

## Enhance Vitamin B<sub>12</sub> Production by Online CO<sub>2</sub> Concentration Control Optimization in 120 m<sup>3</sup> Fermentation

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### Abstract

Great amounts of carbon dioxide generated by *Pseudomonas denitrificans* during high aerobic vitamin B<sub>12</sub> fermentation, while the influence of CO<sub>2</sub> concentration on vitamin B<sub>12</sub> production remains unclear. In this paper, we present parallel experiments to investigate various levels of inlet CO<sub>2</sub> fractions on the physiological metabolism of *P. denitrificans* in laboratory scale fermentation. The results demonstrated that the oxygen transfer rate, cell growth and glucose consumption were inhibited with CO<sub>2</sub> fraction elevated from 0.03% to 8.86 ± 0.24%, while the most exciting results showed that the specific vitamin B<sub>12</sub> production rate and the yield to glucose were greatly stimulated when dissolved CO<sub>2</sub> increased to 8.86 ± 0.24%. Therefore, the optimal exhausted CO<sub>2</sub> fraction control strategy in 120 m<sup>3</sup> fermenter was established. With the exhaust CO<sub>2</sub> concentration was well controlled at 7.5 ± 0.25% on-line, vitamin B<sub>12</sub> production greatly improved to 223.7 ± 3.7 mg/L, which was 11.2% higher than that of control. This strategy was proved to be significant necessary and effective for successfully scale up optimization in industrial vitamin B<sub>12</sub> fermentation.

**Keywords:** Dissolved carbon dioxide; *Pseudomonas denitrificans*; Vitamin B<sub>12</sub>; Fermentation; Scale up; Optimization

### Introduction

Vitamin B<sub>12</sub> is an important growth factor and has many applications in medicine and nutrition. Because of its increasing need in the world, vitamin B<sub>12</sub> has received much more attention recently. Most of the vitamin B<sub>12</sub> was produced by the culture of aerobic bacterium *P. denitrificans* in industrial production [1]. Due to the actively respiring characters of *P. denitrificans*, the poorly ventilated systems, high back pressure and hydrostatic pressure in large-scale fermenters, high dissolved CO<sub>2</sub> concentration were always encountered in industrial process. As for the significant influence of dissolved CO<sub>2</sub> on the growth and metabolism of many microorganisms [1], it must be taken into consideration in large-scale industrial fermentation.

The effects of CO<sub>2</sub> on growth and product formation in submerged cultures have been investigated in various microorganisms, and the controversy results were always encountered [2,3]. Researches of Gill and Lacoursiere had revealed that cell growth and metabolism of *Pseudomonas fluorescens* and *Escherichia coli* could be stimulated at low dissolved CO<sub>2</sub> level not more than 100 mm Hg pressure or 5% inlet gas phase respectively, while it could be greatly inhibited by the increased CO<sub>2</sub> concentration [4,5]. In yeasts cultivation, the growth rates were seriously decreased under CO<sub>2</sub>-enriched condition [6,7]. For filamentous fungi, in which the secondary metabolites formation was always accompanied with morphology changes [1], it was proved that both the morphology and product formation were negatively affected by the elevated CO<sub>2</sub> conditions [1,8,9]. However, many other reports demonstrated that CO<sub>2</sub> could also have positive effect on the metabolism of some microorganisms [10], in which existed high CO<sub>2</sub>-fix pathway activity, the anaplerotic reactions were very important for replenishing the metabolites generated from tricarboxylic acid cycle's pool. Researches had shown that enhanced CO<sub>2</sub> could stimulate the activity of key enzymes for the anaplerotic reaction in *Corynebacterium glutamicum* [11,12] and succinate-producing strain [13] by accelerating the accumulation of glutamate and succinate. The former studies on industrial vitamin B<sub>12</sub> fermentation have shown that *P. denitrificans* has high affinity to oxygen, and accompanied with higher CO<sub>2</sub> evolution rate during fermentation [14]. However, at present, there was

not clearly understood of the effect of CO<sub>2</sub> on metabolism of vitamin B<sub>12</sub> biosynthesis.

Therefore, the aim of present study was to evaluate the effect of CO<sub>2</sub> on production and metabolism of vitamin B<sub>12</sub> produced by *P. denitrificans*. Furthermore, an optimal industrial fermentation strategy based on online and CO<sub>2</sub> controlling strategy was successfully applied to enhance industrial vitamin B<sub>12</sub> production.

### Materials and Methods

#### Microorganisms and medium

The industrial strain *P. denitrificans* used in this study was donated by Huarong pharmacy corporation (Shijiazhuang, China). The cultivations were carried out at a temperature of 32°C and agitation speed of 260 rpm. The seed medium was composed of (g/L): sucrose 40, corn steep liquor 20, betaine 5, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> 1, (NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub> 2, MnSO<sub>4</sub>·H<sub>2</sub>O 0.8, COCl<sub>2</sub>·6H<sub>2</sub>O 0.02, MgO 0.3, 6-dimethyl-benzimidazole (DMBI) 0.01, ZnSO<sub>4</sub>·7H<sub>2</sub>O 0.01, CaCO<sub>3</sub> 1.5, the pH was adjusted to 7.2-7.4 by NaOH.

The fermentation medium *lab scale and industrial scale* was composed of (g/L): glucose 80, corn steep liquor 45, betaine 14, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> 1, KH<sub>2</sub>PO<sub>4</sub> 0.75, COCl<sub>2</sub>·6H<sub>2</sub>O 0.075, MgO 0.5, DMBI 0.05, ZnSO<sub>4</sub>·7H<sub>2</sub>O 0.08, CaCO<sub>3</sub> 1, pH 7.2-7.4. Feed media (FM) for the fed-batch fermentation were as follows (g/L), FM-1: Glucose 200, DMBI 0.15, COCl<sub>2</sub>·6H<sub>2</sub>O 0.15 and FM-2: betaine 30, COCl<sub>2</sub>·6H<sub>2</sub>O 0.3, DMBI 0.3.

