The Studies of Antioxidant and Antimicrobial Potentials of the Leaf Extract of Bauhinia monandra Plant

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Abstract

In the present study, the dried leaf (144g) of B. monandra plant was extracted with 95% ethanol. The crude extract was partitioned into ethyl acetate and n-hexane layers to afford fractions with golden brown and greenish yellow colours respectively. The fractions were concentrated and coded EFBM and HFBM. The preliminary phytochemical screening conducted on the crude extract and ethyl acetate fraction (EFBM) revealed the presence of flavonoids, tannins, steroids, terpenoids, saponin, cardiac glycoside and phenols. The antioxidant activity of EFBM and HFBM was evaluated using DPPH radical scavenging assay with ascorbic acid as a reference standard. However, the result of analysis revealed that EFBM exhibited high antioxidant (IC50 µg/mL=0.010) activity than HFBM (IC50 µg/m = 5.564) and reference standard ascorbic acid (IC50 µg/mL=30). The antibacterial activity of EFBM and HFBM against three gram-negative bacterial strains (Escherichia coli, Pseudomonas aeruginosa and Klebsiella oxytoca) was also evaluated and it was found that EFBM exhibited highest antibacterial potential. In contrast to this, HFBM showed negligible or zero activity against all the bacteria strains. The results of the GC-MS analyses of EFBM and HFBM showed the presence of nine different phytochemicals. In EFBM, oleic acid which is classified as monounsaturated omega-9-fatty acid with percentage concentration, (40.76%) was the major compound as identified by the GC-MS and 4-hydroxy-5-methyl-3-propyl-2-hexanone with percentage concentration, (42.70%) was identified as the major compound in HFBM. The results of this studies demonstrated that folk medicine can be as effective as modern medicine to combat pathogenic microorganisms. The millenarian use of the leaf of this plant in folk medicine suggests that it represents an economic and safe alternative to treat infectious diseases.

Keywords: Phytochemicals; EFBM; HFBM; Antioxidant; Antimicrobial; GC-MS

Introduction

Antioxidants from plant extract are compounds that demonstrated biological activity which can protect the body from damage caused by free radical-induced oxidative stress [1]. Many of today’s diseases that result from imbalance between formation and neutralization of pro oxidant are associated with oxidative stress. Oxidative stress is initiated by free radicals, which seek stability through electron pairing with biological macromolecules such as proteins, lipids and DNA in healthy human cells and cause protein and DNA damage along with lipid peroxidation. These changes contribute to cancer, atherosclerosis, cardiovascular diseases ageing and inflammatory diseases [2,3]. The uses of medicinal herbs are increasingly gaining acceptance even among the literates in the urban settlements, probably because of their effectiveness, affordability, availability, low toxicity, acceptability [4], and also to the increasing inefficiency of many modern drugs used for the control of many infections such as typhoid fever, gonorrhoea, diabetes, tuberculosis as well as increase in resistance by several bacteria to various antibiotics and increase cost of prescription drugs for the treatment of manifestations caused by microorganism and to establish chemical constituents of ethyl acetate and n-hexane fractions of ethanolic leaf extract of B. monandra. Therefore, the present study is an endeavour to find out good natural antimicrobial and antioxidant compounds for the treatment of manifestation caused by microorganism and to establish chemical constituents of ethyl acetate and n-hexane fractions of ethanolic leaf extract of B. monandra. In the present study, ethyl acetate and n-hexane fractions of the ethanolic leaves extract of Bauhinia monandra plant were tested for their phytoconstituents, antioxidant using DPPH assay and antimicrobial potential against various pathogenic bacteria strains. The GC-MS analysis of these fractions was also carried out.

Material and Methods

Material

Chemicals: 1,1-diphenyl-2-picrylhydrazyl (DPPH) and ascorbic acid (Vitamin C) were obtained from Sigma-Aldrich (Germany). Acetic acid, sodium hydroxide, n-Hexane, ethyl acetate, hydrochloric acid (HCl), tetraoxosulphate (VI) acid (H2SO4), chloroform (CHCl3),

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ethanol (C<sub>2</sub>H<sub>5</sub>OH) and ferric chloride were obtained from the chemical store of the Department of Chemistry, Kwara State University, Malete, Nigeria. Solvents were redistilled before use while reagents were used without further purification. All other chemicals and reagents were of analytical reagent grade.

