tr>
<tr>
<td>9</td>
<td>45 ± 0.07</td>
<td>30 ± 0.45</td>
<td>5.3</td>
<td>***</td>
</tr>
<tr>
<td>10</td>
<td>44 ± 0.10</td>
<td>25 ± 0.03</td>
<td>4.7</td>
<td>++</td>
</tr>
<tr>
<td>11</td>
<td>43 ± 0.01</td>
<td>18 ± 0.18</td>
<td>4.1</td>
<td>++</td>
</tr>
<tr>
<td>12</td>
<td>38 ± 0.09</td>
<td>15 ± 0.03</td>
<td>3.4</td>
<td>+</td>
</tr>
</tbody>
</table>

Table 4: Influence of different pH values for production of biosurfactant in terms of El: Emulsification Index; ODA: Oil Displacement Assay; DCA: Drop Collapse Assay; (+: less dispersed drop; ++: moderate dispersed drop; +++ highly dispersed drop of surfactant).

**Effect of pH on production of BS**

The production of BS was tested over a wide range of pH (3-12) of medium with wheat bran and grease waste (1:1) and 1% NaNO₃ as nitrogen source. Growth was observed at all condition but performance of surfactants was observed with significant variations under different pH ranges (Table 4).

Surfactant shows the changes in the activity of emulsion formation in different pH ranges a positive increase was observed from 2 to 8 (maximum activity with E24, 47.61% in oil and 38.9% in diesel (Figure 2) and further some significant reduction was observed from pH range 8-12.

**Foam height analysis:**

Foaming ability of BS was determined 23% in 10 ml crude surfactant. To improve this observation further stability of foam was continuously monitored after every 5 minutes for 1 hr and after 1 hour 3% reduction was found in initial foam height (2 m) (Figure 3).

**Effect of solvent system for extraction of BS:**

Another aspect which was fundamental to analysis of suitable solvent for collection of cell free extracts (a source of crude BS) from fermented bran was application of various buffers (pH 3-10) and their comparative study with water as a solvent system. From Table 5, it was again expressed reduction but overall it cumulative effects was very low compare to water as a solvent with maximum activity 43% in case of oil and 22.06% in case of hydrocarbon.

**Analysis through thin layer chromatography:**

Characterization of BS collected by precipitation method was determined by TLC method. A red color spot was observed after the use of ninhydrin reagent and a yellow color spot was developed by using grease waste and wheat bran (1:1) which previously found for BS extraction. Obtained results were shown the dependency of all buffer system was depicted as week solvent comparative to water for collection of cell free extracts (a source of crude BS) from fermented bran was application of various buffers (pH 3-10) and their comparative study with water as a solvent system. From Table 5, it was again expressed reduction but overall it cumulative effects was very low compare to water as a solvent with maximum activity 43% in case of oil and 22.06% in case of hydrocarbon.

**Table 2: Influence of nitrogen content (%) for biosurfactant production in terms of EI: Emulsification Index; ODA: Oil Displacement Assay; DCA: Drop Collapse Assay; (+: less dispersed drop; ++: moderate dispersed drop; +++ highly dispersed drop of surfactant).**

<table>
<thead>
<tr>
<th>Substrate concentration (Wheat bran and grease ratio)</th>
<th>El (oil) (%)</th>
<th>El (diesel) (%)</th>
<th>ODA (cm)</th>
<th>DCA</th>
</tr>
</thead>
<tbody>
<tr>
<td>1:1</td>
<td>43 ± 0.30</td>
<td>22 ± 0.07</td>
<td>3 ± 0.17</td>
<td>++</td>
</tr>
<tr>
<td>1.5:1</td>
<td>45 ± 0.06</td>
<td>24 ± 0.01</td>
<td>4 ± 0.30</td>
<td>+++</td>
</tr>
<tr>
<td>2:1</td>
<td>43 ± 0.10</td>
<td>18 ± 0.50</td>
<td>2 ± 0.2</td>
<td>++</td>
</tr>
<tr>
<td>2.5:1</td>
<td>40 ± 0.22</td>
<td>18 ± 0.03</td>
<td>1 ± 0.72</td>
<td>+</td>
</tr>
<tr>
<td>3:1</td>
<td>37 ± 0.13</td>
<td>15 ± 0.04</td>
<td>1 ± 0.51</td>
<td>+</td>
</tr>
</tbody>
</table>

Table 3: Effects of substrate ratio for biosurfactant production in terms of EI: Emulsification index; ODA: Oil Displacement Assay; DCA: Drop Collapse Assay; (+: less dispersed drop; ++: moderate dispersed drop; +++ highly dispersed drop of surfactant).

