

Editorial

Membrane Distillation: Principles, Applications and Perspectives

Criscuoli A*

Institute on Membrane Technology (ITM-CNR), via P. Bucci 17/C, Rende (CS) 87030, Italy

*Corresponding author: Criscuoli A, Institute on Membrane Technology (ITM-CNR), via P. Bucci 17/C, Rende (CS) 87030, Italy, Tel: +39-0984-49211; E-mail: a.criscuoli@itm.cnr.it

Received date: August 29, 2017; Accepted date: August 31, 2017; Published date: September 03, 2017

Copyright: © 2017 Criscuoli A. This is an open-access article distributed under the terms of the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

Abstract

Membrane Distillation (MD) is a membrane-based operation able to give 100% theoretical ions rejection and to efficiently work with high concentrated brines. Both features make MD of interest for the purification of wastewater, the production of ultra-pure water and the concentration of brines produced in desalination. MD, also integrated with other membrane operations, can be a valuable way to improve the performance of separation processes.

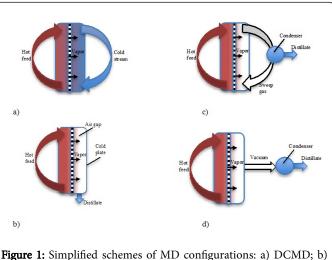
Keywords Membrane distillation; Wastewater treatment

Introduction

MD is a thermally-driven membrane operation where the vapor molecules evaporate from the feed thanks to a difference of vapor pressure created across the membrane. Through the evaporation it is possible to produce a pure distillate and to concentrate the feed. The membranes employed are hydrophobic and porous. Polymeric membranes made of polypropylene (PP), polyvinylidene fluoride (PVDF), polytetrafluoroethylene (PTFE) are the most used, although some studies on zeolite and ceramic membranes have been also carried out. If compared to conventional distillation units, MD is able to work with smaller footprint and lower operating temperatures. Typical operating temperatures are around 50°C-80°C and waste heat or renewable sources, like solar energy, can be used to heat the feed. If compared to pressure-driven membrane operations, MD operates at lower pressures (usually at atmospheric pressure) and is able to treat high concentrated streams, due to the lower effect of concentration polarization on its performance. Its potentialities have been proven for many types of separations, as well documented by the huge amount of papers published in the field [1-24].

Main Principles and Applications of Membrane Desalination

There are four main configurations of MD, that differ in the way the difference of vapor pressure is created and the distillate recovered (Figure 1). Direct Contact Membrane Distillation-DCMD (Figure 1a) is the simplest one, using a cold stream to promote the difference of temperature across the membrane; in this case vapor condenses directly into the cold stream. Air Gap Membrane Distillation-AGMD (Figure 1b) presents an air gap in which the vapor diffuses before condensing. Sweep Gas Membrane Distillation-SGMD (Figure 1c) uses a sweep gas to remove the permeating vapor, while Vacuum Membrane Distillation-VMD (Figure 1d) applies vacuum. In both cases, the vapor is liquified in an external condenser.



AGMD; c) SGMD; d) VMD.

The efficiency of the MD is strongly dependent on the membrane properties. To make the process happening, it is essential to work with hydrophobic membranes. Pore size should be of the order of few microns and has to be chosen with the objective of working with reasonable fluxes and high Liquid Entry Pressure (LEP), to reduce the wetting risk. A typical value of pore size is around 0.2 μ m. High porosity values should be preferred, for both enhancing the flux and reducing the heat loss by conduction through the membrane matrix, particularly for DCMD. Thickness plays also an important role, influencing the flux, the mechanical resistance and the heat loss by conduction.

Being a thermally-based process, the efficiency of MD is often evaluated in terms of Gained Output Ratio (GOR), that gives information on how much of the supplied heat has been used for the evaporation (useful heat).

MD finds application in different sectors, as summarized in Table 1.

Desalination Treatment of gas produced waters Treatment of waters contaminated by toxic compounds, like boron, arsenic, uranium, fluoride, etc. Treatment of brines and high-concentrated waters Treatment of radioactive waters Ultra-pure water production	
Treatment of gas produced waters Treatment of waters contaminated by toxic compounds, like boron, arsenic, uranium, fluoride, etc. Treatment of brines and high-concentrated waters Treatment of radioactive waters Ultra-pure water production	Concentration and purification of wastewaters (dairy effluents, olive mill wastewaters, textile effluents, etc.)
Treatment of waters contaminated by toxic compounds, like boron, arsenic, uranium, fluoride, etc. Treatment of brines and high-concentrated waters Treatment of radioactive waters Ultra-pure water production	Desalination
Treatment of brines and high-concentrated waters Treatment of radioactive waters Ultra-pure water production	Treatment of gas produced waters
Treatment of radioactive waters Ultra-pure water production	Treatment of waters contaminated by toxic compounds, like boron, arsenic, uranium, fluoride, etc.
Ultra-pure water production	Treatment of brines and high-concentrated waters
	Treatment of radioactive waters
Fruit juice concentration	Ultra-pure water production
	Fruit juice concentration
Dehydration of solid particles	Dehydration of solid particles

 Table 1: Some fields of application of MD.