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## Determination of dry cell weight (DCW) and residual sugar concentration

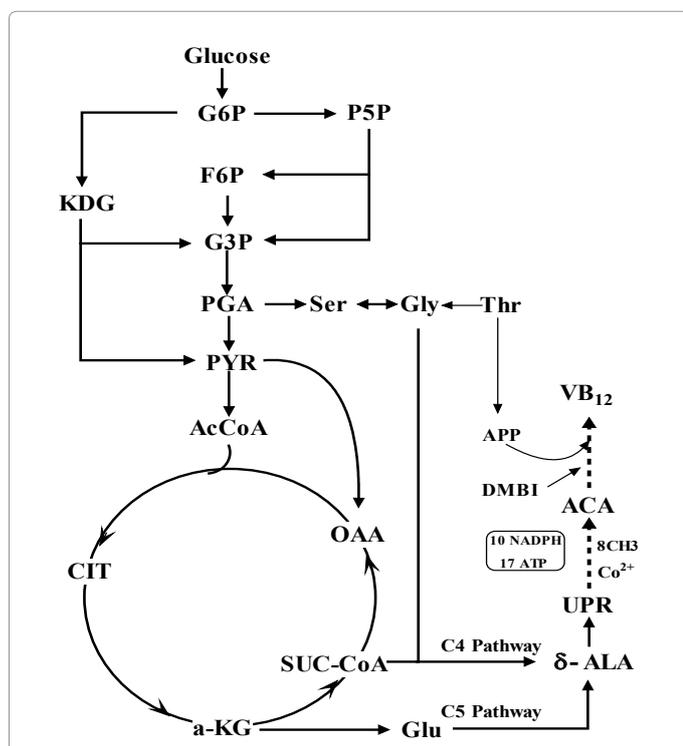
For determination of biomass concentration, 10 ml ferment broth was centrifuged at 4,000 g for 10 min. After washing twice with distilled water, the cell precipitate was dried to a constant weight at 105°C. The residual sugar concentration was measured by anthrone method [15].

## Determination of Pyruvate Carboxylase and intracellular δ-aminolevulinic acid (δ-ALA)

Vitamin B<sub>12</sub> is a highly complicated molecular structure [16,17] that is biosynthesized from eight molecules of δ-aminolevulinic acid (δ-ALA). This δ-ALA precursor can be generated by either of two pathways (Figure 1): In the C4 pathway, δ-ALA is made from glycine and succinyl CoA by the action of ALA synthase [1]. The δ-ALA was determined according to the literature [18]. For determination of Pyruvate Carboxylase, 10 ml of samples were taken quickly and centrifuged at 4°C, 12,000 g for 5 min, the obtained cells were frozen in -80°C nitrogen. After several times of repeated freezing and thawing, cell pellets were washed with 20 mM Tris-HCl (pH 7.6) for ultrasonication. The sonicated cell suspension was centrifuged and the cell extract was used for the enzyme assay. Pyruvate carboxylase activity was determined by the method of Uy [19].

## Measurement of organic acids and amino acids

For the analysis of the extracellular organic acids and amino acids



**Figure 1:** Proposed overview of the pathway of vitamin B<sub>12</sub> biosynthesis by *P. denitrificans* connects with central metabolism. Abbreviations: ACA, adenosyl cobyric acid; AcCoA, acetyl-CoA; APP, aminopropanol-o-2-phosphate; CIT, citrate; DMBI, dimethyl benzimidazole; F6P, fructose-6-phosphate; Gly, glycine; Glu, glutamate; G3P, glyceraldehyde-3-phosphate; G6P, glucose-6-phosphate; KDG, 2-Keto-3-deoxy-6-phosphogluconate; OAA, oxaloacetate; P5P, pentose-5-phosphate; PGA, 3-phosphate-glycerate; PYR, pyruvate; Ser, serine; SUC-CoA, succinyl-CoA; Thr, threonine; UPR, uroporphyrinogen III; α-KG, α-ketoglutarate; δ-ALA, δ-aminolevulinic acid.

Parameters	
Diameter (T, mm)	3,800
Bottom type	Dished
Filled volume (VL, L)	105,000
Capacity (V, L)	120,000
Baffle length (L, mm)	7600
Baffle width (w, mm)	15
Gas pipe inlet vent diameter (d, mm)	125
Impellers:	
6-Blade concave disc turbine impellers (D, mm)	1,150
3-Blade Rushton disc turbine impellers (D, mm)	1,219
3-Curved-blade disc turbine impellers (D, mm)	1,219
Standard pitched blade disc turbine impellers (D, mm)	1,219

**Table 1:** Geometric parameters of the 120 m<sup>3</sup> fermenter.

in metabolism, the HPLC systems (Agilent 1100, USA) were equipped with an Aqua Sep C18 (250 × 4.6 mm, 5 μm, ES, USA) and Zorbax Eclipse AAA column (150 × 4.6 mm, 5 μm, Agilent, USA) respectively, the determination conditions were the same to that of previously reported [14].

## Quantification of vitamin B<sub>12</sub> in the broth

The vitamin B<sub>12</sub> concentration was determined by HPLC. Broth samples (25 ml), into which 2.5 ml of 8% NaNO<sub>2</sub> and 2.5 ml of glacial acetic acid were added, were boiled for 30 min. Then the upper aqueous phase was measured using HPLC system equipped with C<sub>18</sub> column (4.6 mm, 25 cm, 5 μm) and UV detector (361 nm). 5% of acetonitrile was used as the mobile phase with flow rate of 1.0 ml min<sup>-1</sup> at 25°C combined with 95% of 0.25 M sodium acetate anhydrous, the pH was adjusted to 3.6 by acetic acid.

## Fermentation process design

Firstly, fed-batch fermentation was carried out in a 50 L turbine-agitated bioreactor (Shanghai Guoqiang Inc., China) with 30 L working volume. The cultivation conditions were same to that of previous report. When the cultivation turned into higher vitamin B<sub>12</sub> biosynthesis phase, 2.5 L broth was incubated into four parallel 5 L bioreactors respectively (Shanghai Guoqiang Inc., China) for further investigation. Fermentations were performed at 32°C, stirrer speed was 450 rpm, and the pressure in the tank was all kept at 0.1 MP. The inlet gas stream was consisted of a defined mixture of air and pure carbon dioxide with the aeration rate at 1 vvm. The mixture was prepared by thermal mass flow controllers (Mass-Trak, SIERRA, Netherlands). In addition to the control process where air was the input gas, processes were also performed with 3.32 ± 0.12%, 8.86 ± 0.24%, and 13.84 ± 0.27% CO<sub>2</sub> addition (v/v air) in the inlet gas mixture. Meanwhile, the dissolved CO<sub>2</sub> concentration in fermentation was determined with using dissolved carbon dioxide concentration electrode (Mettler).