**Table 2: Influence of nitrogen content (%) for biosurfactant production in terms of EI: Emulsification Index; ODA: Oil Displacement Assay; DCA: Drop Collapse Assay; (+: less dispersed drop; ++: moderate dispersed drop; +++ highly dispersed drop of surfactant).**

<table>
<thead>
<tr>
<th>Nitrogen (1%)</th>
<th>El (oil) (%)</th>
<th>El (diesel) (%)</th>
<th>ODA (cm)</th>
<th>DCA</th>
</tr>
</thead>
<tbody>
<tr>
<td>NaNO₃</td>
<td>43 ± 0.15</td>
<td>22 ± 0.09</td>
<td>3 ± 0.15</td>
<td>++</td>
</tr>
<tr>
<td>Peptone</td>
<td>40 ± 0.10</td>
<td>20 ± 0.12</td>
<td>3 ± 0.01</td>
<td>++</td>
</tr>
<tr>
<td>Yeast</td>
<td>45 ± 0.61</td>
<td>23 ± 0.13</td>
<td>4 ± 0.75</td>
<td>+++</td>
</tr>
<tr>
<td>malt</td>
<td>37 ± 0.23</td>
<td>17 ± 0.21</td>
<td>2 ± 0.93</td>
<td>+</td>
</tr>
</tbody>
</table>

**Table 3: Effects of substrate ratio for biosurfactant production in terms of EI: Emulsification index; ODA: Oil Displacement Assay; DCA: Drop Collapse Assay; (+: less dispersed drop; ++: moderate dispersed drop; +++ highly dispersed drop of surfactant).**
**Table 5**: Optimization of different solvent system for extraction of biosurfactant from fermented substrate in terms of EI: Emulsification Index; ODA: Oil Displacement Assay; DCA: Drop Collapse Assay; (+: less dispersed drop; ++: moderate dispersed drop; +++: highly dispersed drop of surfactant).

<table>
<thead>
<tr>
<th>Solvent system</th>
<th>EI (oil) (%)</th>
<th>EI (diesel) (%)</th>
<th>ODA (in cm)</th>
<th>DCA</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sodium citrate buffer 3</td>
<td>27 ± 0.11</td>
<td>9 ± 0.17</td>
<td>1 ± 0.40</td>
<td>+</td>
</tr>
<tr>
<td>Sodium citrate buffer-4</td>
<td>28 ± 0.20</td>
<td>10 ± 0.30</td>
<td>1 ± 0.1</td>
<td>+</td>
</tr>
<tr>
<td>Sodium citrate buffer 5</td>
<td>30 ± 0.05</td>
<td>12 ± 0.51</td>
<td>0 ± 0.95</td>
<td>+</td>
</tr>
<tr>
<td>Sodium citrate buffer 6</td>
<td>33 ± 0.12</td>
<td>13 ± 0.61</td>
<td>1 ± 0.70</td>
<td>+</td>
</tr>
<tr>
<td>Sodium carbonate buffer 7</td>
<td>35 ± 0.19</td>
<td>15 ± 0.21</td>
<td>2 ± 0.4</td>
<td>+</td>
</tr>
<tr>
<td>Sodium carbonate buffer 8</td>
<td>29 ± 0.05</td>
<td>12 ± 0.05</td>
<td>1 ± 0.05</td>
<td>+</td>
</tr>
<tr>
<td>Sodium carbonate buffer 9</td>
<td>19 ± 0.01</td>
<td>10 ± 0.02</td>
<td>nd</td>
<td>nd</td>
</tr>
<tr>
<td>Sodium carbonate buffer 10</td>
<td>13 ± 0.08</td>
<td>8 ± 0.14</td>
<td>nd</td>
<td>nd</td>
</tr>
<tr>
<td>Water</td>
<td>43 ± 0.31</td>
<td>22 ± 0.06</td>
<td>3 ± 0.19</td>
<td>++</td>
</tr>
</tbody>
</table>

Application of BS for oil recovery

While applying 0.6 volume of cell free BS to the oil saturated sand column, to recover oil trapped (43.2%) in the column, 16.5 ± 2 AOR was achieved upon 24 h of incubation (Table 7).

Discussion

On the basis of results obtained in current study it can be stated that production of BS through solid state fermentation is much advantageous than submerged fermentation. Especially in case of BS where SSF emphasize to less energy and cost requirement and very important foaming nuisance in fermentation system, which generally found as drawback in Submerged Fermentation (SmF). Various types of strains of bacteria and fungus have been reported for the efficient production of secondary metabolites under SSF by many scientists [21-24].