Conclusion and Future Perspectives

MD has a good potential for the purification and concentration of streams and the growing number of researches on MD clearly documents the interest in the field. It is expected that improvements in the membrane properties and the development of membrane modules specifically designed will drive the application of MD at large scale.

References

- El-Bourawi MS, Ding Z, Ma R, Khayet M (2006) A framework for better understanding membrane distillation separation process. J Membr Sci 285: 4-29.
- 2. Khayet M (2011) Membranes and theoretical modeling of membrane distillation: A review. Adv Colloid Interface Sci 164: 56-88.
- 3. Alkhudhiri A, Darwish N, Hilal N (2012) Membrane distillation: A comprehensive review. Desalination 287: 2-18.
- Camacho LM,Dumée L, Zhang J (2013) Advances in membrane distillation for water desalination and purification applications. Water 5: 94-196.
- Wang P, Chung TS (2015) Recent advances in membrane distillation processes: Membrane development, configuration design and application exploring. J Membr Sci 474: 39-56.
- 6. Ashoor BB, Mansour S, Giwa A, Dufour V, Hasan SW (2016) Principles and applications of direct contact membrane distillation (DCMD): A comprehensive review. Desalination 398: 222-246.
- Belafi-Bako K, Koroknai B (2006) Enhanced water flux in fruit juice concentration: coupled operation of osmotic evaporation and membrane distillation. J Membr Sci 269: 187-193.
- Song L, Ma Z, Liao X (2008) Pilot plant studies of novel membranes and devices for direct contact membrane distillation-based desalination. J Membr Sci 323: 257-270.
- 9. Mericq J, Stephanie L, Cabassud C (2010) Vacuum membrane distillation of seawater reverse osmosis brines. Water Res 44: 5260–5273.
- Simone S, Figoli A , Criscuoli A, Carnevale MC, Rosselli A, Drioli E (2010) Preparation of hollow fibre membranes from PVDF/PVP blends and their application in VMD. J Membr Sci 364: 219-232.
- 11. Pal P, Manna AK (2010) Removal of arsenic from contaminated groundwater by solar-driven membrane distillation using three different commercial membranes. Water Res 44: 5750-5760.

- 12. Yarlagadda S, Gude VG, Camacho LM, Pinappu S, Deng S (2011) Potable water recovery from As, U, and F contaminated ground waters by direct contact membrane distillation process. J Hazard Mater 192: 1388-1394.
- Shaffer DL, Chavez LHA, Ben-Sasson M, Romero-Vargas Castrillón S, Yip NY, et al. (2013) Desalination and reuse of high-salinity shale gas produced water: Drivers, technologies and future directions. Environ Sci Technol 47: 9569-9583.
- 14. Singh D, Prakash ZP, Sirkar KK (2013) Deoiled produced water treatment using direct-contact membrane distillation. Ind Chem Eng Res 52: 13439-13448.
- 15. Criscuoli A, Bafaro P, Drioli E (2013) Vacuum membrane distillation for purifying waters containing arsenic. Desalination Special Issue: Membrane Distillation. Desalination 323: 17-21.
- 16. Liu H, Wang J (2013) Treatment of radioactive wastewater using direct contact membrane distillation. J Hazard Mater 261: 307-315.
- Khan EU, Martin AR (2014) Water purification of arsenic contaminated drinking water via air gap membrane distillation. Period Polytech Mech Eng 58: 47-53.
- Boubakri A, Bouguecha SAT, Dhaouadi I, Hafiane A (2015) Effect of operating parameters on boron removal from seawater using membrane distillation process. Desalination 373: 86-93.
- Duong HC, Chivas AR, Nelemans B, Duke M, Gray S, et al. (2015) Treatment of RO brine from CSG produced water by spiral-wound air gap membrane distillation- a pilot study. Desalination 36: 121-129.
- Kezia K, Lee J, Weeks M, Kentish S (2015) Direct contact membrane distillation for the concentration of saline dairy effluent. Water Res 81: 167-177.
- 21. Bouchrit R, Boubakri A, Hafiane A, Bouguecha AITS (2015) Direct contact membrane distillation: Capability to treat hyper-saline solution. Desalination 376: 117-129.
- 22. Criscuoli A, Carnevale MC, Drioli E (2016) Study of the performance of a membrane-based vacuum drying process. Sep Purif Technol 158: 259-265.
- 23. Dao TD, Laborie S, Cabassud C (2016) Direct As (III) removal from brackish groundwater by vacuum membrane distillation: Effect of organic matter and salts on membrane fouling. Sep Purif Technol 157: 35-44.
- 24. Criscuoli A (2016) Improvement of the Membrane Distillation performance through the integration of different configurations. Chem Eng Res Des 111: 316-322.