## Fermentation in 120 m<sup>3</sup> fermenter

Three-stage fermentation was performed in 150 L, 9 m<sup>3</sup>, and 120 m<sup>3</sup> fermenters gradually, involving two stages of seed growth and one stage of fed-batch vitamin B<sub>12</sub> fermentation. The primary seed, secondary seed and large-scale fermentation were cultivated with the methods as previous reported [14]. The geometry and dimensions of 120 m<sup>3</sup> fermenter tanks and agitators were given in Table 1.

The inlet and exhaust gas ingredients were analyzed by a mass spectrometer (MAX300-LG, Extrel). Oxygen Uptake Rate (OUR) and Carbon dioxide Evolution Rate (CER) was calculated and collected

online Oxygen Uptake Rate (OUR), and were controlled by adjusting agitation and aeration conditions.

### Carbon dioxide control strategy model in 120 m<sup>3</sup> fermenter

The culture medium always enriched with CO<sub>2</sub> due to the high evolution rate of growing cells during fermentation process [20], especially in industrial scale fermenters with high hydrostatic pressures, high cell density or high pressure [21,22]. Previous researches revealed that *P. denitrificans* has high affinity to oxygen, with the dissolved oxygen concentration maintained nearly zero, therefore the oxygen transfer rate (OTR) was equal to OUR during the steady fermentation process. The OTR was determined by the agitation, aeration and the pressure, so the equation was used to create their relationship. Meanwhile, the active respiration make the CO<sub>2</sub> levels rise rapidly in large scale fermenters. In order to implement online control the oxygen supply levels and exhausted carbon dioxide concentration, we establish a response model about the OTR and exhausted CO<sub>2</sub> concentration related to aeration and agitation

$$OUR \approx OTR = m \times F^\alpha \times R^\beta \times P^\gamma \quad (\text{Eq.1})$$

Where m=correction coefficient;  $\alpha$ ,  $\beta$ ,  $\gamma$ =response coefficient of flow, agitation and pressure relate to oxygen transfer rate; OUR=oxygen uptake rate (mmol L/h); OTR=oxygen transfer rate (mmol L/h); F=air flow (L/min); P=pressure (MPa); R=agitator speed (rpm).

Moreover, the relation between oxygen uptake rate and carbon dioxide evolution rate has presented by RQ which shown in Equation 2. In addition, equation 3 shows online determination formula of carbon dioxide evolution rate.

$$RQ = \frac{CER}{OUR} \quad (\text{Eq.2})$$

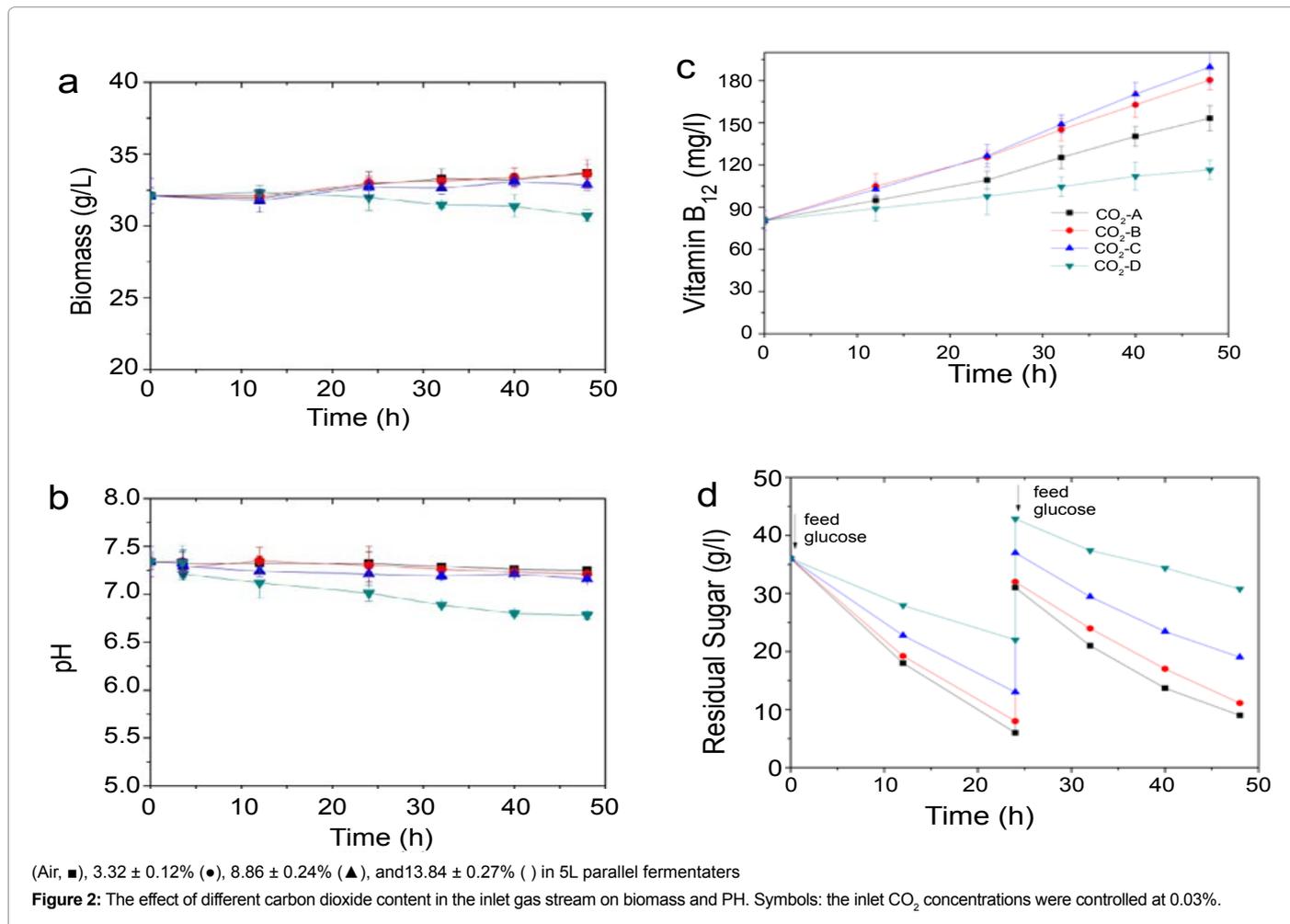
$$CER = \frac{F (E_{CO_2} - 0.03)}{V \cdot 22.4} \quad (\text{Eq.3})$$