In present work production and effectiveness of BS was tried to enhance through optimizing various physiochemical conditions and selecting appropriate media for SSF. However results were promising using wheat bran and soyabean oil (emulsification activity 56% with oil and 37% with diesel) but for further study wheat bran and grease waste (emulsification activity 43% with oil and 22% with diesel) was considered in order to find out good alternative tool for remediation of oil spills and hydrocarbon contaminated soil sites etc. By using hydrocarbon based grease waste as substrate for production, BS produced might be specific for petroleum products removal or

**Antimicrobial activity of BS**

The results of antimicrobial activity of lipopeptide produced by *P. chrysogenum* SNP5 against pathogenic bacteria *S. aureus* and *P. aeruginosa* are shown in table 6. Presence of clear halos indicating that antimicrobial activity of the compound against gram +ve (*S. aureus*) and gram-ve (*P. aeruginosa*) bacteria with a hallo diameter of mean values 1.67 cm and 1.93 cm respectively (Table 6).

**Table 6**: Antimicrobial activity with partially purified biosurfactant.

<table>
<thead>
<tr>
<th>S.No.</th>
<th>Name of strain</th>
<th>Zone of inhibition (cm)</th>
<th>Mean value</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td><em>S. aureus</em></td>
<td>1.6</td>
<td>1.67</td>
</tr>
<tr>
<td>2</td>
<td><em>P. aeruginosa</em></td>
<td>2.0</td>
<td>1.93</td>
</tr>
</tbody>
</table>

**Table 7**: Parameters for oil recovery using water flood and lipopeptide bio surfactant from sand pack glass column.

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Sand Pack Column (SPC)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>SPC1</td>
</tr>
<tr>
<td>PV (ml)</td>
<td>5.5</td>
</tr>
<tr>
<td>OORP (ml)</td>
<td>4</td>
</tr>
<tr>
<td>Sorwf (ML)</td>
<td>1.7</td>
</tr>
<tr>
<td>Sorfb(ml)</td>
<td>0.6</td>
</tr>
<tr>
<td>Swi (%)</td>
<td>72</td>
</tr>
<tr>
<td>Soi (%)</td>
<td>27</td>
</tr>
<tr>
<td>Oreocwf (%)</td>
<td>42</td>
</tr>
<tr>
<td>AOR (%)</td>
<td>18</td>
</tr>
</tbody>
</table>

FTIR analysis

The FTIR spectra of partially purified BS are shown in figure 5. Characterization absorption peak were found at 3398 cm⁻¹ ranging from 3100 cm⁻¹ to 3600 cm⁻¹. This feature typically confirmed presence of carbon and amino groups and caused due to stretching vibrations of C-H and N-H bonds characterized to carbon containing amino group. Two other sharp bands at 2926 cm⁻¹ and 2850 cm⁻¹ showed presence of long alkyl chain (-CH₂- and -CH₃-). The medium band near 1637 cm⁻¹ attributed to CO·NH bend which confirms presence of peptide group in the 286 molecule. Other significant peaks observed at 1384 cm⁻¹ correspond to C-H vibrations. A medium stretch peak around 1084 cm⁻¹ signifies to presence of C-O-C stretch of ester. And at last a sharp peak at the region around 669 cm⁻¹ and 770 cm⁻¹ also represented -CH₂ bending of alkenes.

**Figure 5**: FTIR spectra of partially purified biosurfactant.
recovery, whereas soybean oil is lipid based. Yeast extract as nitrogen source has shown good impact (displacement of oil is 4 cm and destabilization of surfactant drop 1.4 cm compare to water was 0.4 cm on an oil coated glass surface) on BS production, however NaN03 was also found enhancer (displacement of oil 3.5 cm and destabilization of surfactant drop 1.1 cm in compare to water 0.4 cm on an oil coated glass surface). Yeast extract being complex and mix kind of nitrogen source might have served as fortifying rather than NaN03. Selection of substrates and nitrogen sources has major effect on the production of BS [6,11]. *P. chrysogenum* showed different preferance on complex and simple nitrogen sources and substrates for BS production [4].

While optimizing the ratio of Wheat Bran (WB) and grease waste for production it was observed that with increase ratio of Grease Waste (GW), emulsification activity reached up to 45% with oil and 24% with diesel at a ratio of 1.5:1 w/w of grease waste and wheat bran respectively without use of other complex nitrogen sources. But with more than this ratio reduction in emulsification index and oil displacement activity from 22-15% and 3-1 cm respectively was observed due to its complexity in nature and hard to utilize completely.