**Microorganisms:** Three-gram negative bacteria namely, *Escherichia coli*, *Klebsiella oxytoca* and *Pseudomonas aeruginosa* used in this study were collected from the department of Microbiology, Kwara State University, Malete, Nigeria. Nutrient agar was used as the growth media for the bacteria.

**Instrument:** GC-MS analysis: GC-MS analysis on the fractions was carried out in a Shimadzu GC-MS-QP-2010 plus chromatography with capillary column (methyl phenyl siloxane, 60 mm × 0.25 mm); ionization detector; carrier gas, helium, flow rate 1.6 ml/min; column oven temperature 4 min in 600°C and 8 min in 160-2800°C; injection temperature, 2500°C; viscosity compound time, 0.2 sec., washing volume, 8 µL., split ratio 1.0. The MS parameters were as follows: ion source temp., 2000°C., interface temperature, 2500°C, solvent cut time, 2.5 min., scan speed, 769 amu/s, start time, 30 min, end time 27 min, start m/z, 40.0 and end m/z, 400. Identification was based on the molecular structure, molecular mass and calculated fragments. Interpretation on mass spectrum GC-MS was conducted using the database of National Institute Standard and Technology (NIST 05). The name, molecular weight and structure of the components of the test fraction were ascertained. The spectrum of the unknown component was compared with the spectrum of the component stored in the NIST library.

**Methods**

**Collection and identification of plant sample:** The leaves of *Bauhinia monandra* were collected from Kwara State University, Malete environ in September, 2014. The plant was identified at the department of Plant biology, Kwara State University, Malete.

**Preparation of sample:** Fresh leaves of *B. monandra* were collected and air dried in the laboratory at a temperature of 25°C for 2 weeks. They were subsequently pulverized into fine powder weighing one hundred and forty four grams (144 g)

**Extraction:** 144 g of the powdered leaves was extracted with distilled ethanol (1.4 L) for three days. The extract (greenish dark) was decanted; filtered and concentrated using rotary evaporator at 78°C to afford dark crude extract (16.8 g and 11.6 %).

**Solvent-solvent partitioning of ethanol crude extract of *B. monandra* leaf**

**Partitioning with n-hexane:** 5.0 g of the crude extract was dissolved in distilled water (50 mL) and filtered. The filtrate (brown) was taken in a separating funnel and n-hexane (100 mL) was added. The funnel was shaken vigorously and allowed to stand for a few minutes. The n-hexane fraction was collected. The process was repeated three times. Then ethyl acetate fraction was concentrated using rotary evaporator at 400°C to afford golden brown fraction which was coded as (EFBM).

**Partitioning with ethylacetate:** 5.0 g of the crude extract was dissolved in distilled water (50 mL) and filtered. The filtrate (brown) was taken in a separating funnel and ethylacetate (100 mL) was added. The funnel was shaken vigorously and allowed to stand for a few minutes. The ethyl acetate fraction was collected. The process was repeated three times. Then ethyl acetate fraction was concentrated using rotary evaporator at 400°C to afford golden brown fraction which was coded as (EFBM).

**Preliminary phytochemical screening of crude extract and fraction (EFBM)**

Chemical tests were carried out on the ethanolic extract of *B. monandra* leaf and its fraction (EFBM) using standard procedures to identify the constituents as described by [17-19]. The specific procedure involved for the evaluation of a particular group of chemicals is mentioned below.

**Tannins:** One ml of water and 1-2 drops of ferric chloride solution were added in 0.5 ml of extracted solution. Blue colour was taking as indication for tannins.

**Flavonoids (Alkaline Reagent Test):** Extract was treated with few drops of sodium hydroxide solution. Formation of intense yellow colour, which becomes colourless on addition of dilute acid, indicates the presence of flavonoids.

**Steroids:** Two ml of acetic anhydride was added to 0.5 g ethanolic extract of sample with 2 ml H<sub>2</sub>S<sub>O</sub><sub>4</sub>. The color changed from violet to blue indicating the presence of steroids.

**Terpenoids (Salkowski test):** Five 5 ml of aqueous extract was mixed in 2 ml of chloroform, and concentrated H<sub>2</sub>S<sub>O</sub><sub>4</sub> (3 ml) was carefully added to form a layer. A reddish brown colouration of the interface was formed to show the presence of terpenoids.

