

Dissipation Behavior of ¹⁴C-Deltamethrin in Egyptian Soil Columns in Absence and Presence of Chlorpyrifos under Subtropical Conditions

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Abstract

The dissipation of deltamethrin in silt clay under subtropical climate conditions was studied.¹⁴C- Deltamethrin labeled at gem–dimethyl groups of cyclopropane ring was used alone in one sets of columns and in presence of chloropyrifos in the other sets and kept under field conditions for eleven months. Radioactivity was concentrated in the upper zone of the columns 0-10 cm. The total dissipation of the insecticide amounted to 31%; 52% of the applied dose in silt clay soil in absence and presence of chloropyrifos, respectively. The soil binding capacity increased with time whereby the extractable ¹⁴C-residues simultaneously decreased. The nature of methanolic extract of soils was determined by chromatographic analysis. Deltamethrin was the main degradation product of the extractable residues in addition to 3-phenoxybenzaldehyde, 3-phenoxybenzoic acid and 3-phenoxybenzyl alcohol.

Keywords: ¹⁴C-Deltamethrin; Chloropyrifos; Soil columns; Dissipation

Introduction

The persistence of pesticides in soil has both economic and environmental significance and is often used as a key parameter in pesticide risk assessment. An appreciable proportion of the insecticides applied to plants often reaches the soil and eventually become incorporated into it [1]. Dispersion of pesticides and their transformation products within the soil, or from the soil to the environment, is influenced not only by the properties of the pesticides and the soil but also by the climatic conditions. Soil properties known to influence the persistence of pesticides include soil pH [2] moisture [3] organic matter content [4-6], redox status [7], and mineral constituents [8]. The synthetic pyrethroids are among the most potent and effective insecticides, accounting for more than 30% of the world market of insecticides. The low toxicity of these insecticides to mammals, birds and their limited soil persistence has encouraged their widespread use in agriculture [9,10] as a substitute for more toxic or recalcitrant organochlorines or organophosphates [11]. The combination of insecticides from different groups may overcome resistance of insects to pesticides. Several organophosphate insecticides have been mixed with pyrethroids [12,13]. For example, pyrinex quick (deltamethrin+chloropyrifos) has a reduced volatility, inhalation toxicity, phytotoxicity and is safe for the operators. Grain protectants including fenitrothion, malathion, pirimifosmethyl (organophosphorus compounds), deltamethrin (a synthetic pyrethroid), diatomaceous earth (DE), had proven effective against grain insects when used alone and in combination [13-15] Deltamethrin[(S)a-cyano-3-phenoxybenzyl-(1R,3R)-3-(2,2-dibromo-vinyl)-2,2dimethylcyclopropane-1-carboxylate] is a synthetic a-cyano pyrethroid insecticide. This insecticide is considered to be useful insecticide due to its high insecticidal activity with low mammalian toxicity and limited soil persistence [16-18]. Most investigations done efforts focused on the photochemistry and metabolism of deltamethrin in plants and animals. Limited information is available about the fate and movement of deltamethrin and its residues in soil, this is probably due to its extremely low solubility in water [19-21]. Chloropyrifos, a broad spectrum chemical, is the most extensively used organophosphate insecticide in agriculture [22]. It is registered for the control of soil insects and some foliar insects on a wide range of economic crops, as well as for house hold use [23].

The aim of the present work is to study the movement, degradation and dissipation of ¹⁴C-deltamethrin in absence and presence of the organophosphorus insecticide.

Materials and Methods

Soil

A silt clay soil having different characteristics were used in this study (clay 33.60%; silt 55.15%; sand 11.25%; organic matter 0.95%; pH7.78). The used soil was air-dried and passed through a 2-mm sieve before use. The soil was stored below 0°C until use. At the start of the experiment, soil was thawed and air dried overnight.

Experimental Set-up

Hard polyvinylchloride cylinders (PVC) (50 cm length×5 cm.i.d.) open at both ends, were used for the present study. Columns were inserted in soil in the field, two weeks prior to applications. Soil columns received regular inputs of fertilizer and were kept moist by adding water when necessary, and left in the open air under field conditions. Two sets of PVC columns were used; the radiochemical was appropriately diluted with the pure compound. Each column was spiked at the top of the soil column using a micro pipette with 10 mg of the diluted preparations /kg soil (containing) (1.0 μ Ci) of the radiochemical dissolved in 100 μ l of water or acetone and two sets of columns received the above dose of ¹⁴C-deltamethrin in addition to 50 mg of pure chloropyrifos. The experiment was run over 11 successive months.