## Results and Discussion

### Effect of CO<sub>2</sub> on the dynamic of vitamin B<sub>12</sub> fermentation

For investigating the effects of dissolved CO<sub>2</sub> levels on vitamin B<sub>12</sub> biosynthesis, the stationary cell growth broth in 50 L fermenter was transferred to the four 5 L reactors with 2.5 L working volume simultaneously. The fermentations were performed with operating conditions held constant to ensure that dissolved carbon dioxide was the only major variable. Samples were taken individually for analysis of biomass, glucose, inter-metabolites, and vitamin B<sub>12</sub> concentrations; and the data points represent mean values from duplicate experiments. A gas-mixing unit was used to give input CO<sub>2</sub> concentration in the influent gas of 0.03%, 3.32 ± 0.12%, 8.86 ± 0.24%, and 13.84 ± 0.27% respectively.

The time course of cell growth and pH under the four different CO<sub>2</sub>



addition were illustrated in Figure 2. During those four conditions, the dissolved oxygen concentration were all maintained at nearly zero levels, while the cell growth haven't any effect when the CO<sub>2</sub> addition elevated from 0.01% to 8.86 ± 0.24%; However, with influent CO<sub>2</sub> concentration increased to 13.84 ± 0.27%, the cell growth rate was greatly inhibited and biomass concentration decreased to 30.7 g/L after 48h cultivation, which was 8.9% lower than that of control (33.68 g/L) (Figure 2a). These might be caused by the respiration inhibition of CO<sub>2</sub> on biological membranes and cytoplasmic enzymes [23], and which may also interact with lipids of the cell membrane, the higher CO<sub>2</sub> concentration would greatly inhibit cell growth [1,9]. It could be clear seen from Figure 2b that pH had hardly any different to the control when CO<sub>2</sub> addition increased to 8.86 ± 0.24%, while the pH was remarkably decreased when the influent CO<sub>2</sub> elevated to 13.84 ± 0.27%.

The effect of influent CO<sub>2</sub> on vitamin B<sub>12</sub> production rate, glucose consumption rate and specific carbon dioxide evolution rate were summarized in Table 2. Results revealed the inhibitory effects of CO<sub>2</sub> concentration on the specific glucose consumption rate (SGCR). Little decrease of SGCR was appeared with CO<sub>2</sub> addition increased to 3.32 ± 0.12%; However, the SGCR were dramatically decreased to 0.14 and 0.10 mmol/gDCW/h when the influent CO<sub>2</sub> increased to 8.86 ± 0.24% and 13.84 ± 0.27% respectively, which were 19.2% and 46.7% lower than that of control (0.18 mmol/gDCW/h).

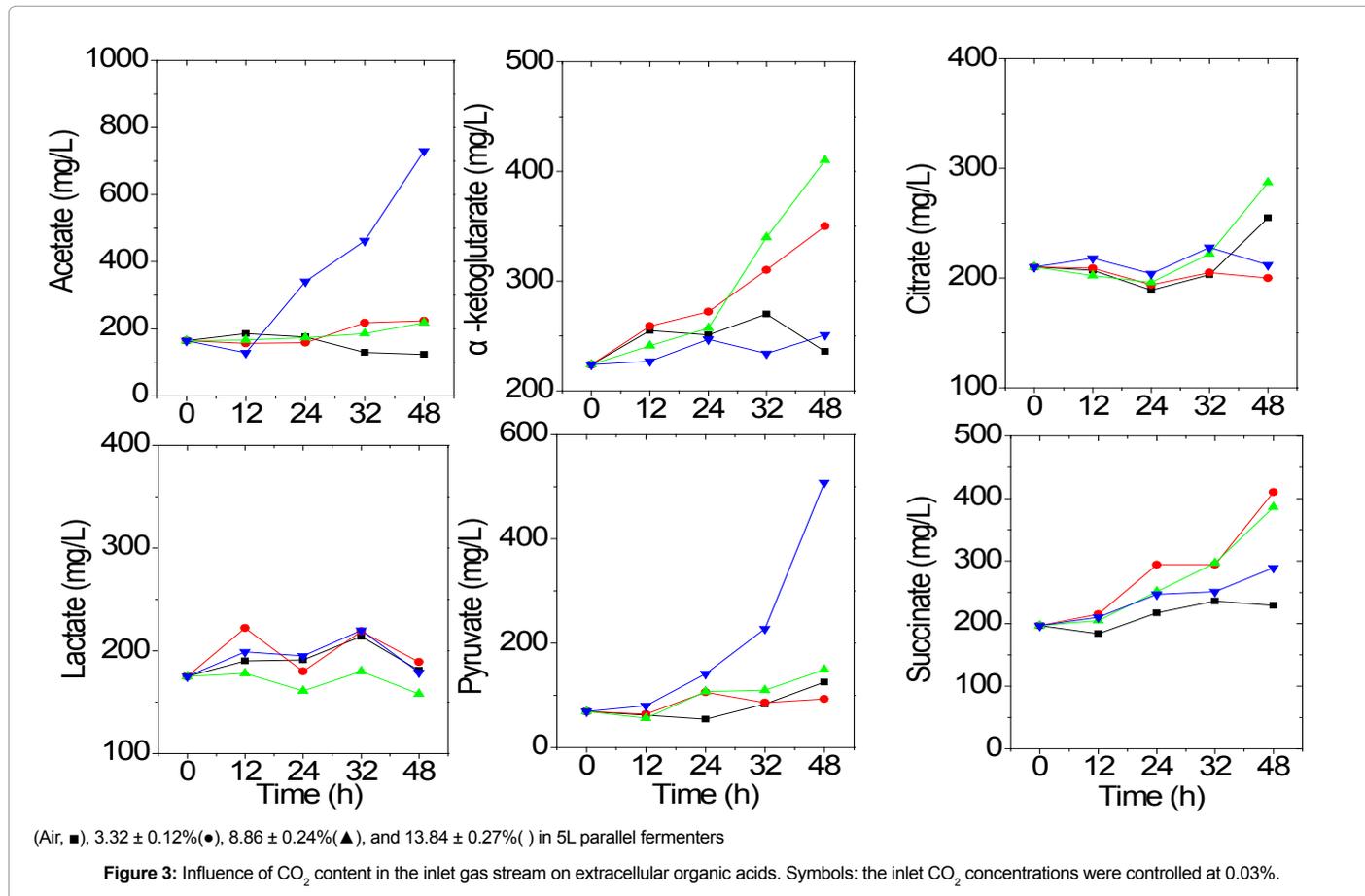
Just different from cell growth and SGCR, vitamin B<sub>12</sub> productions were greatly stimulated when influent CO<sub>2</sub> increased from 0.03% to 8.86 ± 0.24% (Figure 2c). The highest specific vitamin B<sub>12</sub> production