Use of various buffer as a solvent systems showed high impact over recovery of crude BS and found water as best solvent in compare to other solvents. Similar findings were also observed by other scientist [22]. However, in this study the optimization of BS production was confirmed through some indirect methods like Emulsification index, Oil displacement assay and Drop collapse. These are very sensitive and effective methods to confirm the production of BS [11,23]. Abuoseoud, et al. [11] had optimized the carbon source, nitrogen source and C/N ratio for bio surfactants production on the basis of emulsification index [11]. In similar manner bio surfactant production optimization was reported by Kiran et al. [23], where optimization of BS production by *Nocardiopsis lucentensis* MSA04 with various carbon sources in solid-state cultivation had been performed by using Emulsification index reported of about E24=25% with wheat bran as substrate [23]. Colla et al. [24] have studied simultaneous production of lipases and BSs by submerged and solid-state bioprocesses with a fungal strain *Aspergillus* sp. and they reported that the production of BS was very low in SSF (emulsification activity after 24 hours was 2.85%) compare to SmF (emulsification activity after 24 hours was 42.67%). Therefore this current study with *P. chrysogenum* SNP5 by using grease waste as substrate for BS production with their optimized emulsification value 43% would be a good benchmark among already reported BSs and also can be accentuate Solid state cultivation of BSs production. And using grease waste as substrate for production make the process cost effective, sustainable and innovative.

BS obtained after extraction was precipitated out and characterized through thin layer chromatography. Results confirmed that BS produced by *P. chrysogenum* SNP5 by using grease waste as substrate was lipopeptide in nature. Presence of peptide and lipid was confirmed on the basis of Rf value 0.68 and 0.69 obtained with ninhydrin and iodine respectively during TLC.

Earlier study confirmed the TLC data with the Rf value of 0.68 and 0.70 for lipopeptide in case of iodine treatment [25] and in case of lipopeptide isolated from *Kocuria marina* BS-15 also confirmed Rf value 0.68 [26]. It was again verified from FTIR spectrum of BS which has shown vibration at 2926 cm⁻¹ and 2850 cm⁻¹ and 1384cm⁻¹ suggested the presence of aliphatic chain of fatty acids, at 3398 cm⁻¹ indicating presence of C-H and N-H bonds stretching of peptides. A vibration near 1647 cm⁻¹ showed the presence of CO-NH bond and vibration near 1082 cm⁻¹ signified presence of C-O-C stretch of ester. The FTIR analysis was quite similar to result obtained by Thavasi et al. [27] They found spectra at 2,852, 2,923, 1,421, and 1,465 cm⁻¹ for the C-H stretching mode. 3,383 and 1,647 cm⁻¹ for C-O and N-H vibration and the C-O ester bonds were observed at 1,058 cm⁻¹ [27].

Surfactant produced by *P. chrysogenum* SNP5 was lipopeptide in nature, hence pH, solvent and buffer have shown high impact on its activity, stability and emulsification efficiency etc. due to presence of free NH2, C=O and COOH group. It can be characterized further for its molecular structure in detail.

Lipopeptides are commercially known for their antimicrobial activity against Multi Drug Resistance (MDR) strains [20]. The partially purified lipopeptide was shown good antimicrobial activities of 1.67cm and 1.93 cm against gram +ve (*S. aureus*) and gram –ve (*P. aeruginosa*) respectively. This is little contrasting to previous reports where lipopeptides have been reported mostly against gram +ve bacteria [28]. The current scenario of pathogenic strain against drug resistance depends upon indiscriminate use of regular common antibiotics. Both strain are a very common example of MDR and generally show resistance against Methicillin and Streptomycin. Hence these new lipopeptide produced from *P. chrysogenum* SNP5 can gives an alternative to already existing antimicrobial drugs [6]. So finally BS produced due to its light molecular weight and lipopeptide in nature would be most desirable and useful in food, pharmaceutical like, medicine, agriculture industries and for bioremediation purpose like oil tanks clean up [29-31].

BS produced as crude extract was also tested for it application in oil recovery which is very challenging and costly efforts in petroleum and oil industries. And it was found very efficient even in crude form for recovery of trapped oil in column. Hence could be very useful in petroleum and oil industries, also it can be applied for pipeline washing due to its both the properties i.e., antimicrobial property to check microbial corrosion and surface activity for washing out retained oil from the surface.

**Conclusion**

In present study SSF using grease waste as substrate for BS production had been found promising due to its cost effectiveness and single step downstream processing, BS produced had shown an effective emulsification and oil displacement activity against crude oil as well as it has shown a potential antimicrobial activity also. Therefore, BS produced may be utilized towards the solid waste management, in oil recovery and in food processing. The BS produced by *P. chrysogenum* SNP5 was characterized as lipopeptide derivative, which is different from the liposaccharide BS produced by *P. citrium*, reported by Demorais et al. [15] and finally create a potential alternative over the various other chemical surfactants. Now a day’s solid waste management is a critical issue and this strain or its surface active compounds can be used as a good alternative for biostimulation or bioaugmentation respectively. Further study may open new horizon for industries.

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**References**