Chemicals

¹⁴C-Deltamethrin radiolabeled at gem-dimethyl groups of the cyclopropane ring was supplied by Roussel-Uclaf, Paris, France. It had a specific activity 695.6MBq/mg and a radiometric purity of 98%. Pure- non labelled deltamethrin and its main degradation

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products were synthesized according to known procedures [24]. Pure chlorpyrifos, on the other hand was prepared according to Fakhr et al [25]. All chemicals used in this study were of analytical grade. Pure deltamethrin was synthesized by adding equimolecular amounts of 3-phenoxybenzaldehyde cyanohydrins (A) and dibromo acid chloride(B) in tetrahydrofuran and stirring for 24 h at room temperature, followed by heating at 40°C for 6 h and then left to cool at room temperature. 1, 4-Dioxane and saturated aqueous sodium carbonate were added and the mixture was stirred for 1h. The mixture was extracted with ethyl acetate, washed with water, dried over anhydrous magnesium sulphate, and concentrated in a vacuum. The solid residue was crystallized from ethanol to give white crystals (1) m.p.100-102 C°; (Scheme 1). Pure chlorpyrifos was prepared,by condensation of thiophosphoryl chloride with two equivalents of ethanol in presence of pyridine as acid binding reagent to give O,Odiethyl phosphorochloridothioate as colourless oil. The latter was condensed with an equimolecular amount of 3,5,6-trichlorpyridinol for 3 h at room temperature in presence of sodium carbonate and freshly distilled dimethylformamide to give chlorpyrifos (O,O-diethyl-O-3,5,6-trichloro-2-pyridyl phosphorothioate) which was crystallized from methanol to give pure white crystals which melted at 42-43°C. as shown in scheme 1.

Sampling and extraction

Samples were taken at various intervals over 11 months at 1, 30, 60, 90, 120, 150, 180, 210, 240, 270, 300 and 330 days after treatment with insecticides. Each time, the columns were removed and sliced into 0-5, 5-10, 10-20, 20-30, and 30-45 cm sections carefully at random for analysis. Soil was air dried, mixed thoroughly and then three subsamples were analyzed from each section of column. Soil was usually stored at below 0°C till analyzed. Samples ($30g \times 3$) from each section of columns were extracted in a Soxhlet apparatus with methanol for 6 hours. Further extraction of the soil gave no further extractable residues of the pesticide.

Analysis of Extractable ¹⁴C-Residues

The nature of ¹⁴C-residues in the methanolic extract was determined by thin layer chromatography (TLC) and high performance liquid chromatography (HPLC) of samples taken over 11 months following application of the radioactive insecticide. The TLC analysis was conducted on precoated silica-gel plates (F-254, Merck, Germany) with 25 mm gel thickness. Authentic substances were run alongside as references. Spots of deltamethrin and its metabolities were made visible on developed plates by using UV lamp at 254 nm and made visible by spraying with Iodine [26].Deltamethrin and its metabolities appear as yellowish brown spots against white background. The following systems were used for development:

System A: n-Hexane-ether (4:1, v/v) System B: Benzene-chloroform (1:1, v/v)

System C: Pet.Ether₆₀₋₈₀ – Ethyl acetate (5:1, v/v).

The HPLC analysis was performed on a Waters-Association Model 510 equipped with a U6K Loop injector and a Model 484 variable wave length detector was used. For the identification, 10 $\mu C_{_{18}}$ columns protected with a pre -column filter was used. HPLC grade acetonitrile and water were degassed in an ultrasonic bath just before use. The mobile phase consisted of acetonitrile and water, (8: 4, v/v), (System D) and the flow rate was 1 ml/min. The detection was set by monitoring the UV absorbance at λ 190 nm.

