Neuroprotective Effects of *Pterois volitans* Venom against Alcohol Induced Oxidative Dysfunction in Rats

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**Abstract**

The *P. volitans* generally is an edible fish, it protein are considered to be a source of food for human and the venom use for the development of new drugs. Pharmacological character of its venom were characterised during the study period. The lionfish venom shows the neuroprotective efficacy in alcohol intoxicated albino rat brain. These findings are further confirmed by histopathological observations. Therefore, the lionfish venom could be used as a neuroprotective agent.

**Keywords:** *P. volitans*; Venomous sting; Neuroprotective efficacy; Histopathology

**Introduction**

In the industrial world people are having some kind of problem which may be largely man-made and iatrogenic in origin. Still, we ignore and dismiss their growing incidence to factors which we assume are not preventable. Alcohol abuse is one of the major problems in Indian continent and is responsible for a significant percentage of hospital admission. Ethanol molecule is small and soluble in both water and lipids. It permeates all tissues of the body and affects most vital functions of virtually all organs including brain, liver, kidney, heart and pancreas [1]. Metabolism of alcohol leads to the generation of free radicals and the chain reaction of lipid peroxidation that causes damage to the brain and other vital organs. Therefore inhibition of free radical generation is important in providing protection against hepatic damage. Toxicity due to the alcohol consumption in the body leads to the changes in the main organs, especially the brain. The morbidity and mortality rate of heavy alcohol drinkers have since been reported to have reached 6.1% [2]. Alcoholic liver disease (ALD) is one of the most serious results of chronic alcohol abuse in the world [3,4] have reported that the factors that mediate the occurrence of ALD are acetaldehyde, oxidative stress, hypoxia, immune response and membrane alterations. Antioxidants play a significant role in protecting living organisms from the toxic effect of various chemicals by preventing free radical formation [5]. Now a days, alcohol produces various dysfunctions in the human being. A number of fish has venomous sting, including the family Scorpaenidae. Venomous species are only a few [6], as people consider it for edible. Due to venom extracts storing have some technical difficulties [7], the source of marine organisms especially fish, remain a largely unused source of novel compounds. The piscine venoms are effective on the responses to the venoms of all species of fish. Numerous fish venoms are effective on the toxicity of *P. volitans* administered to rats and the incidence of oxidative stress and neurotoxicity are made.

**Materials and Methods**

Specimens of *P. volitans* (Figure 1) were obtained from the local aquarium, killed (by cooling), and the venomous spines were removed and stored in 10% glycerol solution at -80°C. The spines were thawed and ground in a chilled mortar and pestle in 10% glycerol solution. The spines were then centrifuged (7000 g, 10 mins) the supernatant removed, the pellet resuspended again in approximately 1 ml of 10% glycerol and recentrifuged. The final supernatants were pooled, assayed for protein concentration using a Bio-Rad Dc protein assay kit, and adjusted to a concentration of 1mg/ml protein, before being aliquoted and stored at -20°C until use and the venom was prepared as described by [11]. The protein was estimated by [18]. The concentration was adjusted to 1 mg/ml; aliquoted and stored at -20°C until use.

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Stabilization of the venom

The venom preparation was maintained at various temperatures for 1 hour (-80°C, -20°C, 0°C, 4°C, 37°C and 100°C) immediately brought back to room temperature and assayed for activity using rat bio assay. The venom preparation was kept at optimum temperature in different buffers namely 50 mM acetate buffer with pH ranging from 3 to 6.2 and 50 mM tris-HCl buffer with pH ranging from 7-10. The optimum pH was determined by performing the rat bioassay. The venom was stored at -20°C in the presence of sodium aside for six months. Aliquots were withdrawn at every month and their activity was determined by rat lethality assay.