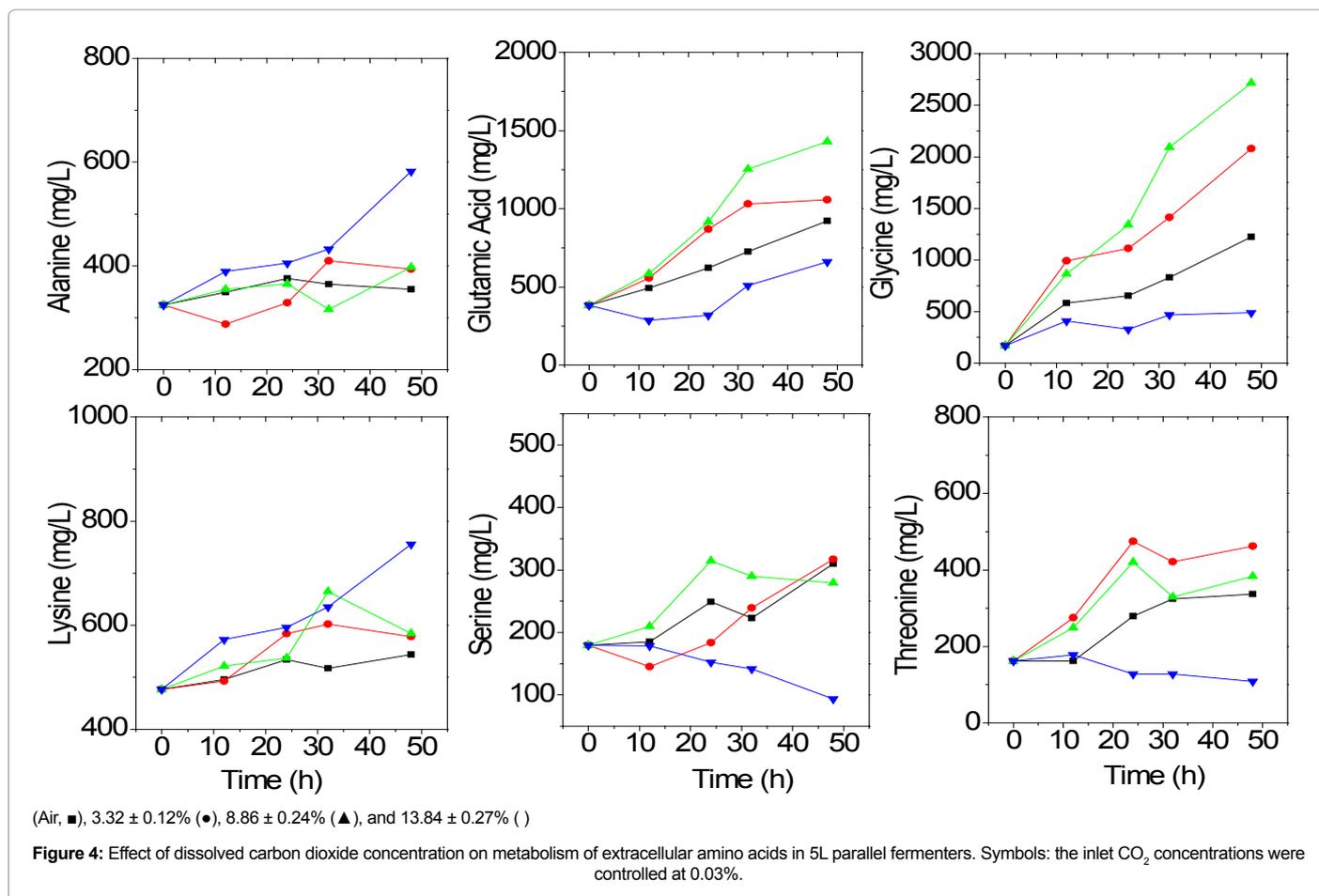
rate reached to 69.4 µg/gDCW/h under 8.86 ± 0.24% CO<sub>2</sub> addition, 53.5% higher than that of control (45.2 µg/gDCW/h). While vitamin B<sub>12</sub> productions and specific production rate were all significantly inhibited when the CO<sub>2</sub> concentration raised to 13.84 ± 0.27%. Analysis of vitamin B<sub>12</sub> yields to the glucose consumption ( $Y_{p/s}$ ) (Figure 2d) showed that the maximum yield of 480.5 µg/mmol (vitamin B<sub>12</sub>/ glucose consumption) were obtained when the influent CO<sub>2</sub> reached 8.86 ± 0.24%, while 13.84 ± 0.27% CO<sub>2</sub> addition led the yields dramatically decreased to lowest levels of 248.3 µg/mmol. Exhaust gas analysis showed that specific