Radioactivity measurement

The obtained extracts (1mL x3) were mixed with dioxane-

based Scintillator and determined in a liquid scintillation counter (LSC) [27]. Bound ¹⁴C-residues in extracted soil were determined by combustion (100mg) in a Harvey Biological Oxidizer (OX-600). For quench correction, an internal standard technique was used. For radio scanning of TLC plates, the plates were scraped at 1cm increments into vials, mixed with scintillation cocktail and counted by LSC. For each column, the overall residue load, and hence persistence, was calculated by summing increment loads of the column.

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Results and Discussion

Dissipation of 14C-Deltamethrin in Silt Clay Soil

The data given for ¹⁴C-deltamethrin dissipation in silt clay soil over 11 months using in-situ soil columns are shown in Table 1. The initial concentration of 14C-deltamethrin was found to be 98.5% after one day of application (zero time). Most of the radioactivity was concentrated in the upper zone (0-5 cm) in control columns. This layer contained about 89 and 67% after 6 and 11 months, respectively. The amount of extractable ¹⁴C-deltamethrin residues in the upper 0-5cm zone decreased from 98.5% at zero time to 94.5, 86.7 and 52% after 3, 6 and 11 months, respectively. After 10 months, a trace of extractable insecticide residues (0.44%) was observed in the 5-10cm zone. The amount of bound residues (non-extractable) showed a gradual slow increase during the time of experiment. The binding became clear after 4 months especially in the upper zone (0-5cm) and reached its maximum 9% and 14% after 10 and 11 months, respectively. The recovery percent ranged between 67-98.5% in the upper 0-10 cm zone. The percentage of dissipation represented 31% of the total applied dose at the end of the experiment (Table 1).

-Radioactivity applied per column =1 μ Ci (37 MBq) +10 mg non-labelled deltamethrin /kg silt clay soil (control). Results are mean of two columns

On the other hand, the presence of chlorpyrifos insecticide mixed with deltamethrin in silt clay soil columns, led to a faster downward movement of deltamethrin. The percentage of dissipation represented 52% of the total applied dose in 11 months as shown in Table 2 .The extractable deltamethrin residues in the upper 0-5 cm zone decreased from 97.8% at zero time to 19%, after 11 months. After 3 months, a trace of extractable insecticide residues (1.2%) was observed in the 5-10 cm zone and reached its maximum (10%) at the end of the experiment. The binding became clear after 3 months (2%) and reached (15 and17%) in the 0-10 cm zone at 10 and 11 months reached respectively. After 6 months, a trace of the insecticide residues was observed in 10-15 cm zone.

Radioactivity applied per column =1 μ Ci (37 MBq) +10 mg non-labelled deltamethrin + 50 mg non-labeled chloropyrifos /kg silt clay soil (treated)

Results are mean of two columns

The obtained data as shown in Tables 1 and 2, revealed that deltamethrin dissipated more readily in the presence of chlorpyrifos. These data are in agreement with other reported findings where deltamethrin was immobile in soil that had a higher organic matter content and / or higher clay content than sand [28] Moreover 93% of the applied deltamethrin remained within the 0-2cm layer, and only 0.5-1% of the applied dose could be extracted from the subsequent lower depths 2-6cm layers [29-32]. It is known that the pyrethroid insecticide deltamethrin is classified as a low mobile compound in soil according to the pesticide mobility classification system [28]. Kaufman et al. [33] found that deltamethrin was practically immobile in soil