The experiments were carried out in adult male Wistar rats (180-200 g) procured from Central Animal House (CAH), Faculty of Medicine, Annamalai University. The animals were housed under the conditions of constant temperature (24 ± 2°C) and humidity (50-60%), under a 12 h light/12 h dark photoperiod in the CAH of the University. Water and standard laboratory food (Hindustan Lever Ltd., Mumbai, India) were provided ad libitum. The experimental protocol used in this study was approved (Reg. No.450/2007/CPCSEA) by the Institutional Animal Ethical Committee (IAEC) for the purpose of control and supervision on experimental animals of Rajah Muthiah Medical College, Annamalai University, Annamalainagar, India. After acclimatization, the animals were divided into 6 groups, consisting of 6 rats in each group (Figure 2).

Experimental protocol

Group 1: Control rats received normal saline 100 µl orally for four weeks.

Group 2: Rats orally received with 20% alcohol in saline for four weeks.

Group 3: Rats orally received the 20% alcohol with 1% of LD₅₀ dose (42.5 µg/kg) of venom in 100 µl of sterile saline (2.125 lg/kg BW by ip—low dose: LD₅₀) for four weeks.

Group 4: Rats orally received the 20% alcohol with 1% of LD₅₀ dose (42.5 µg/kg) of venom in 100 µl of sterile saline (4.25 lg/kg BW by ip—medium dose: MD) for four weeks.

Group 5: Rats orally received the 20% alcohol with 1% of LD₅₀ dose (42.5 µg/kg) of venom in 100 µl of sterile saline (24.25 lg/kg BW by ip—high dose: HD) for four weeks.

Group 6: Rats received the (2.125 lg/kg BW by ip—medium dose: MD) for four weeks.

The animals were maintained in their respective groups for four weeks. Food and fluid intake and body weights were measured weekly. At the end of the experimental period, the animals were anesthetized using ether and sacrificed by cervical decapitation. Brain tissues were excised, washed and homogenized in 0.1M Tris-HCl-0.001M EDTA buffer (pH 7.4) and centrifuged at 12,000 rpm for 30 min at 4°C. The supernatant was collected and used for the experiments. Brain tissue (250 mg) was sliced into pieces and homogenised in appropriate buffer in cold condition (pH 7.0) to give 20% homogenate (w/v). The homogenate were centrifuged at 1000 rpm for 10 min at 0°C in cold centrifuge. The supernatant was separated and used for the assay of various biochemical estimations [8,19].

Determination of acetyl cholinesterase activity

Acetyl cholinesterase (AChE) activity was determined in brain using acetylcholine iodide as a substrate according to the method of [20]. In this method AChE in samples hydrolyzes acetyl thiocholine iodide into thiocholine and butyric acid. The thiocholine reacts with 5,5'-dithiobis-2-nitrobenzoic acid to form 5-thio-2-nitrobenzoic acid. The yellow colour intensity is measured spectrophotometrically at 412 nm (Elico-SL177, Elico LTD. Hyderabad, Andhra Pradesh, India).

Estimation of TBARS content

The concentration of thiobarbituric acid reactive substances (TBARS) were estimated in brain by the method of [21] using 1,1’,3,3’-tetramethoxypropane as the standard.

Determination of non-enzymatic antioxidants

The levels of reduced glutathione (GSH) in brain homogenate were
determined by the method of [22], based on the reaction with Ellman’s reagent (19.8 mg DTNB in 100 ml of 0.1% sodium citrate). The levels of total sulphhydryl groups (TSH) were measured after the reaction with 5,5’-dithiobis-2-nitrobenzoic acid using the method of [23].

Determination of enzymatic antioxidants

Superoxide dismutase (SOD) activity in the brain was assayed by the method of [24] based on the inhibition of formation of NADH-phenazine methosulphate-nitroblue tetrazolium complex. One unit of SOD corresponds to the amount of enzyme causing 50% reduction of nitro blue tetrazolium/min/mg of protein. Catalase (CAT) was assayed by the method of [25] by quantitating the H₂O₂ consumed after the enzymatic reaction. Dichromate in acetic acid was used as the coloring agent. The activity of catalase is expressed as μmoles H₂O₂ consumed/min/mg of protein. GPx activity was assayed in brain by the method of [26].