**Cardiac glycosides (Keller-Killani test):** Five ml of aqueous extract was treated with 2 ml of glacial acetic acid containing one drop of ferric chloride solution. This was under layered with 1 ml of concentrated tetraoxosulphate (VI) acid. A brown ring of the interface indicates the presence of deoxyugar characteristic of cardenolides.

**Alkaloids:** Dragendorff’s reagent test was conducted for detection of alkaloids: 0.5 g of aqueous extract was dissolved in 5 ml of 1% HCl and the mixture was kept for 2 minutes in water bath. 1 ml of filtrate is treated with dragendorff’s reagent. Turbidity or precipitation occurs in alkaloid rich samples. The percentage of deoxyugar characteristic of cardenolides was taking as indication for presence of alkaloids.

**In vitro assessment of the biological activities of HFBM and EFBM**

**DPPH antioxidant assay:** The ethylacetate and hexane fractions (EFBM and HFBM) from the ethanolic extract of the leaf of *B. monandra* were subjected to antioxidant test in order to determine the free radical scavenging power. The radical scavenging capacity was determined according to the method described by Cervato et al. with minor modifications. The DPPH was prepared freshly and kept away from light after preparation. 1.0 mL of fraction with different concentrations (500, 250, 200, and 100 µg/mL) was separately mixed with 3.0 mL of 60 µM methanolic solution of DPPH radicals; the mixture was kept in room temperature in the dark for 30 min and the absorbance measured at 517 nm. Blank experiment was also carried out to determine the absorbance of DPPH before interacting with the fraction. The decrease in absorbance of DPPH on addition of test samples in relation to the positive control (Ascorbic acid) was used to calculate the percentage inhibition (% Inhibition) following the equation:

\[
I \% = 100 \times \frac{Abs_{control}-Abs_{sample}}{Abs_{control}}
\]

The IC<sub>50</sub> (the concentration that inhibited 50% of the free DPPH radicals) determined.

**Data analysis:** The 50% inhibition (IC<sub>50</sub>) was calculated from dose-response-inhibition nonlinear regression equation of each fraction. The results represent the mean ± standard error of the mean values of duplicate experiments. IC<sub>50</sub> values were approximated with statistical significance p ≤ 0.01 and with high regression coefficients.
Antimicrobial Assay

Preparation of Mueller Hinto Agar (MHA): MHA was prepared according to the manufacturer’s instruction 36 g of MHA was dissolved in 1 L of sterile distilled water. It was allowed to homogenize before sterilizing inside the autoclave at 121°C for 15 minutes. The sterilized MHA was aseptically poured into sterile disposable petri dishes and allowed to set.

Preparation of bacteria inoculum: Innoculum of Pseudomonas aeruginosa, Klebsiella oxytoca and Escherichia coli was prepared according to Kirby-Bauer method. Little quantity of the bacteria culture was suspended in 9 ml sterile normal saline and serially diluted until the colour of the serially diluted bacteria and normal saline resembles the colour of 0.5 McFarland standard. The suspension was swabbed on the surface of the prepared MHA plates.

Antimicrobial susceptibility test: All the inoculated plates were bored at the center with 6 mm cork borer and each extract was introduced to the hole with the aid of capillary tubes. The plates were allowed to stand uprightly for about 20 minutes before they were transferred to the thermostatically stable incubator. All the plates were incubated at 35 ± 2°C (Normal body temperature).

Results and Discussion

Results

Result of the phytochemical screening: The preliminary phytochemical screening conducted on the ethanol extract of B. monandra and its ethyl acetate fraction (EFBM) demonstrated the presence of flavonoids, phenols, steroids, alkaloid, glycoside, tannins, saponin and terpenoids as shown in Table 1.

Results of In vitro assessment of the biological activities of HFBM and EFBM

Result of DPPH Antioxidant assay: The result of antioxidant studies of HFBM and EFBM using DPPH free radical scavenging assay with ascorbic acid as reference standard is depicted in Table 2.

Result of the Antimicrobial Assay: The result of antimicrobial activities of HFBM and EFBM against three gram-negative bacteria is as shown in Table 3.

Results of GC-MS analysis

Result of GC-MS analysis of EFBM: The result of GC-MS analysis of EFBM revealed many phytochemicals, which contribute largely to the medicinal activities of plants. These phytochemicals, with their molecular weight, molecular formulae, percentage area, mass peak and retention index are depicted in Table 4. The total number of nine compounds were identified and the major compound was oleic acid with percentage composition, (40.76%), the next abundant compound was found to be hexadecanoic acid with percentage concentration (21.75%) (Figure 1).