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Recovery, %	¹⁴ C-Residue concentrations							Sempling Date
	Total		Bound		Extractable		(Cm)	(month)
	%	μ±S.D	%	µg ± S.D	%	µg± S.D		. ,
98.51	98.51	9.851 ± 0.0	0.01	0.001 ± 0.00	98.50	9.85 ± 0.00	0-5	1 day
98.10	98.10	9.81 ± 0.20	0.12	0.012 ± 0.00	98.00	9.80 ± 0.02	0-5	1
97.29	97.29	9.729 ± 0.06	0.29	0.029 ± 0.0	97.00	9.70 ± 0.30	0-5	2
95.11	95.11	9.511 ± 0.03	0.61	0.061 ± 0.0	94.50	9.45 ± 0.09	0-5	3
92.85	92.74 0.11	9.274 ± 0.02 0.011 ± 0.00	0.94 0.01	0.094 ± 0.01 0.001 ± 0.0	91.80 0.08	9.18 ± 0.07 0.008 ± 0.00	0-5 5-10	4
89.25	89.20 0.03	8.92 ± 0.05 0.003 ± 0.00	5.0 0.03	0.50 ± 0.02 0.003 ± 0.0	84.20 0.03	8.42 ± 0.04 0.003 ± 0.00	0-5 5-10	5
88.67	88.42 0.25	8.842 ± 0.04 0.025 ± 0.00	1.72 0.08	0.172 ± 0.01 0.008 ± 0.00	86.70 0.17	8.67 ± 0.07 0.017 ± 0.01	0-5 5-10	6
84.79	84.63 0.16	8.463 ± 0.06 0.016 ± 0.00	5.73 0.04	0.573 ± 0.03 0.004 ± 0.0	78.90 0.12	7.89 ± 0.06 0.012 ± 0.00	0-5 5-10	7
82.18	81.93 0.256	8.193 ± 0.07 0.0256 ± 0.0	7.13 0.056	0.713 ± 0.03 0.0056 ± 0.0	74.80 0.20	7.48 ± 0.08 0.020 ± 0.00	0-5 5-10	8
77.70	77.25 0.451	7.725 ± 0.04 0.0451 ± 0.0	8.45 0.061	0.845 ± 0.005 0.0061 ± 0.0	68.80 0.39	6.88 ± 0.003 0.039 ± 0.00	0-5 5-10	9
71.74	71.22 0.527	7.122 ± 0.06 0.0527 ± 0.01	9.32 0.087	0.932 ± 0.03 0.0087 ± 0.0	61.90 0.44	6.19 ± 0.04 0.044 ± 0.00	0-5 5-10	10
67.14	66.36 0.78	6.736 ± 0.03 0.078 ± 0.01	13.96 0.17	1.396 ± 0.05 0.017 ± 0.0	52.40 0.61	5.24 ± 0.02 0.061 ± 0.001	0-5 5-10	11

Table 1: Dissipation of ¹⁴C-Deltamethrin only in silt clay soil columns during 11 months

columns, where 97 % of the ¹⁴C-activity remained in the upper 0 - 2.5 cm layer and these results are in line with the obtained results. Its immobility in the environment is attributed to its strong adsorption on particles and its insolubility in water. [28-32] On the other hand, the bound ¹⁴C-residues increased gradually as time of incubation increased due to hydrophobic property of pyrethroids, thus leading to formation of bound residues [28-34]

from the sampled top soil layer, and enhanced microbial / chemical degradation due to variable soil temperature and moisture conditions. The soil surface photolysis may, also accelerate the dissipation of the pesticide under field conditions [35]. Also, the soil microorganisms are believed to play an important role in the release and further degradation of bound residues of organochlorine (DDT) [36,37].Organophosphorus (chlorpyrifos) [38] and herbicides (prometryn) [39].

Generally, it is found that the loss in the amount of recovery is probably caused by surface runoff transport, leaching of the pesticide In the presence of chlorpyrifos, the percentage of the obtained insecticide residues is 46% of the applied dose in silt clay (Tables 2)