CO <sub>2</sub> fraction	%	0.03	3.32 ± 0.12	8.86 ± 0.24	13.84 ± 0.27
CO <sub>2</sub> conc.	mmol/L	0.001 ± 0.0006	1.16 ± 0.07	3.05 ± 0.09	4.81 ± 0.12
$P_{ra}$	ug/L/h	1521.9	2164.2	2279.1	750.0
SPR <sup>b</sup>	ug/g(DCW)/h	45.2 ± 0.9	64.4 ± 2.1	69.4 ± 1.2	23.7 ± 1.6
SGCR <sup>c</sup>	mmol/g(DCW)/h	0.18	0.17	0.14	0.10
$Y_{p/s}$	ug/mmol	252.9	382.4	480.5	248.3
SCER <sup>d</sup>	mmol/g(DCW)/h	1.03	0.90	0.73	0.54
CO <sub>2</sub> /Glucose	C(mol)/C(mol) %	94.6 ± 2.1	89.2 ± 1.9	85.6 ± 1.5	93.3 ± 1.3

<sup>a</sup>,  $P_{ra}$ : Vitamin B<sub>12</sub> production rate  
<sup>b</sup>, SPR: specific vitamin B<sub>12</sub> production rate  
<sup>c</sup>, SGCR: specific glucose consumption rate  
<sup>d</sup>, SCER: specific carbon dioxide evolution rate

**Table 2:** Yield factors of vitamin B12 fermentation under various influent CO<sub>2</sub> concentration for 48h with the cultivations of stationary phase biomass transferred from a well cultured 50L fermentor.





carbon dioxide evolution rate (SCER) of *P. denitrificans* decreased with elevated CO<sub>2</sub> concentration in input gas. The statistics of the percent of glucose used for CO<sub>2</sub> generation were decreased from 94 ± 2.1% to 85 ± 1.5% with influent CO<sub>2</sub> increased from 0.03% to 8.86 ± 0.24%; while under the highest CO<sub>2</sub> concentration of 13.84 ± 0.27%, nearly 93.3 ± 1.3% of the glucose consumption was seriously transformed to CO<sub>2</sub>. In conclusion, maintained an appropriate input CO<sub>2</sub> concentration around 8.86 ± 0.24% was positive for the economical vitamin B<sub>12</sub> production.

Previous researches on the effects of elevated CO<sub>2</sub> on the industrial producing strains of bacteria, yeasts, and fungi [1,9] revealed that the inhibitory action of product biosynthesis was probably caused by the decreased substrate consumption with elevated CO<sub>2</sub> concentration. In our research, however, glucose consumption decreased with the increased influent CO<sub>2</sub> concentration, while the yields of vitamin B<sub>12</sub> based on glucose consumption increased greatly with the elevated influent CO<sub>2</sub> concentration. These results indicated that there must exist abnormal metabolism characters of *P. denitrificans* under the various influent CO<sub>2</sub> concentrations. The reasons were further investigated by organic acid and amino acid analysis of the central metabolism.

#### Effect of CO<sub>2</sub> on the inter-metabolites in fermentation

The researches on vitamin B<sub>12</sub> biosynthesis had demonstrated that the concentrations of the precursor of δ-aminolevulinic acid (δ-ALA) played an important role on vitamin B<sub>12</sub> biosynthesis [16]. δ-ALA could be synthesized from C4-pathway (from glycine and succinyl

CoA by δ-ALA synthesis) or C5-pathway (from glutamate by glutamyl-tRNA reductase) associated with central metabolism [24]. The dynamic processes of extracellular organic acids concentration under different CO<sub>2</sub> levels were shown in Figures 3 and 4.

When influent CO<sub>2</sub> increased from 0.03% to 8.86 ± 0.24%, α-ketoglutarate and succinate concentrations greatly increased, which were actually accelerated the biosynthesis of glutamate, and the other organic acids such as pyruvate, acetate, lactate and citrate hadn't any different. However, the acetate and pyruvate were accumulated much higher when influent CO<sub>2</sub> reached 13.84 ± 0.27%. These were probably caused by the inhibition of respiration under high CO<sub>2</sub> concentration, and the resulted high NADH accumulation in the cell really repressed the activity of pyruvate dehydrogenase and TCA cycle, and then attributed to the accumulation of acetate and pyruvate.

The dynamic analysis on the concentrations of extracellular amino acids (Figure 3) showed that glutamate and glycine synthesis rate was significantly enhanced with elevated CO<sub>2</sub> concentration. Under 8.86 ± 0.24% CO<sub>2</sub> addition, glutamate and glycine concentration reached 1430.2 mgL<sup>-1</sup> and 2716.4 mg/L, respectively, which were 55% and 122% higher than that of control. Threonine reached highest synthesis rate when inlet CO<sub>2</sub> increased to 3.32 ± 0.12%. These metabolites accumulation could be accelerate the precursor's biosynthesis for vitamin B<sub>12</sub> synthesis.

δ-ALA is the critical precursors of vitamin B<sub>12</sub> biosynthesis, the time course of δ-ALA concentration under different levels of inlet CO<sub>2</sub>

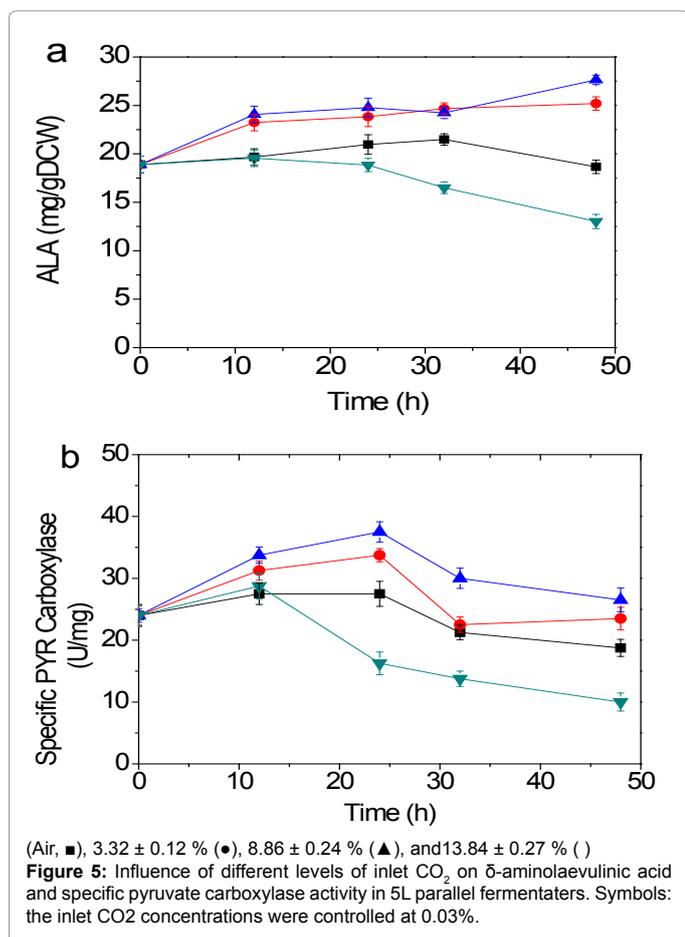
were shown in Figure 5a, experiment results demonstrated the positive effects of moderate CO<sub>2</sub> concentration exerted to the δ-ALA synthesis. In bioprocess sparged with 8.86 ± 0.24% CO<sub>2</sub> addition, the δ-ALA concentration reached the top point at 26.4 ± 1.2 mg/gDCW, which was 40.4% higher compared to that of control (18.6 ± 0.9 mg/gDCW). However, the δ-ALA biosynthesis was seriously inhibited with 13.84 ± 0.27% CO<sub>2</sub> addition, the final δ-ALA concentration was lowest as 13.02 ± 0.5 mg/gDCW, 30% lower than that of control.