Result of GC-MS analysis of HFBM: The result of GC-MS analysis of HFBM revealed nine different phytochemicals. These compounds with their molecular formulae, molecular weight, retention index, percentage area and mass peak are presented in Table 5 below. The most abundant compound was 9-octadecenoic acid (oleic acid), (20.08%) (Figure 2).

Discussion

In the present study, the results of phytochemical screening demonstrated the presence of flavonoids, terpenoids, steroids, alkaloid, cardiac glycoside, saponin, tannins and phenols (Table 1). Thus the presence of these phytochemicals in leaf of B. monandra plant may be associated with its uses by traditional medicine practitioners in healthcare system in the treatment of diabetes and other infections. From (Table 2), it can be inferred that the EFBM with lower minimum inhibition concentration (IC50, µg/mL=0.010) showed significantly high antioxidant activity than HFBM with (IC50, µg/mL=5.564) and the reference standard ascorbic acid (IC50, µg/mL=30). The antioxidant activity of EFBM is strongly due to the presence of flavonoids and phenols contents as established by qualitative phytochemical screening in this study. Table 3 presents the results of antibacterial activity of EFBM and HFBM against three gram-negative bacteria. It can be deduced that EFBM has broadest spectrum of activity on the tested bacteria. The results show that its activity against Pseudomonas aeruginosa was significantly higher than Escherichia coli, and Klebsiella oxytoca. In contrast to this, HFBM showed negligible or zero activity against all the tested bacteria strains. The zero activity of HFBM against tested bacteria strains in this study is in agreement with the findings of [20]. The antimicrobial activity of EFBM may be due to the presence of phytochemicals such as phenols, flavonoids, alkaloids, terpenoids which are naturally biosynthesised in this plant and serve as defense mechanism against any pathogenic microorganisms [21]. The gas chromatogram of EFBM and HFBM depicted the relative concentrations of various phytochemicals getting eluted as a function of retention time (RT). The relative concentration of the isolated compound is indicated by the heights of the peaks. The mass spectrometer analysis the compounds eluted at different times so as to identify the nature and the structures of the compounds. The large compounds fragment into smaller compounds giving rise to appearance of peaks at different m/z ratios. In this study, nine different compounds were identified in both EFBM and HFBM by GC-MS as depicted in (Tables 4 and 5). In EFBM, fatty acids such as 9-octadecenoic acid (oleic acid), hexadecanoic acid and octadecanoic acid (stearic acid), with percentage concentrations [40.76, 21.75 and 14.01] are the most abundant compounds respectively, both constituting 72.51% of the total. These compounds also present in HFBM, but at a lower concentration as compared to EFBM. The

<p>|</p>
<table>
<thead>
<tr>
<th>Phytochemicals</th>
<th>Ethanolic Flower Extract of B. monandra</th>
<th>Ethylacetate fraction of B. monandra</th>
</tr>
</thead>
<tbody>
<tr>
<td>Flavonoids</td>
<td>++</td>
<td>++</td>
</tr>
<tr>
<td>Poly phenol</td>
<td>++</td>
<td>++</td>
</tr>
<tr>
<td>Cardiac Glycoside</td>
<td>++</td>
<td>++</td>
</tr>
<tr>
<td>Terpenoids</td>
<td>++</td>
<td>++</td>
</tr>
<tr>
<td>Tannins</td>
<td>++</td>
<td>++</td>
</tr>
<tr>
<td>Steroid</td>
<td>++</td>
<td>++</td>
</tr>
<tr>
<td>Saponin</td>
<td>++</td>
<td>++</td>
</tr>
<tr>
<td>Alkaloids</td>
<td>++</td>
<td>++</td>
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| Table 1: Result of phytochemical screening of ethanolic leaf extract of B. Monandra.

<table>
<thead>
<tr>
<th>Test fractions</th>
<th>IC50, µg/ML</th>
</tr>
</thead>
<tbody>
<tr>
<td>HFBM</td>
<td>5.564</td>
</tr>
<tr>
<td>EFBM</td>
<td>0.010</td>
</tr>
<tr>
<td>Ascorbic acid</td>
<td>30.00</td>
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</tbody>
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| Table 2: IC50 of test fractions using the DPPH assay.