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Sampling Date (month)		¹⁴ C-deltamethrin residues						
	Column Zone	Extractable		Bound		То	Total	
	(Cm)	μg ± S.D	%	μg ± S.D	%	μg ± S.D	%	
1 day	0 – 5	978 ±0.01	97.8	0.02 0.00	0.2	9.80 ± 0.01	98.0	98.00
1 -	0 – 5	9.35 ± 0.05	93.5	0.06 ± 0.01	0.6	9.41 ± 0.06	94.1	04.70
	5 – 10	0.06 ± 0.01	0.6	0.002 ± 0.03	0.02	0.062 ± 0.09	0.60	94.70
2	0 – 5	8.81 ± 0.03	88.1	0.08 0.04	0.8	8.89 ± 0.05	88.9	00.20
	5 – 10	0.08 ± 0.02	0.8	0.06 ± 0.05	0.6	0.14 ± 0.06	1.4	90.30
3	0 – 5	8.51 ± 0.03	85.1	0.1 0.03	1.0	8.61 ± 0.01	86.1	00.00
	5 – 10	0.12 ± 0.01	1.2	0.09 ± 0.02	0.9	0.21 ± 0.00	2.1	88.20
	0 – 5	7.80 ± 0.06	78.0	0.30 0.01	3.0	8.1 ± 0.04	81.0	95.00
4	5 – 10	0.30 ± 0.05	03.0	0.10 ± 0.00	1.0	0.4 ± 0.01	4.0	65.00
5	0 – 5	7.10 ± 0.04	71.0	0.45 0.01	4.5	7.55 ± 0.05	75.5	04.50
	5 – 10	0.4 ± 0.01	4.0	0.20 ± 0.02	2.0	0.6 ± 0.04	6.0	81.50
6	0 – 5	6.50 ± 0.02	65.0	0.52 0.03	5.2	7.02 ± 0.01	70.2	
	5 – 10	0.4 ± 0.03	4.0	0.25 ± 0.01	2.5	0.65 ± 0.06	6.5	76.71
	10 – 15	0.001 ± 0.01	0.01	-	-	0.001 ± 0.01	0.01	
	0 – 5	5.82 ± 0.05	58.2	0.65 0.01	6.5	6.47 ± 0.05	64.7	72.64
7	5 – 10	0.50 ± 0.03	5.0	0.29 ± 0.03	2.9	0.79 ± 0.03	7.9	
	10 – 15	0.003 ± 0.04	0.03	0.001 ± 0.04	0.01	0.004 ± 0.01	0.04	
8	0 – 5	5.0 ± 0.06	50.0	0.75 0.02	7.5	5.75 ± 0.04	57.5	67.56
	5 – 10	0.7 ± 0.01	7.0	0.30 ± 0.03	3.0	1.00 ± 0.06	10.0	
	10 – 15	0.005 ± 0.02	0.05	0.001 ± 0.01	0.01	0.006 ± 0.05	0.06	
9	0 – 5	3.8 ± 0.01	38	0.83 0.05	8.3	4.63 ± 0.01	46.3	58.40
	5 – 10	0.8±0.00	8.0	0.40 ± 0.04	4.0	1.20 ± 0.04	12.0	
	10 – 15	0.009 ± 0.02	0.09	0.002 ± 0.01	0.02	0.011 ± 0.03	0.1	
10	0 – 5	2.80 ± 0.03	28	0.95 0.00	9.5	3.75 ± 0.02	37.5	
	5 – 10	0.90 ±0.05	9	0.50 ± 0.05	5.0	1.40 ± 0.03	14.0	51.63
	10 – 15	0.01 ±0.04	0.1	0.003 ± 0.06	0.03	0.013 ± 0.01	0.13	
	0 – 5	1.85 ± 0.01	18.5	1.10 0.01	11.0	2.95 ± 0.05	30.0	
11	5 – 10	1.00 ± 0.03	10.0	0.55 ± 0.02	5.5	1.55 ± 0.01	15.5	45.55
	10 – 15	0.05 ± 0.02	0.5	0.005± 0.00	0.05	0.055 ± 0.04	0.55	

Table 2: Dissipation of ¹⁴C-deltamethrin in presence of chlorpyrifos in silt clay soil columns during 11 months



%	Retention Time (R _t)		Componds		
	Sys.D	Sys.C	Sys.B	Sys.A	
60-70	8.55	0.78	0.84	0.64	I
10	5.11	0.38	0.12	0.27	II
7	4.62	0.72	0.75	0.60	111
5	2.02	0.2	0.05	0.04	IV
2	4.05	0.10	0.03	0.01	V

Table 3: R, and R, values of deltamethrin and its degradation products

.These data indicate that movement of deltamethrin and dissipated is faster in the presence of chlorpyrifos under the field condition . It is worthy to mention that significant interactions between pesticides applied in combination, in terms of their persistence in soils and toxicity to crops and insects, have been documented [40,41]. The nature of extractable residues showed that deltamethrin represents 60-70% of the applied dose by TLC and HPLC analysis, in addition to, at least three degradation products (25%) at the end of experiment as shown in Figure 1 and Table 3. It is obvious that the preferred degradation pathway of deltamethrin occurs via hydrolysis of the ester linkage followed by oxidation. These metabolites appear to have no tendency to accumulate in soils [42].