Experiment results showed that the biosynthesis of glycine, succinic acid, and glutamate were elevated with the increased influent CO<sub>2</sub> concentration from 0.03% to 8.86 ± 0.24%, which actually augmented δ-ALA generation for higher vitamin B<sub>12</sub> biosynthesis.

### Effect of CO<sub>2</sub> on pyruvate carboxylase activity

Previous researches have illustrated that anaplerotic reaction catalysed by pyruvate carboxylase was mostly existed for replenishing the TCA cycle in *Pseudomonas* spp [25]. In this work, the dynamic analyses of specific pyruvate carboxylase activity in different input CO<sub>2</sub> addition were shown in Figure 5b. In the process sparged with 8.86 ± 0.24% CO<sub>2</sub>, the maximal specific activity reached 37.2 U/gDCW, which was 35% higher than that of control (27.5 U/gDCW).

Tricarboxylic acid (TCA) cycle is one of the main central pathways in aerobic bacteria. It is responsible for the complete oxidation of acetyl-CoA and provision of precursors for many TCA-related metabolites biosynthesis. As precursors of the metabolites succinyl-CoA and glutamate for δ-ALA biosynthesis were all generated from



F(L/min)	R(rpm)	P(MPa)	OUR (mmol/L/h)	
			Determined	Predicteda
1263.1	78	0.031	47.38	45.71
929.9	78	0.034	40.66	41.77
406.4	50	0.031	24.73	24.20
600.5	90	0.041	39.10	39.72
850.4	78	0.032	37.51	39.80
1523.2	110	0.040	63.2	61.68
1224.5	81	0.032	46.07	46.41
746.3	79	0.029	36.65	37.23
960.0	55	0.035	38.42	36.21
959.4	78	0.033	40.77	41.95
709.2	77	0.028	36.74	35.77
625.5	81	0.033	42.58	36.39
1121.8	78	0.027	43.82	42.32

a calculated from the obtained equation 2 with the corresponding process parameters  
**Table 3:** The determined and predicted OUR values at various aeration, agitation and pressure levels in the 120 m<sup>3</sup> bioreactor.

TCA cycle. In the rapid VB<sub>12</sub> synthesis phase, the relative high demand of succinyl-CoA and glutamate lead to an “overflow” of corresponding compounds of TCA cycle. Therefore, TCA cycle has to be replenished continuously through anaplerotic reaction for oxalacetate generation to maintain the cycle running.

The experiment results demonstrated that elevated levels of CO<sub>2</sub> concentration had significant effect on vitamin B<sub>12</sub> fermentation. When the influent CO<sub>2</sub> concentration increased from 0.03% to 8.86 ± 0.24%, the specific vitamin B<sub>12</sub> production rate and the yields of vitamin B<sub>12</sub> to glucose were stimulated. However, cell growth, glucose utilization, specific vitamin B<sub>12</sub> production rate, and the yields of vitamin B<sub>12</sub> to glucose were all greatly inhibited when CO<sub>2</sub> addition up to 13.84 ± 0.27%. Therefore, the influences of CO<sub>2</sub> on industrial vitamin B<sub>12</sub> production have to take into consideration.

### Industrial fermentation strategy based on online exhaust CO<sub>2</sub> controlling

**Exhaust CO<sub>2</sub> concentration controlling strategy:** With the process parameters determined under various agitation speed, air flow, and pressure obtained from 120 m<sup>3</sup> fermentation tank are shows in Table 3, we obtained relevant respond coefficient through regression analysis of these parameters between OTR and Process parameters determined (Eq.1). Then, oxygen transfer rate and oxygen uptake rate could be calculated according to these parameters (Eq.4).

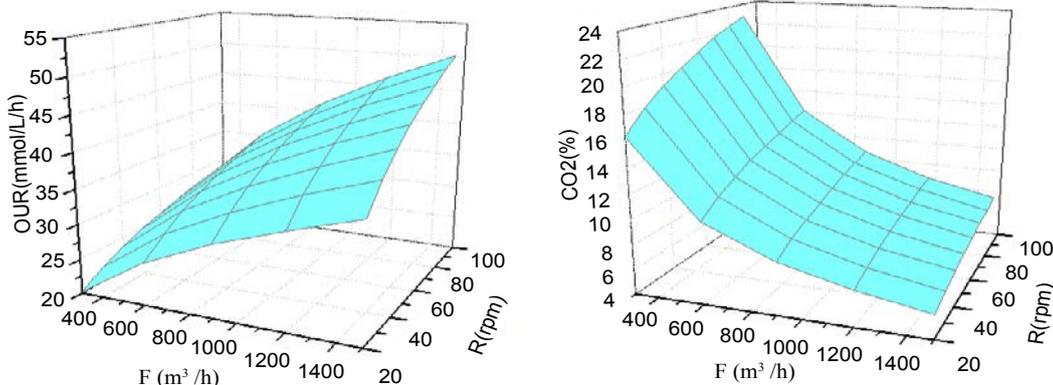
$$OUR \approx OTR = 1.0074 \times F^{0.3729} \times R^{0.4635} \times P^{0.2472} \quad (\text{Eq.4})$$