Processing of tissues for histopathological studies

One half of brain was fixed in 10% formalin solution. The fixed tissues were processed, embedded in paraffin and sectioned (5-6 μ). The sections were stained with hematoxylin and eosin (H and E) and observed under microscope. The brain sections were examined for gliosis, pycnosis, spongiosis, inflammatory infiltrate, oedema and meningeal changes.

Statistical analysis

Values are given as the mean ± S.D. Significant difference between the means of the six groups was statistically analyzed by Duncan’s Multiple Range Test (DMRT). The significance levels was set at P<0.05 for all the tests. Statistical analysis was performed using SPSS 11.0 software package (SPSS, Chicago, IL, USA).

Results

The bioassay test shows that the venom was more stable and active at the optimum pH of at 7-8.

Brain AChE

The activity of AChE in brain of control and experimented rats are shown in Figure 3. The activities of AChE in brain was significantly (P<0.05) decreased in alcohol treated rats when compared with control rats whereas the administration of venom in alcohol intoxicated rats significantly (P<0.05) increased the activities of AChE to near normal levels when compared with alcohol treated rats.

Oxidative stress marker in brain

The data presented in Figure 4 shows the changes in the levels of brain lipid peroxidation indices thiobarbituric acid reactive substances (TBARS). The rats treated with alcohol, the levels of brain TBARS was increased (P<0.05) significantly when compared with control rats. Administration of venom in alcohol intoxicated rats significantly (P<0.05) decreased the level of these oxidative stress marker in brain when compared with alcohol treated rats.

Non-enzymatic antioxidants in brain

The level of non-enzymatic antioxidants (GSH and TSH) in the brain of control and experimental rats were shown in Table 1. The levels of GSH and TSH were significantly (P<0.05) decreased in the brain tissues of alcohol intoxicated rats, when compared to control rats. Administration of venom in alcohol treated rats significantly (P<0.05) depleted the levels of GSH and TSH contents in brain when compared with alcohol treated rats.

Enzymatic antioxidants brain

The activities of enzymatic antioxidants (SOD, CAT and GPx)
levels of TBARS with concomitant improvement in the levels of brain enzymatic and non-enzymatic antioxidant defences.

Since the brain AChE activity is an important regulator of the behavioural processes, the decreased level of AChE in brain might be one of the indicators for alcohol induced toxic manifestations. Administration of venom in alcohol intoxicated rats restored the activity of AChE in brain. An important aspect of venom’s role in neuroprotection is by the attenuation of radical formation through its acetylcholine (Ach) or a cholinomimetic compound present in the fish venom. Accumulating lines of evidence had suggested that the availability of Ach lead to increased levels of acetylcholinesterase (AChE) activity [28]. Several studies have examined the role of oxidative stress in developmental alcohol-mediated enzymatic and non-enzymatic depletion, possibly via the formation of free radicals [29-31]. GSH is the most abundant non-protein thiol that maintains the cellular redox status and providing first line of antioxidant protection against oxidative stress in brain. Thiols are potent cheaters capable of mobilizing even intracellularly bound alcohol and also provide an antioxidant defence function by removing alcohol from the site of deleterious oxidant reactions. The diminished levels of GSH and TSH in alcohol intoxication could be due to increased utilization to overwhelm the production of free radicals by alcohol and subsequent lipid peroxidation in brain. The activities of SOD, CAT and GPx, which were further decreased in the presence of Alcohol. SOD, CAT and GPx constitute the principal component of enzymatic antioxidant system against oxidative stress. SOD dismutates O2- to H2O2, which is removed by GPx and catalase. Moreover, the active compound of lionfish venom (cholinomimetic) may contribute to improve the tissue thiol pools, which could be associated with a reduction in alcohol induced oxidative threat and increased antioxidant status in venom administrated alcohol-treated rats.

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References

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