References

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- Cousins IT, Beck AJ, Jones KC (1999) A review of the processes involved in the exchange of semi-volatile organic compounds (svoc) across the air –soil interface. Sci total environ 228: 5-24.
- Agrawal HC, Singh DK, Sharma VB (1994) Fate of parathion in soil under subtrobical field conditions. J Environ Sci Health B 29: 189-194.
- Ross DJ (1987) Soil microbial biomass estimated by the fumigation-incubation procedure: seasonal fluctuations and influence of soil moisture content. Soil Biol Biochem 19: 397-404.
- Coleman JM(1983) Biological strategies of nutrient cycling in soil systems, Adv. In Ecol Res 13: 1-55.
- Schnurer J, Clarholm M, Rosswall T (1985) Microbial biomass and activity in an agriculture soil with different organic matter contents. Soil Biol Biochem 17: 611-618.
- Duxbury JM (1989) Soil organic matter as a source and sink of plant nutrients. In: Coleman DC (eds) Dynamics of Soil Organic Matter in Tropical Ecosystems, Univ. Hawaii Press, Honolulu.
- Siddaramappa R, Sethunathan N (1975) Persistence of gamma-BHC and beta-BHC in Indian rice soils under flooded conditions. Pestic Sci 6: 395-403.
- Glass BL (1972) Relation between the degradation of DDT and iron redox system in soils. J Agric Food Chem 20: 324-327.
- Koprucu SS, Koprucu K, Ural MS (2006) Acute Toxicity of the synthetic pyrethroid deltamethrin to fingerling European catfish, Silurus glanis L Bull. Environ Contam Toxicol 76: 59 – 65.
- Roberts TR, Hutson DH (1999) Metabolic pathways of agrochemicals. Part 2: Insecticides and fungicides, Cambridge, UK: The Royal Society of Chemistry.
- 11. Katsuda Y (1999) Development of and future prospects for pyrethroid chemistry. Pestic Sci 55: 775–782.
- Arthur FH, Zehner JM (1994) Degradation and efficacy of deltamethrin plus chlorpyrifos methyl as protectants of wheat stored in Southeast Georgia (USA).
- Daglish GJ (1998) Efficacy of six grain protectants applied alone or on combination against three species of Coleoptera. Journal of Stored Product Research 34: 263-268.
- White NDG, Leesch, JG (1996) Chemical Control. In: ubramanyam, B. and Hagstrum, D., W. (eds) Integrated Management in Stored Products. Marcel Dekker Inc, New York, NY.
- Daglish GJ (1994) Efficacy of several mixtures of grain protectants on paddy and maize. In: Highley E, Wright EJ, Banks HJ and B. R. Champ (eds)

Proceedings of the 6th International Conference on Stored-Product Protection, Canberra, Australia, 650, University Press, Cambridge, U.K.