The carbon dioxide fraction in exhausted gas was calculated from a mutual relation between RQ, CER, and OUR (Eq. 2,3,5)

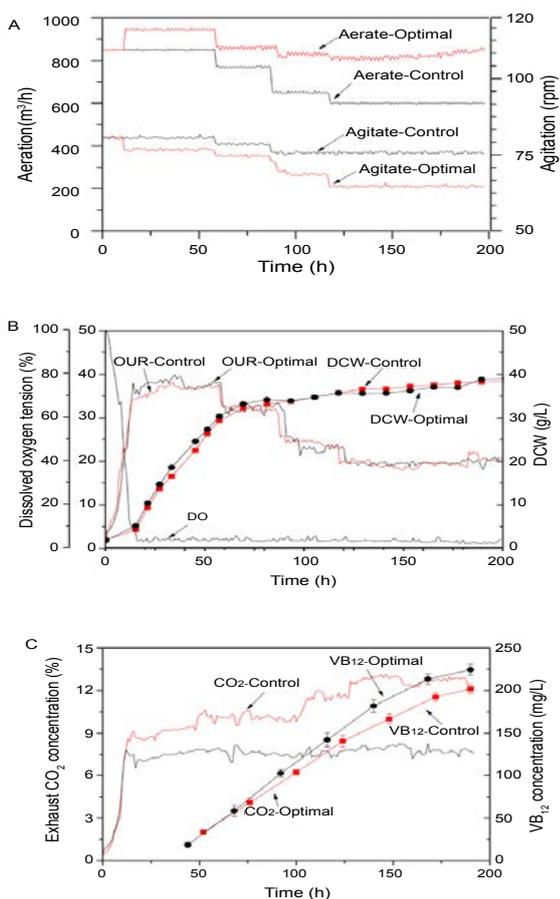
$$E_{CO_2} = RQ \times F^{-0.36271} \times R^{0.4635} \times P^{0.2472} \times V \times 22.57 + 0.03 \quad (\text{Eq.5})$$

During the vitamin B<sub>12</sub> fermentation process, glucose was used as the main carbon substrate, the respiration quotient (RQ) of this strain was always maintained at 1.03 ± 0.02 in microorganism growth period and RQ was calculated as 0.95 ± 0.04 in vitamin B<sub>12</sub> synthesize period. Then OUR and exhausted CO<sub>2</sub> concentration from 120 m<sup>3</sup> fermentation tank could be obtained by adjusting agitator speed and air flow according to the created model above (Figure 6).

**Scale-up fermentation based on exhaust CO<sub>2</sub> concentration control:** Based on the results obtained from 5 L vitamin B<sub>12</sub> fermentation, the optimal controlling strategies based on exhausted



**Figure 6:** The response of oxygen uptake rates and exhaust CO<sub>2</sub> concentrations as a function of aeration and agitation in 120 m<sup>3</sup> large scale fermenter (based on the obtained equation 4 and equation 5).



**Figure 7:** Time course of industrial VB<sub>12</sub> fermentation with the exhaust CO<sub>2</sub> concentration controlled at throughout at 7.5 ± 0.25 %, (optimal) and the control without exhaust CO<sub>2</sub> controlled (control) in 120 m<sup>3</sup> fermenters and the effect of CO<sub>2</sub> concentration on DCW (b) and VB<sub>12</sub> concentration (c) by correlation adjustment of oxygen supply conditions (a).

CO<sub>2</sub> concentration were implemented for vitamin B<sub>12</sub> fermentation in 120 m<sup>3</sup> fermenters. The exhausted CO<sub>2</sub> concentration was controlled at 7.5 ± 0.25% by adjusting aeration and agitator speed, also OUR

was maintained at needed level. This strategy greatly prevented the inhibition of high dissolved CO<sub>2</sub> concentration on vitamin B<sub>12</sub> synthesis.

Time course of OUR, dry cell weight, vitamin B<sub>12</sub> production, and exhausted CO<sub>2</sub> concentration were shown in Figure 7. By control the aeration and agitation throughout the fermentation process, the changes of OUR were maintained at similar levels in fermentation processes (Figure 7b), the optimal strategy by stepwise reduction of oxygen uptake rate under dissolved oxygen limiting level during fermentation process was used for oxygen supply levels control [11]. Meanwhile, the exhausted CO<sub>2</sub> concentration changed greatly and maintained at the suitable levels with the adjustment of agitation and aeration conditions. In the contrast conditions, the exhaust CO<sub>2</sub> concentration increased greatly to 9.1 ± 0.15%, cell growth rate was lower than that of optimal strategy, especially when it elevated to 12.0 ± 0.24% after 75 h cultivation, the vitamin B<sub>12</sub> production rate was gravely inhibited (Figure 7c).

The results presented demonstrated that cell growth and vitamin B<sub>12</sub> production were greatly stimulated by controlling a lower CO<sub>2</sub> concentration of 6.0 ± 0.24% during the early fermentation phase. The highest vitamin B<sub>12</sub> production rate was achieved when exhausted CO<sub>2</sub> concentration were controlled at 7.5 ± 0.25%. The maximum vitamin B<sub>12</sub> production reached 223.7 ± 3.7 mg/L at 190 h, which was 11.2% higher than that of without CO<sub>2</sub> concentration controlled (201.4 ± 2.9 mg/L). It can be observed that higher production would be realized through simultaneously regulate the appropriate OUR and CO<sub>2</sub> concentration levels are more beneficial to vitamin B<sub>12</sub> synthesis.

## Conclusion

CO<sub>2</sub> was produced in aerobic and anaerobic fermentations through decarboxylation reactions in a number of metabolic pathways, and hence has an influence on the performance of microbial cultivations [1,9]. Especially in industrial large bioreactors with high total pressure, poorly ventilated system, and inefficiently aerated fermenters. In this work, we investigated the effects of dissolved CO<sub>2</sub> concentration on vitamin B<sub>12</sub> fermentation of *P. denitrificans*, established the correlation of exhausted CO<sub>2</sub> concentration and the activity of the microbial metabolism by inter-scale observation and data association. The novel and optimal fermentation controlling strategy based on OUR and exhausted CO<sub>2</sub> concentration was successfully applied in industrial scale up of vitamin B<sub>12</sub> fermentation.

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