<table>
<thead>
<tr>
<th>Fraction</th>
<th>Bacteria</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Escherichia coli</td>
</tr>
<tr>
<td></td>
<td>Pseudomonas aeruginosa</td>
</tr>
<tr>
<td></td>
<td>Klebsiella oxytoca</td>
</tr>
<tr>
<td>EFBM</td>
<td>16.50 mm</td>
</tr>
<tr>
<td></td>
<td>25.00 mm</td>
</tr>
<tr>
<td></td>
<td>11.00 mm</td>
</tr>
<tr>
<td>HFBM</td>
<td>0.00 mm</td>
</tr>
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<td></td>
<td>0.00 mm</td>
</tr>
<tr>
<td></td>
<td>0.00 mm</td>
</tr>
</tbody>
</table>
|          | Table 3: Result of Antimicrobial Assay of EFBM and HFBM.

s/n | Compound | Ret. Index | MW | % area | mass peak
--- | --- | --- | --- | --- | ---
1 | 4-hydroxy-5-methyl-3-propyl-2-hexanone, | 1185 | C_{16}H_{32}O_{2} | 172 | 42.7 | 41,43,58,72,85,100
2 | 9-octadecenoic acid (oleic acid) | 2175 | C_{18}H_{34}O_{2} | 282 | 20.08 | 27,41,55,69,83,97,123,137,264
3 | 2,2-dimethyl pentanal | 821 | C_{7}H_{14}O | 114 | 11.26 | 27,41,55,69,83,97,123,137,264
4 | Hexadecanoic acid | 1968 | C_{16}H_{32}O_{2} | 256 | 9.41 | 27,41,43,60,73,85,98,115,129,157,171,185,213,256
5 | Octadecanoic acid (stearic acid) | 2167 | C_{18}H_{36}O_{2} | 284 | 4.98 | 27,41,43,58,83,85
6 | Hexadecanoic acid | 1968 | C_{16}H_{32}O_{2} | 256 | 9.41 | 27,41,43,60,73,85,98,115,129,157,171,185,213,256
7 | 9,2-octadecadien-1-01,(z,z) | 2069 | C_{18}H_{34}O | 266 | 5.16 | 41,55,67,81,95,110,121,266
8 | 4-tridecene,(z) | 1321 | C_{13}H_{26} | 182 | 8.03 | 27,41,55,69,70,84,98,182
9 | Pentafloropropionic acid,octadecyl ester | | | | | 41,48,57,71,97,111,125,252

Table 4: The result of GC-MS analysis of EFBM.

s/n | Compound | Ret. Index | molecular formula | molecular weight | % area | mass peak
--- | --- | --- | --- | --- | --- | ---
1 | 4-hydroxy-5-methyl-3-propyl-2-hexanone, | 1185 | C_{16}H_{32}O_{2} | 172 | 42.7 | 41,43,58,72,85,100
2 | 9-octadecenoic acid (oleic acid) | 2175 | C_{18}H_{34}O_{2} | 282 | 20.08 | 27,41,55,69,83,97,123,137,264
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9 | Pentafloropropionic acid,octadecyl ester | | | | | 41,48,57,71,97,111,125,252

Table 5: The result of GC-MS analysis of HFBM.

major compound in HFBM was found to be 4-hydroxy-5-methyl-3-propyl-2-hexanone and its biological activity is yet to establish. The total percentage concentration of the fatty acids in HFBM was 34.47.

Conclusion

Bauhinia monandra leaf is a rich source of phytochemicals with proven antioxidant and antimicrobial activities. The phytochemical analysis conducted on B. monandra extract and its ethyl acetate fraction (EFBM) revealed the presence of flavonoids, cardiac glycoside, steroids, phenolics, terpenoids, alkaloid, and saponin. The presence of these phytochemicals confirmed the uses of this plant in folkloric usage of the leaf of studied plant and suggest that EFBM possesses compounds with antioxidant and antimicrobial properties that can be further explored for in vivo and clinical studies so as to justify its antioxidant and antimicrobial activities. The results of this studies demonstrated that folk medicine can be as effective as modern medicine to combat pathogenic microorganisms. The millenarian use of the leaf of this plant in folk medicine suggests that it represent an economic and safe alternative to treat infectious diseases.

Reference