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- He F, Deng H, Ji X, Zhang Z, Sun J, et al. (1991) Changes of nerve excitability and urinary deltamethrin in sprayers. Int Arch Occup Environ Health 62: 587-590.
- Baeza-Squiban A, Bouaicha N, Santa-Maria A, Marano F (1990) Demonstration of the excretion by Dunaliella bioculata of esterases implicated in the metabolism of deltamethrin, a pyrethroid insecticide Bull. Environ Contam Toxicol 45: 39 - 45.
- Anand SS, Bruckner JV, Haines WT, Muralidhara S, Fisher JW, et al. (2006) Characterization of deltamethrin metabolism by rat plasma and liver microsomes. Toxicol Appl Pharmacol 212: 156 – 166.
- Hill BD, Inaba DJ, Schaalje GB (1992) Persistence of deltamethrin in baled alfalfa hay. J Agric Food Chem 40: 2493–2496.
- Raha P, Banerjee H, Das AK, Adityachaudhury N (1993) Persistence kinetics of endosulfan, fenvalerate, and decamethrin in and on eggplant (Solanum melongena L.). J Agric Food Chem 41: 923–926.
- 21. Westcott ND, Reichle RA (1993) Deltamethrin residues on Saskatoon berries. J Agric Food Chem 41: 2153–2155.
- 22. Larson SJ, Capel PD, Majewski MS (1997) Pesticides in Surface Waters: Distribution, Trend and Governing Factors. Ann Arbor Press Chelsea.
- Kidd H, James DR (1991) The agrochemicals Handbook A0791/Aug91. (3rd edition), Unwin Brothers Limited Old Working, Surney.
- Lee N, McAdam DP, Skerritt JH (1998) Development of immunoassays for type II synthetic pyrethroids, Hapten design and application to heterologous and homologous assays. J Agric Food Chem 46: 520-534.
- 25. Fakhr IMI, Hegazi B, Khatab MS, Zayed SMAD (1998) Egypt. J Chem 41: 145.
- Singla A, Dhingra S, Dhingra V (2012) Detection and identification of deltamethrin in autopsy material. J Punjab Acad Forensic Med Toxicol 12: 22-24.
- L' annunziate MF (1979) Radiotracers in agricultural chemistry IAEA Chapter 4, Page 96. Academic press, Vienna, Austria.
- Beulke S, Dubus IG, Brown CD, Gottesburen B (2000) Simulation of pesticide persistence in the field on the basis of laboratory data – a review. J Environ Qual 29: 1371–1379.
- Coats JR, Symonik DM, Bradbury SP, Dyer SD, Timson LK, et al. (1989) Toxicology of synthetic pyrethroids in aquatic organisms: An overview. Environ Toxicol Chem 8: 671-679.
- WHO (1990) Deltamethrin. Environmental health criteria 97. World Health Org. Geneva.
- 31. Hengel MJ, Mower CR, Shibamoto T (1997) New method for analysis of pyrethroid insecticides: Esfenvalerate, cis-permethrin, and trans-permethrin, in surface waters using solid-phase extraction and gas chromatography. Bull Environ Contam Toxicol 59: 171-178.
- Selim HM, Zhu H (2002) Retention and mobility of deltamethrin in soils: 2. Transport 1. Soil science 167: 580-589.
- Kaufman DD, Russel BA, Helling CS, Kayser AJ (1981) Movement of cypermethrin, decamethrin, permethrin and their degradation products in soil. J Agric Food Chem 29: 239 – 245.
- 34. Laabs V, Amelung W, Fent G, Zwch W, Kubiak R (2002) Fate of 14C-labeled soybean and corn pesticides in tropical soils of Brazil under laboratory conditions. J Agric Food Chem 50: 4619–4627.
- House WA, Long JA, Rate JE, Parker A, Orr DR (2000) Occurance and mobility of the insecticide permethrin in rivers in the southern Humber catchment, UK. Pest Manag Sci 56: 597-606.
- Hussain A, Maqbool U, Asi M (1994) Studies on dissipation and degradation of 14C-DDT and 14C-DDE in Pakistani soils under field conditions. J Environ Sci Health 29: 1-15.
- Zayed SMAD, Mostafa IYEI-Arab AE (1994) sChemical and biological release of 14C-residues from soil treated with 14 C – PP- -DDT. JEnviron Sci Health Part B 29: 169-175.
- Menon P, Gopal M, Prasad R (2004) Dissipation of chlorpyrifos in two soil environments of semi- Arid India. J Environ. Sci. Health 39: 517-531.

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- Khan SU, Ivarson KC (1981) Microbiological release of unextracted (bound) residues from an organic soil treated with prometryn. J Agri Chem 29: 1301-1303.
- Kaufman DD(1977) Biodegradation and persistence of several acetamide, acylanilide, azide, carbamate and organophosphate combinations. Soil Biol Biochem 9: 49-57.
- 41. Zayed SM, Farghaly M, Mahdy F, Soliman SM (2008) Impact of repeated pesticide applications on the binding and release of methyl 14C- monocrotophos and U-ring labelled 14C-carbaryl to soil matrices under field conditions. J Environ Sci Health B 43: 595-604.
- Zhang LZ, Khan SU, Akhtar MH, Ivarson KC (1984) Persistence, degradation, and distribution of deltamethrin in an organic soil under laboratory conditions. J Agric Food Chem 32: 1207-1211.